



Nanoparticle Based Sensors and Organic Nanospintronic transistors

Hadi Rasam AlQahtani

Supervisors

Dr Martin Grell

Dr Dan Allwood

**A thesis submitted in partial fulfillment for the
degree of Doctor of Philosophy with Integrated Studies
(New Route PhD) in Nanoelectronics and Nanomechanics**

Faculty of Science

Department of Physics and Astronomy

May 2013

Abstract

The work presented in this doctoral thesis is mainly divided into two main parts: nanoparticle swelling based sensors and organic nanospintronics.

In the swelling based sensors work, three novel experimental methods for enhancing the sensitivity of gold core shell nanoparticle (Au-CSNP) films are presented. The first method utilises a long ligand of alkane-thiols, e.g. dodecanethiols and a significant response was obtained for alkanes with long carbon chain such as decane ($C_{10}H_{22}$) which is found in petrol. The sensitivity of swelling- based gold core- shell nanoparticle vapour sensors can be enhanced considerably by cooling sensors below ambient temperature. We found that the sensitivity to a particular analyte scales with temperature like that of the analyte's saturated vapour pressure and the sensitivity is linked to the analyte's enthalpy of vaporisation. This allows for quantitative prediction of sensitivity enhancement for vapours not yet tested. We demonstrated the detection of low level of a biogenic odour that is released by E.coli bacteria (1-decanol odour) at a partial pressure in the order 100 ppb using Au-CSNPs decorated with a long -OH terminated ligand. This is an exceptionally low limit of detection for swelling- based sensors, and relies firstly, in the careful matching of the CSNPs ligands to the targeted odour, and secondly, in the very low volatility of this odour.

In the spintronic part, the organic spin field effect transistor was demonstrated for the first time and about -1400% giant magnetoresistance at was estimated at room temperature using such transistor. This probable GMR effect is exceptionally high and could have a strong impact on the field of organic spintronics. Also, we developed a platform (in-plane spin valve structure based on $Ni_{80}Fe_{20}$ nanostructures) for studying spin transport in organic semiconductors. Using such platform, about -0.4% magnetoresistance was obtained with the electron transporter organic semiconductor PTCDI- C_{13} . Unlike vertical spin valve structures, our in-plane structure does not suffer from the ill-defined organic/ferromagnetic interface as the organic semiconductor will always be deposited on the top of the ferromagnetic contacts. This allows for the depositing of different organic layers on the same device multiple times as long as the organic materials can be washed out by a suitable solvent.

Acknowledgments

All praise and thanks are due to Allah, the one, the Lord of everything. If I were to count His bounties upon me, I would definitely fail. Peace and blessings be upon Prophet Muhammad and all other Prophets.

My deepest gratitude goes to my father Rasam AlQahtani, my mother Hadba AlQahtani, and my wife Maryam AlQahtani. I ask Allah to reward them all for what they have done for me.

I would like to express my deepest thanks and appreciations to my supervisors Dr Martin Grell and Dr Dan Allwood. I have been greatly benefited from their guidance, help, support and fruitful discussions. I was very lucky to work with them as great scientists and friends in the same time. I am indebted to Dr Matt T. Bryan for helping me a lot in the magnetic part of this project and taking me through the wonders of nanoscale patterning and answering my endless questions. Also, I would like to acknowledge the help and support that I received from Dr Tim Richardson who sadly passed away during the writing up of this thesis.

I would like to thank my examiners Prof. Alexei Nabok and Prof. Nigel Clarke who provided encouraging and constructive feedback and I am grateful for their thoughtful and detailed comments.

Special thanks to Dr. Mohammad Al-duraibi for helping me during his summer visit in 2011, Dr Lee Hague and Dr Stuart Brittle for helping me a lot especially in my first year of PhD and Dr. D. Puzzovio for helping in the decanol sensor work.

Thanks also to Dr M.T. Bryan, M.P.P Hodges, and Dr. T.J. Hayward for doing the M-TXM measurement at the ALS beamline, Berkeley, USA. Also I would like to thank Dr P John Thomas and his group in Manchester University for supplying us with gold nanocrystals films that were used as amine sensors.

I would like to acknowledge the technical support that I received from Dan Jackson (in chemistry), Simon Dixon, Paul Kemp-Russell, Pete Robinson, and Chris Vickers (Physics workshop) Paul Hawksworth (in material science and engineering), and Saurabh Kumar (in Kroto nanoscience centre). Also I would like to thank my office and research mates in the Departments of Physics and Materials science & Engineering for their friendliness and offering help when it was needed.

Finally, I would like to thank the kind support that I received from Saudi Cultural Bureau in London throughout my tenure here and King Saud University for the provision of a doctoral fellowship.

Hadi AlQahtani

May 2013

List of Publications and Conferences

Published papers

1. **AlQahtani, H.**, Puzzovio, D., Dragoneas, A., Richardson, T., and Grell, M. (2012). A swelling-based chemiresistor for a biogenic odour. *Talanta*, 99: p. 50-54.
2. **AlQahtani, H.**, Alduraibi, M., Richardson, and T., Grell, M. (2012). Manifold sensitivity improvement of swelling-based sensors. *Phys. Chem. Chem. Phys.*, 14, 5558-5560.
3. **AlQahtani, H.**, Sugden, M., Puzzovio, D., Hague, L., Mullin, N., Richardson, T., and Grell, M. (2011). Highly sensitive alkane odour sensors based on functionalised gold nanoparticles. *Sensors and Actuators B-Chemical*, 160, 399-404.
4. Stansfield, G. L., Vanitha, P. V., Johnston, H. M., Fan, D., **AlQahtani, H.**, Hague, L., Grell, M., and Thomas, P. J. (2010). Growth of nanocrystals and thin films at the water-oil interface. *Philosophical Transactions of the Royal Society a-Mathematical Physical and Engineering Sciences*, 368, 4313-4330.

Papers to be submitted:

1. Planar organic spin valves using nanostructured Ni₈₀Fe₂₀ magnetic contacts.
2. Electrolyte gated organic spin transistor. Under preparation.

Conferences

1. Talk will be given in Sensors 2013 AAMG conference, June 2013, London, UK.
2. Poster in JSPS York-Tohoku Research Symposium on "Magnetic Materials and Spintronics" June 2013, University of York, York, UK.
3. Poster in SPINOS, Sep. 2012, University College of London, London, UK.
4. Talk in IEEE NANO 2012, 22-24 Aug. 2012, Birmingham, UK.
5. Poster presentation in European Conference on Organized Films (ICOMF 12), July 2011, Sheffield Hallam University, Sheffield, UK.
6. Poster presentation in UK & RI Chapter of the IEEE Magnetics Society EGM and Student/RA Workshop, Nov. 2010, University of Sheffield, UK.

Contents

Chapter 1: Introduction and Thesis Organisation

1.1 Introduction and Motivations	1
1.2 Core-Shell Nanoparticle Sensors.....	1
1.3 Organic Nanospintronics.....	3
1.4 Nanostructures: Importance and Classification.....	6
1.5 How this thesis is organised?	8
1.6 References.....	10

Chapter 2: Nanoscale Swelling Based Sensors: Background

2.1 Introduction	12
2.2 Swelling Based Sensors: an Overview.....	12
2.3 Thermodynamics of Evaporation and Swelling	13
2.3.1 Vapour Pressure	13
2.3.1.1 Intermolecular interactions.....	14
2.3.1.2 Effects of the molecular weight	14
2.3.1.3 Saturation, Relative vapour pressure and Lower Explosive Limit.....	15
2.3.1.4 Evaporation Rate.....	16
2.3.1.5 Metabolic Vapour Pressure Build-up.....	18
2.3.2 Clausius-Clapeyron Equation and Enthalpy of Vaporisation	19
2.3.3 Hildebrand Solubility Parameter	20
2.4 Core-Shell Nanoparticles.....	23
2.4.1 Electron Transport in CS-AuNP films	23
2.4.2 Literature Review of Nanoparticle Swelling Based Sensors.....	26
2.5 Conclusion.....	28
2.6 References.....	28

Chapter 3: Nanoscale Magnetism and Spintronics

3.1 Introduction	32
3.2 Nanoscale Magnetism.....	32
3.2.1 Spin and orbital magnetic moments	32
3.2.2 Spin-Orbit Coupling	34

3.2.3 Micromagnetic Energies	35
3.2.3.1 Exchange energy	35
3.2.3.2 Magnetocrystalline energy.....	36
3.2.3.3 Magnetostatic energy	37
3.2.3.4 Magnetostrictive energy	38
3.2.3.5 Zeeman energy.....	38
3.2.4 Magnetic Domains and Domain Walls.....	38
3.4.4.1 Bloch vs Néel Domain Walls	40
3.2.4 Dynamics of domain walls	42
3.2.5 Magnetisation Reversal and Hysteresis.....	43
3.2.6 Nanoscale Effects.....	44
3.2.7 Magnetic Nanowires.....	45
3.3 Magnetoresistance.....	46
3.3.1 Hall Magnetoresistance.....	47
3.3.2 Anisotropic Magnetoresistance	48
3.4 Spintronics.....	49
3.4.1 Spintronic Magnetoresistance.....	49
3.4.2 Interlayer magnetic couplings.....	54
3.4.2.1 Oscillatory Exchange Coupling	52
3.4.2.2 Magnetostatic Coupling.....	52
3.4.2.3 Interface Roughness coupling.....	54
3.4.3 Spin Transport.....	55
3.4.3.1 FM Density of States.....	55
3.4.3.2 Spin injection and transport.....	56
3.5 Conclusion.....	59
3.6 References.....	59
Chapter 4: Organic Spintronics	
4.1 Introduction	62
4.2 Organic Semiconductors	62
4.2.1 Charge Transport in OSCs	65
4.2.1.1 Bulk- limited transport: Mobility.....	65

4.2.1.2 Metal/Organic Interface	67
4.2.1.3 Contact Limited Transport	68
4.2.1.4 Space Charge Limited Current	69
4.2.1.5 Trap Limited Transport.....	70
4.2.2 Organic Field Effect Transistor	71
4.2.2.1 Electrolyte gated OFET	74
4.2.3 Examples of OSCs.....	75
4.3 Organic Spintronics.....	75
4.3.1 Spin diffusion length and relaxation time.....	76
4.3.2 Organic Spin Valve (OSV)	77
4.3.3 State of Art in Organic Spintronics	78
4.3.4 Conductivity Mismatch.....	84
4.4 Conclusion.....	87
4.5 References.....	87
Chapter 5: Experimental Methods	
5.1 Device Fabrication Methods	90
5.1.1 Langmuir-Schaefer Printing.....	90
5.1.2 Oil-Water Interfacial Deposition	92
5.1.3 Self Assembled Monolayers.....	93
5.1.4 Thermal Evaporation	94
5.1.5 Sputtering.....	95
5.1.6 Spincoating.....	96
5.2 Nanoscale Patterning.....	97
5.2.1 Electron Beam Lithography.....	97
5.2.2 Photolithography.....	99
5.3 Characterisation	100
5.3.1 Au CSNPs Characterisation	100
5.3.1.1 Resistance Measurement.....	100
5.3.1.2 Gas Sensing Experiments.....	102
5.3.1.3 Temperature-Controlled Sensing Experiments	104
5.3.2 Planar Organic Spin Valve Device Characterisation.....	106

5.3.2.1 Four-Point Probe Measurement	106
5.3.2.2 Magntoresistance setup	107
5.3.2.3 Magnetic/Organic Transistor Characterising Setup	110
5.3.2.4 Magneto-Optical Kerr Effect Magnetometry	111
5.3.2.5 Magnetic Transmission X-ray Microscopy.....	113
5.4 Structural Characterisation.....	115
5.4.1 Atomic Force Microscopy.....	115
5.4.2 Scanning Electron Microscopy	116
5.5 References.....	117
Chapter 6: Highly Sensitive Nanoscale Swelling Based Sensors	
6.1 Introduction	119
6.2 Chemical Sensors	119
6.2.1 Amine Sensor	119
6.3 Physical Swelling Based Sensors: Novel Approaches	121
6.3.1 Sensors Preparation and Characterisation.....	121
6.3.1.1 Isotherm and Structural Characterisations	121
6.3.1.2 Temperature-dependent Electrical Characterisation.....	125
6.3.2 Highly Sensitive Alkane Sensor.....	127
6.3.2.3 Results and Discussion	130
6.3.3 Cooled CSNP Sensors	134
6.3.3.1 Sensing Film and Controlling Temperature	134
6.3.3.2 Cooling Sensor Model: Our Hypothesis	134
6.3.3.3 Results and Discussion	136
6.3.4 Biogenic Odour Sensor	139
6.3.4.1 Low Volatility Approach: Hypothesis	139
6.3.4.2 Sensing Procedure	140
6.3.4.3 Results and Discussion	141
6.4 Conclusion.....	145
6.5 References.....	145
Chapter 7 : Organic Spin Valves and Spin Transistors	
7.1 Introduction	148

7.2 In-plane Organic Spintronic Platform	149
7.2.1 Device Architecture and Fabrication Process	149
7.2.2 Electrical and Magnetic Characterisations	152
7.2.3 M-TXM Characterisation.....	155
7.2.4 Spin Valve Effect.....	158
7.3 In-plane Organic Spin Field Effect Transistor	163
7.3.1 Electrolyte Gated Organic Spin transistor	163
7.3.2 Results and Discussion	165
7.3.2.1 Electrical Characteristics	165
7.3.2.2 Output Characteristics	166
7.3.2.3 Transfer Characteristics.....	167
7.3.2.4 Magnetic Field Effect under Various Gate Voltage.....	169
7.4 Conclusion.....	171
7.5 References.....	172
Chapter 8: Conclusion and Outlook	
8.1 Conclusion.....	174
8.2 Future Work.....	177

List of Figures

Figure 1.1: A Core-Shell Gold Nanoparticle.....	1
Figure 1.1: Photograph of the Buncefield fire once happened.	3
Figure 1.3: Density of states for (a) bulk and nanosystems of (b) a quantum well, (c) a quantum wire and (d) a quantum dot.....	7
Figure 2.1: Carbon black filled polymer swelling upon exposure to organic vapour	12
Figure 2.2: vapour pressure vs. molecular weight (increases with the number carbon atoms) of n-alkanes	15
Figure 2.3: Evaporation in a container with a constant flow rate.	17
Figure 2.4: quantum tunnelling through (a) thin barrier, (b) thick barrier.	23
Figure 2.5: Tunnelling conduction process between two nanoparticles cores.....	24
Figure 2.6: Conductivity vs. interparticle separation distance of Au-CSNP films with three different ligands: $C_8H_{18}SH$, $C_{12}H_{26}SH$ and $C_{16}H_{34}SH$	25
Figure 3.1: Spin-orbit interaction, from electron's perspective.....	34
Figure 3.2: Two types of exchange coupling: (a) Ferromagnetic and (b) antiferromagnetic.....	36
Figure 3.3: The demagnetising field in a magnetised specimen.	37
Figure 3.4: Different domain configurations and the associated magnetostatic interaction.....	39
Figure 3.5: A Bloch wall vs Néel wall.	41
Figure 3.6: Damped precession motion of a magnetic moment (m) around the effective magnetic field according to the LLG equation.....	42
Figure 3.7: The initial magnetisation and demagnetising curves for ferromagnetic with cubic structure.....	43
Figure 3.8: The effect of reduced dimensionality on the domain structure of an FM sample such as iron.	44
Figure 3.9: Coercivity vs. particle size.....	45
Figure 3.10: Two types of head to head domain walls in magnetic nanowires	45
Figure 3.11: Electron diffusion motion in solids.	47
Figure 3.12: Hall magnetoresistance.	47

Figure 3.13: AMR effect due to spin orbit interaction.....	48
Figure 3.14: Two different geometries of a GMR device.	50
Figure 3.15: Parallel and anti-parallel configurations of a GMR sensor.	51
Figure 3.16: Exchange coupling strength vs. spacer layer thickness for a GMR device ...	52
Figure 3.17: Magnetostatic interaction between two FM layers.....	53
Figure 3.18: Switching behaviour of an array of 20 nm thick Ni ₈₀ Fe ₂₀ wires with two pointed ends.....	54
Figure 3.19: The orange peel coupling caused by the edge roughness.	54
Figure 3.20: schematic showing the density of states for for up-spin and down-spin electrons in: a) ferromagnetic, and b) non-magnetic materials.....	55
Figure 3.21: The effect of having two different spin density of states in a GMR device. ..	57
Figure 3.22: two spin channels model.....	59
Figure 4.1: The benzene ring and its resonance structure.....	63
Figure 4.2:Organic electronic structure vs. inorganic electronic structure.....	64
Figure 4.2: Energy schematics for metal/organic interface when considering: (a) hole injection into p-type organic semiconductor, (b) electron injection into n-type organic semiconductor	67
Figure 4.4: Three transport mechanisms in OSCs: (a) thermoionic emission.....	69
Figure 4.5: Two different structures of OFETs: (a) Bottom gated OFET, (b) Top gated OFET.....	71
Figure 4.6: schematics showing: (a) Typical output characteristics of a field effect transistor: I _{SD} vs V _{SD} at different applied gate voltage, (b) transfer characteristics	73
Figure 4.7: (a) electrolyte-gated OFET, (b) electric double layer (EDL).....	74
Figure 4.8: Planar geometry of an organic spin valve.....	78
Figure 4.9: vertical geometry of an organic spin valve.....	79
Figure 4.10: An intrinsic organic magnetoresistance displayed at room temperature by ITO/PEDOT /Alq ₃ /Ca structure at different electric bias voltages.....	82
Figure 4.11: Gated organic spin valve: (a) device schematic, (b) unsaturated output characteristics due to the short channel effect.....	84
Figure 4.12: Conductivity mismatch.....	85
Figure 5.1:Langmuir-Schaefer printing technique.....	90

Figure 5.2: . A photograph of resulting device	92
Figure 5.3: (a) oil-water interfacial deposition of Au-NPs. (b) the Au-NPs films	93
Figure 5.4: Monolayers resulted from (a) thiols bind to gold and (b) silane bind to OH group in an oxide layer on a silicon or glass substrate.....	94
Figure 5.5: a schematic showing the main components of a thermal evaporator	95
Figure 5.6: Sputtering process.	96
Figure 5.7: a sketch showing the spincoating procedure.	97
Figure 5.8: A photograph for the Raith 150 electron beam lithography machine.	98
Figure 5.9: Defining a metallic nanostructure on a substrate by means of electron beam lithography and thermal evaporation.	99
Figure 5.10: Photolithography patterning for micron and submicron structures.....	100
Figure 5.11: Driving circuit of CS-AuNP films.	101
Figure 5.12: Oscilloscope screenshots, showing V_{in} (1 V/div, top and orange coloured), and V_{out} (50 mV/div, bottom and blue coloured) of dodecanethiol Au CSNP films. Time scale is 1 sec/div.....	101
Figure 5.13: Basic square wave generator circuit.	102
Figure 5.14: Schematic showing the temperature controlled sensing and gas delivery set-up.....	104
Figure 5.15: Thermoelectric elements of n- and p-type based on the Peltier effect, the inset shows a commercial Peltier element.	105
Figure 5.16: a schematic of four-point probe configuration.....	106
Figure 5.17: Magnetoresistance setup connected.	108
Figure 5.18: The graphical user interface (a) and the Block diagram (b) of the MR developed labview code.....	109
Figure 5.19: a schematic showing the contacting method of water gated organic field effect transistors.....	110
Figure 5.20: magnetoresistance measurement setup of the of gated spin valve transistors.	111
Figure 5.21: Schematic showing the basic MOKE magnetometer and its working modes: transverse (T-mode), longitudinal (L-Mode) and polar mode (P-mode).....	112
Figure 5.22: An advanced MOKE setup for ferromagnetic nanostructures	113
Figure 5.23: M-TXM beamline at ALS, Berkley, USA	114
Figure 5.24: a schematic showing the AFM basic structure.	116

Figure 5.25: the basic structure of an SEM.....	117
Figure 6.1: Octylamine odour sensing using gold nanocrystals.....	120
Figure 6.2: Langmuir isotherm of core/shell nanoparticles: (a) Dodecanthiol AuNPs isotherm, (b) Hexanethiol AuNPs isotherm, (c) Undecanolthiol AuNPs.....	123
Figure 6.3: (a) Tapping mode AFM height image rendered in 3d of dodecanethiol Au-CSNP film	124
Figure 6.4: SEM images of hexanethiol Au-CSNP film composed of 4 monolayers deposited by the Langmuir–Schäfer technique	124
Figure 6.5: the change in the electrical resistance of dodecanethiol CS-AuNP films upon varying the temperature between 22 and 50 °C. (c) the resistance percentage change vs. temperature of undecanol thiol CSNP films.....	126
Figure 6.6: (a) Oscilloscope screenshots, showing V_{in} (orange, 1V/div), and V_{out} (blue, 50mV/div) of Au CSNP films prior to exposure, during exposure to 1500 ppm of decane odour, and during recovery.....	128
Figure 6.7: Response of Au CSNP films to hydrocarbon odours of different Hildebrand parameters and hydrogen bonding strength.....	131
Figure 6.8: ‘Headspace’ exposures to decane and hexane. The recovery starting point in each case is indicated by “Off”	132
Figure 6.9: (a) Resistance change under exposure/recovery cycles, beginning with 15 ppm decane odour. (b) $\Delta R/R$ plateau vs. vapour pressure.....	133
Figure 6.10: Resistance change ($\Delta R/R$) vs time of under odour exposure / recovery cycles to 0.1 p_{sat} of decane (1a), toluene (1b), and pentane (1c).....	137
Figure 6.11: Arrhenius plot of temperature- dependant sensitivity for exposure to decane (blue circles), toluene (red squares), and pentane (green triangles).....	138
Figure 6.12: Resistance of Au-undecanolthiol CSNP film under repeated exposure/recovery cycles to 1% p_{sat} (112 ppb) decanol odour.	141
Figure 6.13: Resistance of Au-undecanolthiol CSNP films under exposure/recovery cycle to 1% p_{sat} (112 ppb) 1-decanol (a) and 10% p_{sat} (1.1 ppm) 1-decanol (b)	143
Figure 6.14: Arrhenius- like plot of sensor response at different temperatures for 1% p_{sat} (squares) and 10% p_{sat} (triangles) 1-decanol exposure.....	144
Figure 7.1: (a) planar structure of $Ni_{80}Fe_{20}$ nanowires forming a platform for organic spintronics	150

Figure 7.2: (a) An SEM image of nano-channel (~ 120 nm wide) , scale bar is 100 nm, (b) The probability distributions of the edge roughness amplitude of the channel.....	151
Figure 7.3: Wide and narrow Ni ₈₀ Fe ₂₀ nanowires (~ 120 μm in length) were contacted for electrical measurement.....	152
Figure 7.4: I-V characteristics of narrow and wide Ni ₈₀ Fe ₂₀ nanowires.....	152
Figure 7.5: MOKE Magnetisation hysteresis loops for: The single Ni ₈₀ Fe ₂₀ wires.....	154
Figure 7.6: M-TXM images of (a) positive and (b) negative magnetic field sweeps for the in-plane spin valve structure	157
Figure 7.7: (a) An SEM image of the multi-nanowires planar spin valve structure (scale bar is 10 μm).	158
Figure 7.8: I-V characteristics of the Ni ₈₀ Fe ₂₀ /PTCDI-C13 device.....	159
Figure 7.9: (a) Magnetoresistance loop of Ni ₈₀ Fe ₂₀ /PTCDI-C13 planar nanostructure. (b) Corresponding MOKE loop of similar device.....	160
Figure 7.10: A hysteresis loop obtained by sweeping the field below ±15 mT (the switching field of the narrow wire).....	161
Figure 7.11: a Schematic of our water-gated in-plane organic planar spin transistor....	164
Figure 7.12: Chemical structure of two h-polaron transporter organic semiconductors: (a) P3HT and (b) pBTTT.....	165
Figure 7.13: I-V curves of the planar organic spin valve with P3HT as organic active material: (a) 130 nm for channel length and 20 μm for channel wide (b) 120 nm channel width 60 μm channel length and.....	165
Figure 7.14: Ni ₈₀ F ₂₀ /P3HT interface with the important energy levels that determine the injection efficiency.....	166
Figure 7.15: The output characteristics of water-gated in-plane spin transistor with two different organic semiconductor channels: (a) P3HT and (b) pBTTT.....	167
Figure 7.16: Saturated transfer characteristics at source-drain voltage of -1V for two water gated spin transistors: (a) P3HT with a channel of about 130 nm in legnth and 20 μm in width. (b) pBTTT with a channel width of 40 μm.....	168
Figure 7.17: The effect of magnetic field (100 mT) on the output characteristics of water gated Ni ₈₀ Fe ₂₀ /P3HT/Ni ₈₀ Fe ₂₀ spin transistor at different gate voltages.....	170

List of Tables

Table 2.1: Intermolecular interactions that associated with hydrocarbon solvents	14
Table 2.2: Some physical and thermodynamics properties of several hydrocarbon solvents.	20
Table 2.3: Hildebrand parameter (δ) for a variety of solvents	22
Table 3.1: Easy axes and anisotropy constants in some ferromagnetic materials.	37
Table 3.2: Domain wall parameters for some bulk FM materials.....	41
Table 4.1: Some dielectric constants of inorganic insulators used for OFET gating	74
Table 4.2: Examples of commonly used organic semiconductors in the field of organic spintronics and other fields accompanied with their motilities.	75
Table 4.3: The main achievements in the field of organic spin valves.....	83
Table 6.1: Au-CSNP ligands, chemical formula, dispersion solvents and.	121
Table 6.2: Properties of solvents	130
Table 6.3: Comparison between literature values of enthalpy of vaporisation	138
Table 6.4: Decanol properties: saturation pressure, solubility parameter and enthalpy of vaporisation.....	142
Table 7.1: Switching fields of a planar spin valve device comprising.	161
Table 7.2: Comparing the MR, spin diffusion length and spin relaxation time of our device show in red with the devices reported in literature.....	162