

Optics of atomically thin films and van der Waals heterostructures made from two-dimensional semiconductors



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This thesis is submitted for the degree of

Doctor of Philosophy

September 2015

A mi familia

Acknowledgements

I would like to acknowledge, first of all, my supervisor Prof. Alexander Tartakovskii for his advice and support during my PhD.

Thanks to Dr. Jorge Puebla and Dr. Evgeny Chekhovich for their support and guidance during my first year.

An enormous thank you to Dr. Freddie Withers and Prof. Konstantin Novoselov from the University of Manchester for a very fruitful collaboration that resulted in important publications and two chapters of this thesis.

Thanks to Prof. Manfred Bayer and Dr. Ilya Akimov from TU Dortmund for giving me the opportunity of visiting their labs to measure PL dynamics. Thanks to Dr. Max Sich for his help and patience during measurements in D52.

I would like to thank my colleagues and friends in the LDS group, specially Stefan Schwarz, Dr. Daniel Sercombe, Robert Schofield, Dr. Tillmann Godde, Sasha Skrypka.

A big thank you to Chris Vickers and his team for their technical support and professionalism with cryogenics.

Many thanks to CONACYT and SEP for giving me the financial support for my studies.

Thanks to all my friends and housemates in Sheffield.

Thanks to my best friends in Mexico.

And finally, an enormous thank you to my mother, father, brother and sister.

Abstract

This thesis discusses optical investigations of two-dimensional metal-chalcogenide semiconductor materials and their heterostructures. Topics include a study of continuous wave (cw) and time-resolved photoluminescence (PL) of GaTe and GaSe thin films. Based on experimental evidence, we propose a model explaining the strong PL intensity decrease for thin films as a result of non-radiative carrier escape via surface states. We investigate the stability of thin films of InSe and GaSe using a combination of PL and Raman spectroscopies. By comparing signal intensities in films exposed to ambient conditions for up to 100 hours, we find notable degradation in GaSe and high stability of InSe. We continue our study with the investigation of optical properties of light emitting diodes (LED) made of van der Waals (vdW) heterostructures comprising graphene as transparent contacts, hexagonal boron nitride as tunnel barriers and transition metal dichalcogenides (TMDC), MoS₂ and WS₂, as the semiconductor active regions. Single and multiple 'quantum well' structures were fabricated with an aim to enhance the external quantum efficiency (EQE) under electrical injection. We also present PL characterisation of LEDs based on vdW heterostructures comprising WSe₂ and MoSe₂ as active layers. Temperature dependent experiments show unusual enhancement of the EQE with temperature in WSe₂ in contrast to MoSe₂, where both electroluminescence and PL are reduced with temperature. A theoretical approach to explain this behaviour is proposed, which is based on the strong spin-orbit interaction present in both materials.

List of publications

1. F. Withers, **O. Del Pozo-Zamudio**, S. Schwarz, S. Dufferwiel, P. M. Walker, T. Godde, A. P. Rooney, A. Gholinia, C. R. Woods, P. Blake, S. J. Haigh, K. Watanabe, T. Taniguchi, I. L. Aleiner, A. K. Geim, V. I. Fal'ko, A. I. Tartakovskii, K. S. Novoselov. *WSe₂ light-emitting tunneling transistors with enhanced brightness at room temperature*. Nano Letters **15** (12), 8223-8228 (2015). doi: [10.1021/acs.nanolett.5b03740](https://doi.org/10.1021/acs.nanolett.5b03740).
2. **O. Del Pozo-Zamudio**, S. Schwarz, M. Sich, I. A. Akimov, R. C. Schofield, E. A. Chekhovich, B. J. Robinson, N. D. Kay, O. Kolosov, D. N. Borisenko, N. N. Kolesnikov, and A. I. Tartakovskii. *Photoluminescence of two-dimensional GaTe and GaSe films*. 2D Materials **2** (3), 035010 (2015) doi: [10.1088/2053-1583/2/3/035010](https://doi.org/10.1088/2053-1583/2/3/035010).
3. F. Withers, **O. Del Pozo-Zamudio**, A. Mishchenko, A. P. Rooney, A. Gholinia, K. Watanabe, T. Taniguchi, S. J. Haigh, A. K. Geim, A. I. Tartakovskii, K. S. Novoselov. *Light-emitting diodes by band-structure engineering in the van der Waals heterostructures*. Nature Materials **14**, 301-306 (2015) doi: [10.1038/nmat4205](https://doi.org/10.1038/nmat4205).

In preparation:

1. **O. Del Pozo-Zamudio**, S. Schwarz, J. Klein, R. C. Schofield, E. A. Chekhovich, O. Ceylan, E. Margapoti, A. I. Dmitriev, G. V. Lashkarev, D. N. Borisenko, N. N. Kolesnikov, J. J. Finley, and A. I. Tartakovskii *Photoluminescence and Raman*

investigation of stability of InSe and GaSe thin films. Submitted to Applied Physics Letters (preprint: arxiv.org/abs/1506.05619).

2. **O. Del Pozo-Zamudio**, S. Schwarz, F. Withers, T. Godde, R. C. Schofield, A. P. Rooney, A. Gholinia, K. Watanabe, T. Taniguchi, S. J. Haigh, A. K. Geim, A. I. Tartakovskii, K. S. Novoselov. *Atomically Thin WSe₂ Light-emitting Diodes Embedded in Monolithic Microcavities.* To be submitted in Nano Letters.

Other publications:

1. S. Dufferwiel, S. Schwarz, F. Withers, A. A. P. Trichet, F. Li, M. Sich, **O. Del Pozo-Zamudio**, C. Clark, A. Nalitov, D. D. Solnyshkov, G. Malpuech, K. S. Novoselov, J. M. Smith, M. S. Skolnick, D. N. Krizhanovskii, A. I. Tartakovskii. *Exciton-polaritons in van der Waals heterostructures embedded in tunable microcavities.* Nature Communications **6**, 8579 (2015). doi: [10.1038/ncomms9579](https://doi.org/10.1038/ncomms9579)
2. D. Sercombe, S. Schwarz, **O. Del Pozo-Zamudio**, F. Liu, B. J. Robinson, E. A. Chekhovich, I. I. Tartakovskii, O. Kolosov, A. I. Tartakovskii. *Optical investigation of the natural electron doping in thin MoS₂ films deposited on dielectric substrates.* Scientific Reports **3**, 3489 (2013). doi: [10.1038/srep03489](https://doi.org/10.1038/srep03489)
3. I. J. Luxmoore, R. Toro, **O. Del Pozo-Zamudio**, N. A. Wasley, E. A. Chekhovich, A. M. Sanchez, R. Beanland, A. M. Fox, M. S. Skolnick, H. Y. Liu and A. I. Tartakovskii. *III-V quantum light source and cavity-QED on Silicon.* Scientific Reports **3**, 1239 (2013). doi: [10.1038/srep01239](https://doi.org/10.1038/srep01239)

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Chapter 1

Background and motivation

1.1 Introduction

Modern condensed matter physics have received an addition of a new branch of study in recent years: physics of two-dimensional (2D) materials [1]. This topic gained prominence with the first isolation of a single-layer of graphite, known as graphene, in 2004 [2]. Since then, the library of 2D materials has been rapidly expanding including hexagonal boron nitride (hBN), transition metal dichalcogenides (TMDCs), III-VI compounds and layered oxides [3, 4]. After an extensive research of graphene that followed the advent of this material [5], scientists started paying more attention to other 2D crystals. This has been reflected in the discovery of various novel characteristics such as mechanical [6, 7], electronic and optical [8], thermal [9], superconductive [10, 11], etc. properties. that lead to potential applications in, e.g, flexible devices [12], lasers [13] and biological studies [14]. Heterostructures based on stacks of 2D materials (known as van der Waals heterostructures) showed additional interesting characteristics due to the possibility to combine and manipulate various properties of the constituent materials [15, 16]. However, because of the novelty of the field, there is still a long way for 2D materials to reach the point where applications become a

reality. For this reason, further investigation of many aspects of material properties and new device prototypes need to be conducted. This thesis focuses on the optical properties of two different groups of two-dimensional materials: III-VI compounds and semiconducting transition metal dichalcogenides. Thin films as well as light emitting diodes based on van der Waals heterostructures made of TMDCs sandwiched with hBN and graphene were characterised. The study was performed using a combination of different optical spectroscopy techniques, including continuous and time-resolved photoluminescence, electroluminescence and Raman spectroscopy which will be described in chapter 2 together with the fabrication methods used to produce samples and devices.

1.2 The family of layered materials

A new family of layered materials, metal-chalcogenides, has recently emerged which includes semi-metals, semiconductors and superconductors. Furthermore, the idea of combining these materials in van der Waals heterostructures has been developed recently. Heterostructures have a long and fruitful history in semiconductor physics. They were used to produce electron and optical confinement, engineering of the electronic energy bands and resonant tunnelling, which led to important technologies such as lasers, solar cells and transistors. These heterostructures, with many examples among III-V semiconductor materials, are grown directly on top of each other by very clean techniques such as molecular beam epitaxy. However such growth techniques are usually limited to the lattice-matched materials. The samples produced by transferring two-dimensional films onto each other to form van der Waals heterostructures allow the combination of a wide range of materials and control of doping and thickness with atomic precision. The goal is that functional devices will be fabricated and that they are able to compete with existing technologies based on heterostructures

in a broad range of applications.

Different experimental techniques are needed in order to perform characterisation of two-dimensional materials and van der Waals heterostructures including electronic transport, optical spectroscopy, microscopy, etc.

The 2D materials studied in this work are III-VI semiconductors such as GaSe, GaTe and InSe in chapters 3 and 4; TMDCs such as MoS₂, MoSe₂, WS₂ and WSe₂ in chapters 5 and 6; and, to a lesser extent, graphene and hBN in these latter two chapters. In the following subsections a brief discussion of the properties of these materials will be presented.

1.2.1 Graphene

Graphene is an allotrope of carbon such as the well known and widely studied nanotubes and fullerenes. It consists of carbon atoms in a hexagonal lattice of only one atom thickness separated by 1.42 Å[17]. In the figure 1.1 different carbon allotropes are shown, graphene is the *honeycomb* structure in the right-bottom side of the image.

It has been intensively studied since its first isolation in 2004 by Andre Geim and Konstantin Novoselov[2].

Graphene performance in electronic devices is expected to be of extremely high quality due to its properties. In fig. 1.2 ambipolar electric field effect is observed. This allows charge carrier to be tuned continuously between electrons and holes in high concentrations (10^{13}cm^{-2}) and their mobilities μ can exceed $15\,000\text{cm}^2\text{V}^{-1}\text{s}^{-1}$ [5] as it is shown in the inset of the figure where, with changing V_g , electrons and holes concentration are tuned. This mobility has been observed to be weakly dependent on temperature that makes graphene even more impressive [19].

The ability of graphene to act as a thin and transparent contact for carrier injection is used in this work. Studies of metal-graphene junctions have proved a low resistance

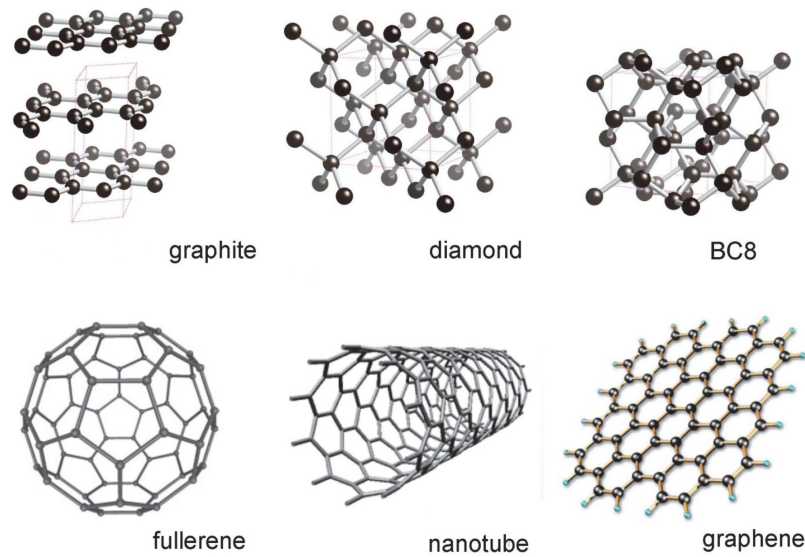


Fig. 1.1 Carbon allotropes[18]. Graphene (bottom right) has a 'honeycomb' structure with carbon atoms separated by 1.42 Å.

in the order of tens of $\Omega \mu\text{m}$ [20–22]. This makes graphene suitable for electronic device fabrication like transistors, photo-detectors or LEDs.

Synthesis of graphene

Since the beginning of extensive research of graphene in 2004 using the now famous mechanical exfoliation method (scotch tape)[2], there have been developed different ways to produce few-to-single layer graphene. Nowadays, the mechanical exfoliation technique is still offering the best quality samples for investigation[24]. The main concern is the fact that it is not viable for mass production processes. In table 1.1 a summary of four fabrication methods show in number of layers, sample size and mobility is presented. These methods are: mechanical exfoliation[23, 25], thermal decomposition of silicon carbide[26], chemical vapour deposition (CVD) on Ni[27] and CVD on Cu[28, 29]. Note that large films in the centimetre scale are already produced with high mobility, however the larger the size, the more structural defects are present in the layer.

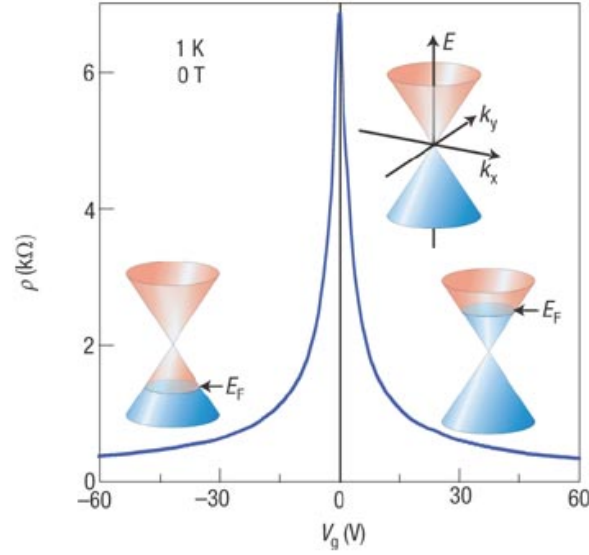


Fig. 1.2 Graphene resistivity ρ plotted as a function of gate voltage V_g [5]. The rapid decrease in resistivity when adding charge carriers indicates their high mobility, in this case $5\,000\text{ cm}^2\text{ V}^{-1}\text{ s}^{-1}$ at 1 K and 0 T. This value depends weakly with temperature. The insets show its conical low-energy spectrum $E(k)$, showing changes in the position of the Fermi energy E_F with changing gate voltage V_g . Positive V_g induce electrons and negative V_g induce holes [2, 3, 23].

As it will be discussed in the chapters 2,5 and 6, the graphene layers used in this work were exfoliated from bulk graphite using the scotch tape method and transferred onto different layers to form heterostructures. It is used as transparent electrodes in LEDs for carrier injection thanks to the properties of ambipolar field effect and high conduction discussed previously.

1.2.2 Boron Nitride

Boron nitride (BN) is a chemical compound that consist on equal number of boron and nitrogen atoms. It exists in different crystalline forms with the hexagonal (hBN) being the most studied one. Like TMDCs, it is widely used as a lubricant in industrial applications [30]. Similar to previously discussed materials, weak interlayer bonding is present in hBN so exfoliation methods can be implemented. Due to its wide band

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Method	Layers	Size	Mobility ($\text{cm}^2\text{V}^{-1}\text{s}^{-1}$)
Exfoliation	1 to 10+	1 mm	15000
Thermal SiC	1 to 4	50 μm	2000
Ni-CVD	1 to 4	1 cm	3700
Cu-CVD	1	65 cm	16000

Table 1.1 Comparison of graphene synthesis methods. Number of layers, sample size and mobility are shown [24]

gap of ~ 6 eV [31], dielectric properties of this material are very attractive.

In fig. 1.3 a schematic of a single layer of hBN is shown. B and N atoms (purple and yellow spheres, respectively) are arranged in a 'honeycomb' structure. Its lattice constant of 2.5 Å is 1.8 % larger than in graphene [32, 33].

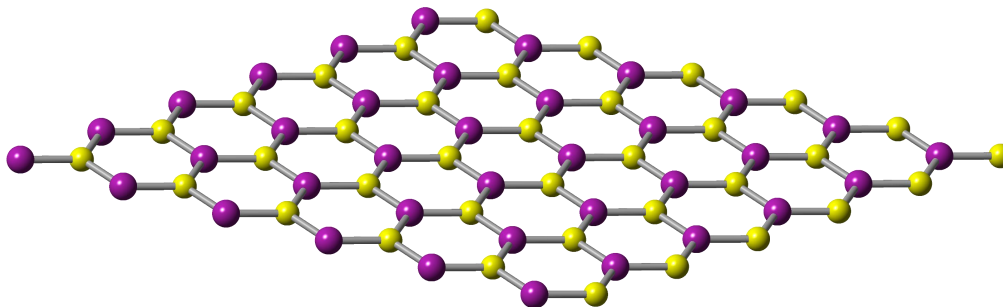


Fig. 1.3 Hexagonal boron nitride single layer. Boron (purple) and Nitrogen (yellow) atoms are arranged in a *honeycomb* structure separated by 2.5 Å.

An important property of hBN is its transparency in a wide range of wavelengths [34] so it has been recently reported to be an ideal layered material for encapsulation and capping of other 2D materials to protect them from oxidation or other atmospheric effects that could produce damage [35] when they are unstable in ambient conditions. In a similar way, it has been proved to be an excellent substrate for graphene electronics [36] reducing carrier inhomogeneity and intrinsic doping [37] in comparison with SiO_2/Si (the most common substrate).

Synthesis of hBN layers

Comparable to other 2D materials, hBN synthesis has been widely investigated resulting in methods similar to those used for graphene. Mechanical exfoliation is, again, the most common and pure method to produce high quality samples with typical sizes of 10-50 μm [38]. Moreover CVD [39] and epitaxial [40] grown hBN has been shown paving the way to the production of samples on the industrial scales.

In this work, hBN is mechanically exfoliated from bulk and used for encapsulation of devices and as a barrier for the quantum well structures (from single to few layers). This will be discussed in detail in chapters 5 and 6.

1.2.3 III-VI semiconductors

III-VI semiconductors are compounds of the form AB where A is an element from the group III of the periodic table and B from the group VI. These compounds are layered crystals with strong covalent in-plane inter-atomic bonding and weaker van der Waals inter-plane bonding[41]. Bulk crystals are stacks of a hexagonal tetralayers in-plane structure of the form $B-A-A-B$ ($A=\text{Ga, In}$; $B=\text{S, Se, Te}$). For the wider studied GaSe, several types of stacking exist leading to different polytypes[42–44]. For GaTe having a monoclinic crystal lattice [45, 46] the polytypic behavior has not been observed [46, 47]. In fig. 1.4 the crystal structure of III-VI compound is presented (III atoms shown in purple, VI in blue): in 1.4A it is seen from the top and the hexagonal lattice is observed. In 1.4B it is viewed from the side where group-III atoms are sandwiched by group-VI atoms, and in 1.4C it is seen from an angle. Table 1.2 presents the structural parameters and band gap energies of the three III-VI semiconductors used in this work.

In the past decades, III-VI compounds attracted attention because of their non-linear optical properties. Some of the most important is their capability as second

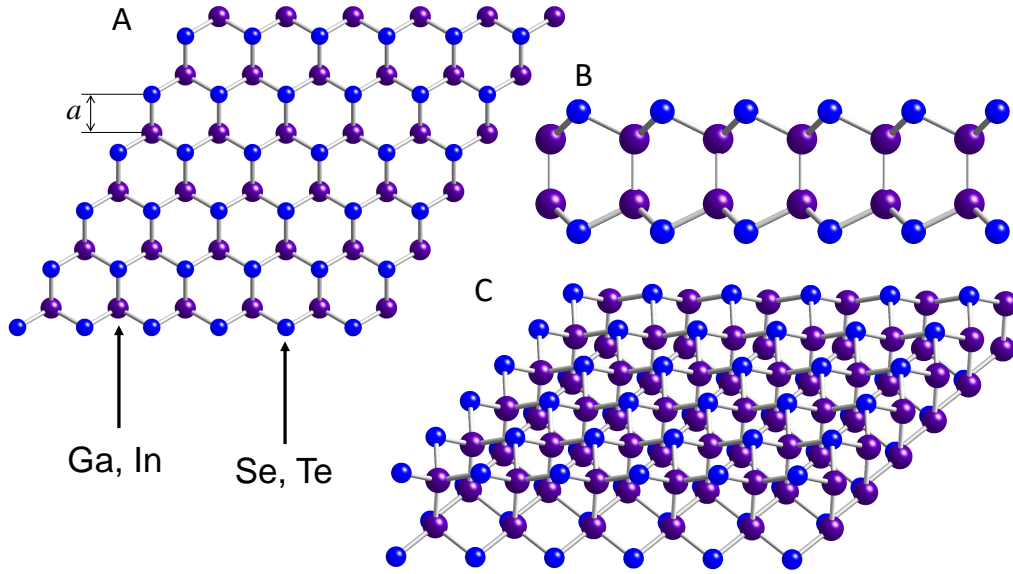


Fig. 1.4 *III – VI* crystal structure of a single layer viewed from (A) top, (B) side and (C) an angle to observe the layered geometry. Purple balls represent atoms of a group-III material, i.e. Ga, In. Blue balls are atoms of a group-VI material, i.e. Se, Te. See table 1.2 for structural parameters a and c .

Compound	a (Å)	c (Å)	E_g (eV)
GaSe	3.755	15.95	2.1
GaTe	4.06	16.90	1.7
InSe	4.00	16.70	1.3

Table 1.2 Structural parameters and band gap energies of III-VI semiconductors[41].

harmonic generation (SHG) materials [48, 49] and also far infra-red properties in the THz range [50].

Properties of sub-micron thick films have been reported only recently. GaSe, GaS and GaTe thin films were investigated for photodetector and sensor applications [51–53], flexible electronics[54] and phototransistors [55]. InSe suggests interesting potential for solar cells [56, 57], room temperature electroluminescence [58] and shows tuning of band gap with film thickness [59, 60]. These studies open up new possibilities for 2D III-VI films in nano- and opto-electronics.

Furthermore, in contrast to molybdenum and tungsten chalcogenides emitting light efficiently only in films with a single unit cell thickness, as it will be discussed later in this chapter, III-VI materials are bright light emitters in a range of thicknesses[59, 60]. This may relax the stringent fabrication requirements and add flexibility for the novel heterostructured devices such as light emitting diodes.

Synthesis of III-VI layers

In the previous section, it was mentioned that III-VI semiconductor materials have not been studied extensively in their layered form. As a consequence, only few reports of different methods of synthesis can be found: mechanical exfoliation, used for graphene in the studies that started the study of 2D materials [2] was employed in the case of InSe[59, 60], GaSe[53] and GaTe [55] and epitaxy was used for GaSe[54].

3D crystals of Gallium-based III-VI materials are usually grown by high-pressure vertical zone melting in graphite crucibles under Ar pressure[61–64]. The gallium mono-chalcogenides used in this work were synthesized from high-purity materials: Ga and Te - 99.9999 %, Se - 99.9995 %. InSe single crystals were grown by the Bridgman-Stockbarger method from a preliminarily synthesized ingot [65].

The thin film samples studied in this work were produced using mechanical exfoliation from bulk materials onto SiO₂/Si substrates. This method will be explained in detail in the next chapter.

1.2.4 Transition metal dichalcogenides

Transition metal dichalcogenides (TMDCs) are a family of compounds of the form MX_2 , where M a transition metal (Mo, W, etc.) and X a chalcogen atom (S, Se or Te). These compounds were largely studied in their bulk form five decades ago in material science and engineering that led to important industrial applications as

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solid lubricants and surface protectors[66–68]. New properties of these materials were discovered when they were produced in monolayer form [3]. The crystal structures of these compounds are very similar to each other and consists of a M atom sandwiched by two X atoms. In fig. 1.5 a schematic of a single layer of a MX_2 single-layer is shown from three different angles. X atoms are presented in yellow and M in blue. In 1.5A it is seen from top, where the hexagonal structure is evident. In 1.5B the single layer is viewed from the side, showing the atomic structure of M atoms sandwiched by two X atoms and in 1.5C the single layer is viewed from a 3D perspective. Bulk crystals are formed by vertical stacking of these single layers separated by the half of the vertical lattice constant c . Table 1.3 shows the structural parameters of the four TMDCs used in this work.

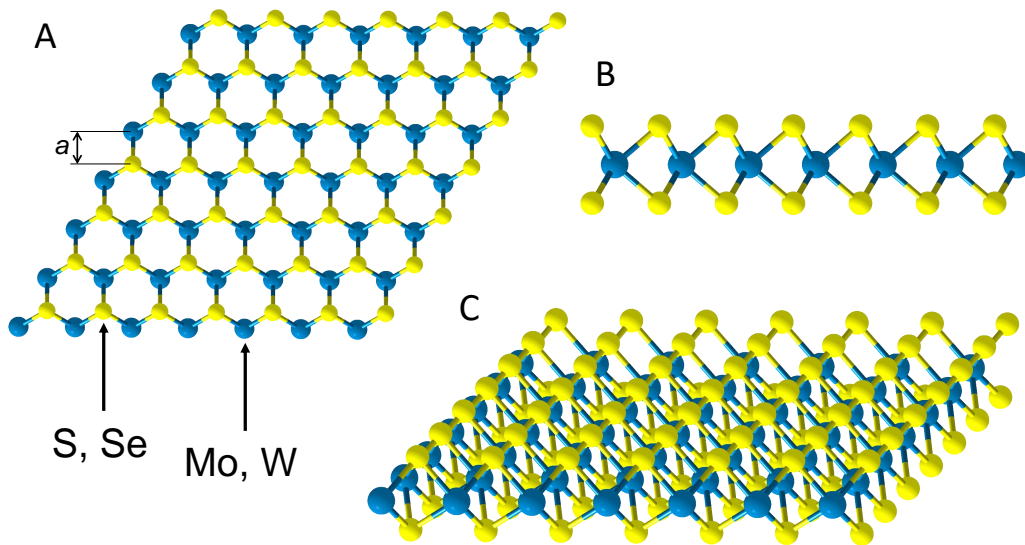


Fig. 1.5 Single layer of a MX_2 single layer viewed from three different angles: (a) Top (b) side and (c) 3D perspective. M atoms are shown in blue and X in yellow. See table 1.3 for parameters of structural parameters of various TMDCs.

Some of the most studied TMDCs in the 2D research such as MoS_2 , MoSe_2 , WS_2 and WSe_2 received much attention in the last few years due to important transformation in their electronic band structure in few monolayer thick films. These are

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Compound	a (Å)	c (Å)	E_g (eV)	Δ_{SO} (meV)	Δ_{cb} (meV)	E_b (meV)	E_c (meV)
MoS ₂	3.18	12.296	1.67	146	3	570	18
MoSe ₂	3.32	12.939	1.44	183	22	550	30
WS ₂	3.19	12.349	1.81	425	32	700	20-40
WSe ₂	3.32	12.976	1.55	461	37	600	30

Table 1.3 Structural parameters, band gap energies E_g , spin-orbit splitting of the valence band Δ_{SO} and the conduction band Δ_{cb} , exciton E_b and trion E_c binding energies of TMDCs used in this work [72–76].

semiconductor materials with an indirect band gap in the bulk form. When they are thinned down to single layer, a transition to a direct band gap occurs[69, 70]. Excitonic effects govern their optical properties[71]. The different band gap energies E_g for the four TMDCs mentioned above are shown in table 1.3.

The electronic structure of TMDCs has been calculated by several theory groups to show the above mentioned transition from indirect to direct band gap semiconductor [70, 72, 77]. In fig. 1.6 a calculated band structure of MoS₂ using density functional theory is presented for (a) bulk, (b) quadrilayer, (c) bilayer and (d) monolayer reported in [70]. In this case the indirect bandgap occurs in the Γ point for bulk and the transition to direct bandgap occurs in the K point. Similar behaviour of the band is present in others TMDCs such as MoSe₂, WS₂ and, more ambiguously, in WSe₂ as was shown in one of the studies [73]. For the latter case, a special investigation of the band structure is discussed in Chapter 6.

Both optical and electronic remarkable properties have been observed in monolayers. Field effect transistors (FETs) made of MoS₂ due to high mobility [78], flexible electronic devices [79] exploit strain tuning effects[72] and solar cells [80] are some examples of the research reported recently.

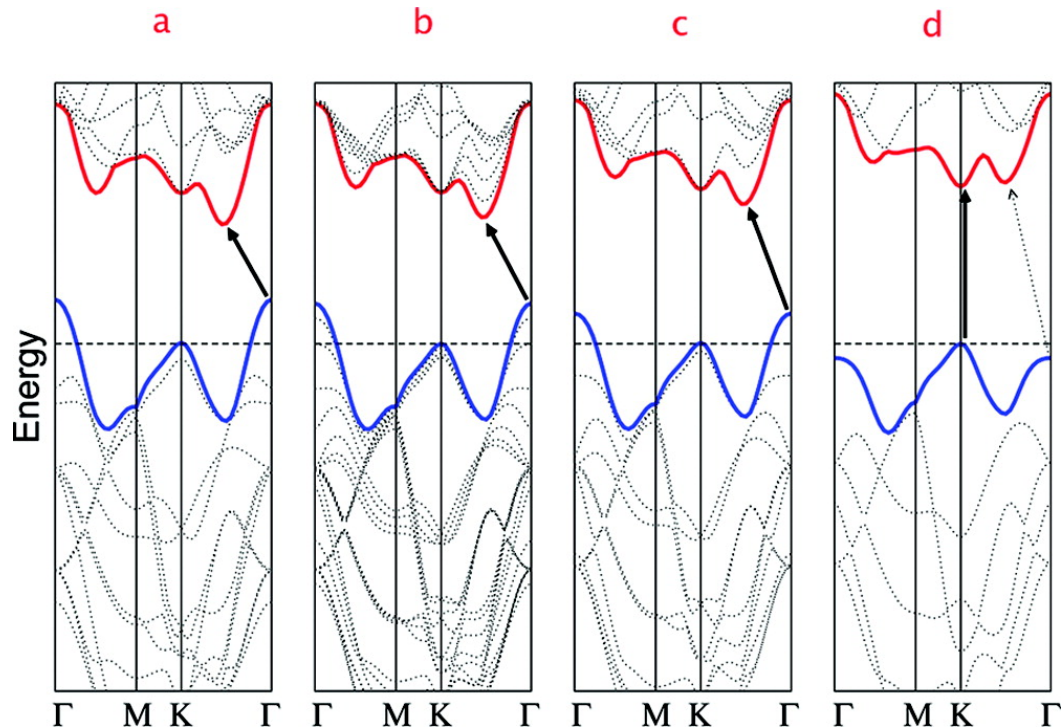


Fig. 1.6 Electronic band structure of (a) bulk (b) quadrilayer (c) bilayer and (d) single layer MoS₂. The transition from indirect to direct band gap is observed [70].

Excitonic properties of TMDCs

In semiconductor physics, an essential quasiparticle called exciton that consists of an electron-hole pair is formed when an electron is excited from the valence band to the conduction band [81]. These quasiparticles can be seen as hydrogen-like atoms due to the Coulomb force that binds the electron and hole. The simplest exciton, formed just by one electron and one hole, has no charge (neutral) and is usually denoted X^0 as it is illustrated in fig. 1.7(a). When more than one electrons or holes form the quasiparticle, they are called charged excitons or trions and denoted X^- , X^+ , depending on the charge state.

In TMDC monolayers, excitons are strongly influenced by the strong spin-orbit coupling producing splitting (Δ_{SO}) in the valence band [82]. This results in the formation of two excitonic transitions, usually referred to as A and B , associated to direct

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optical transitions between band edges [8] at the K points, this can be observed in fig. 1.7(b) where the transitions are still depicted in the single-particle band structure representation with no excitonic effects shown. The giant magnitude of Δ_{SO} in ultrathin TMDC layers is predicted to be from ~ 140 to 460 meV (see table 1.3)[82] and experimentally observable both from PL and absorption spectra like in the case of MoSe₂ in fig. 1.7(c). In general, excitation with light generates free neutral excitons on the TMDCs and, in the case of doped layers, trions can be created as well when $e - h$ pairs bind to additional charge. The exciton binding energies E_b have been predicted by first-principles calculations [83, 84] to be extraordinarily large on the order of hundreds of meV (see table 1.3). Different experimental results have demonstrated these magnitudes including reflection spectra [85], two-photon absorption [86] and scanning tunnelling microscopy [87]. The strong binding energy emerges from the 2D geometry and the large effective masses of electrons and holes [84]. In comparison with III-V semiconductors, the binding energy is two orders of magnitude larger in 2D TMDCs. For example, in GaAs it is ~ 5 meV with an effective electron mass $m_e = 0.063m_0$. For MoS₂ the binding energy is 146 meV with $m_e = 0.37m_0$. It is clear that the value of the electron effective masses play a very important roll here as they differ significantly in both materials. It is worth to compare the exciton Bohr radius a_B (distance between electron and hole) of both types of materials as well. In TMDCs $a_B \sim 1$ nm [84], one order of magnitude smaller than in GaAs. Since the a_B is only ~ 3 times the unit cell dimensions a , the excitons in 2D TMDCs belong to a case in between a typical Wannier-Mott exciton and a Frenkel exciton[88], where a_B is larger than the lattice constant a .

Properties of trions can also be understood from optical spectra. For example, from the energy splitting in PL spectra as shown in 1.7(c), the difference between X^- or X^+ and the neutral exciton X^0 is called trion binding energy E_c and corresponds

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to the energy that binds the additional charge to the exciton. PL experiments show clearly separated X^0 and X^- peaks in TMDCs and range from ~ 20 to 40 meV [75] (see table 1.3). In the inset of 1.7(c) both X^0 and X^- from monolayer MoSe₂ are clearly observable with a separation of 30 meV [89].

Recent theoretical studies have shown that spin-orbit splitting is also relatively large in the conduction band for some TMDC monolayers and the spin ordering affects the excitonic dark and bright states [76]. These effects have not been studied experimentally in detail due to the fact that for MoS₂, the most investigated TMDC, the splitting is of the order of a few meV. In table 1.3 the calculated spin-orbit splitting of the conduction band Δ_{cb} for the materials used in this work are presented. In chapter 6 a temperature dependent study of the electroluminescence of MoSe₂ and WSe₂ will be presented and the effects are mainly explained as a result of their large Δ_{cb} .

Another important property of excitons is their spin. It depends on both electron (\uparrow or \downarrow) and hole (\uparrow or \downarrow) spins and play a key role when a recombination that leads to emission of a photon occurs. For conservation of angular momentum, an exciton should have total spin $S=0$, this is the case when electron and hole have opposite spins ($\uparrow\downarrow$ or $\downarrow\uparrow$), and will be a bright state. For $S=1$ (same spin, $\uparrow\uparrow$ or $\downarrow\downarrow$), their direct recombination is forbidden as momentum conservation cannot be satisfied and the exciton is therefore dark. Optical interband transitions always conserve the spin. Therefore, only the bright exciton can radiatively recombine and result in a photon emission, as the name itself implies. Bright excitons are directly observable from the PL spectra[75].

The excitonic properties of the TMDCs are very important in the study presented in chapters 5 and 6 where LEDs made of these materials are studied and PL and EL spectra are described in terms of excitonic effects.

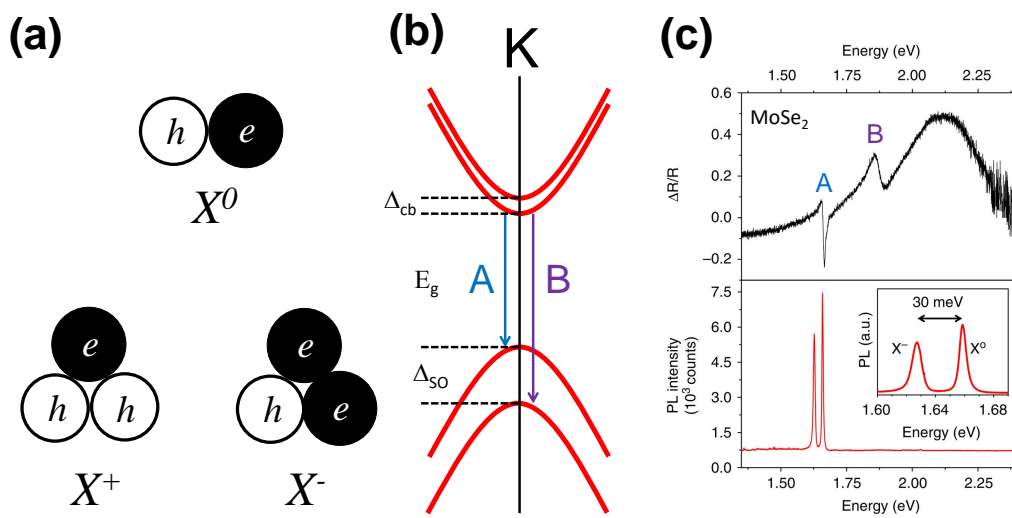


Fig. 1.7 Excitons in TMDCs monolayers. (a) Diagrams of excitons X^0 and trions X^+ , X^- . (b) Diagram of the band structure around the K point of TMDC monolayers. The excitonic transitions A and B due to Δ_{SO} are observed in blue and purple, respectively. (c) Differential reflectance spectra (top) of monolayer MoSe₂ at 20 K shows A and B excitons. X^0 and X^- are observed in the PL spectra (bottom). Inset: zoom to the PL peaks shows a X^- charging energy of 30 meV[89].

Synthesis of TMDCs layers

Mechanical cleaving or exfoliation used for production of graphene [2] has been extended to TMDCs. It is possible to obtain clean high quality samples, but the production rate is generally low and is not suitable for technological applications. As a consequence, further fabrication methods emerged that in principle could be used for a large scale production of TMDCs. This can be obtained, for example, by using chemical vapour deposition (CVD), a common technique used for decades to produce thin films of a large variety of composites including silicon, nitrides, carbon nanotubes and high-k dielectrics [90–92]. In recent years, reports of different CVD grown TMDCs like MoS₂[93–95], WS₂ [95, 96], WSe₂[97, 98] and MoSe₂[99, 100] have been published. The properties that all these works share is that crystals grow in a form of triangles due to hexagonal structure (see fig. 1.5). The layers can be grown on or transferred onto a variety of substrates such as sapphire, quartz, gold or SiO₂ in film size of hundreds of micrometres to a millimetre scale [93–100].

Another elaborated technique is liquid exfoliation, which uses ion intercalation between atomic layers in a liquid environment to split them from each other [101–103]. This process results in a liquid solution that can be span or sprayed onto a substrate producing a large area covered by single layers of the material.

In the next chapter a detailed description of the mechanical exfoliation method will be presented since it was the technique used in this work to produce TMDCs samples.

1.2.5 van der Waals heterostructures

One of the properties that all the materials described previously in this chapter share is the weak interlayer bonding in their bulk form. Due to this, it is natural to think of the stacking of layers of different materials. A typical analogy, proposed by Geim [15], is to imagine the atomic layers as LEGO blocks. When various blocks are stacked on

1.2 The family of layered materials

top of each other, they get strongly attached and form a more complex structure. The strong covalent bonding of the two-dimensional materials act in a similar way. The possibility of making multilayer structures forming new artificial materials has been extensively studied recently [71, 104].

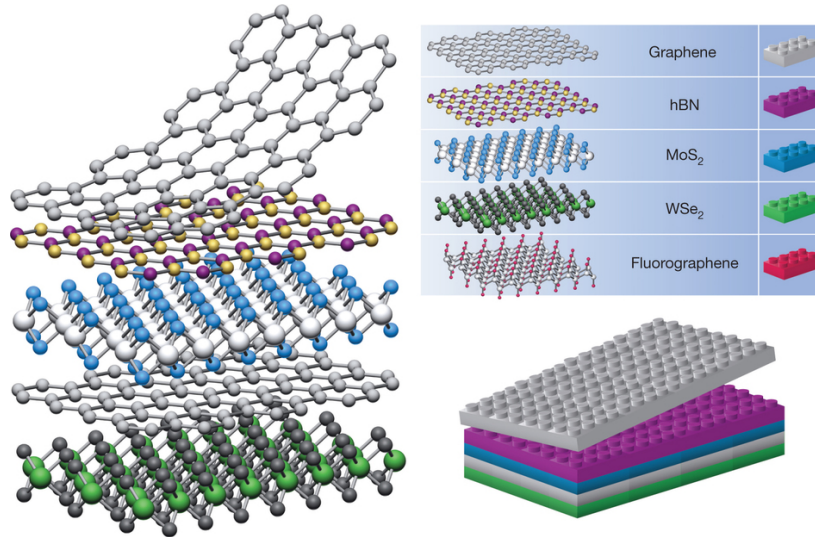


Fig. 1.8 Analogous representation of van der Waals heterostructures with LEGO blocks. The stacking of different classes of layered materials like graphene, MoS₂, hBN, etc. to build more complex structures is possible[15].

Among the reports on vdW heterostructures are the fabrication of field effect tunnelling transistors based on hBN and graphene [104], demonstration of tunnelling diodes with negative differential resistance[105] and photovoltaic devices based on MoS₂-graphene [106] and WS₂-graphene [107]. In terms of optical studies, long-lived interlayer excitons in MoSe₂-WSe₂ heterostructures[108] and LEDs [109] have been demonstrated.

Fabrication of van der Waals heterostructures

In section 2.1.2 the most common method for van der Waals heterostructure will be presented, namely so-called 'peel/lift transfer'. This technique has been the base of

References

almost all the studies mentioned above and consists of the mechanical exfoliation of the desired layers followed by the transfer and stacking on top of each other [110, 111]. In this work LEDs based on vdW heterostructures made of graphene, hBN and TMDCs are built with this method and they will be presented in chapters 5 and 6. These devices were fabricated in laboratories of The University of Manchester by Dr. F. Withers.

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Chapter 2

Experimental Techniques

2.1 Sample fabrication methods

As discussed in the previous chapter, layered materials such as III-VI semiconductors and TMDCs show new properties when they are thinned down to single or few layers. The key techniques used in this work are mechanical exfoliation and peel/lift transfer. A detailed description of both fabrication methods will be presented in the following sections.

2.1.1 Mechanical exfoliation

The most common technique to produce high quality single layers of two-dimensional materials is referred as mechanical exfoliation (or the 'scotch-tape' method). It was demonstrated by Novoselov et al in 2004 in the seminal article of graphene exfoliated from graphite[1]. It is, surprisingly, a rather straightforward procedure that consists of the use of sticky tape to peel layers off bulk crystals[2] as it is schematically shown in Fig. 2.1. The first step is to press a strip of the tape onto a piece of the material (a), then peel it off (b) and press it on a substrate (c) that usually is Si finished with

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SiO₂. This process can be repeated (d) and, eventually, single layers of the crystal will be formed with different dimensions from few microns up to a millimetre in the case of graphene[3] and tens to hundreds of microns for hBN and TMDCs [4].

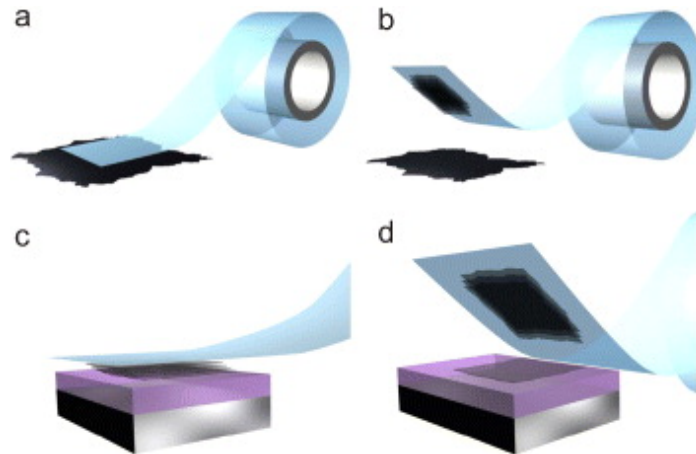


Fig. 2.1 Mechanical exfoliation [2]. (a) Sticky tape is pressed onto a crystal. (b) When the tape is peeled off, a few number of layers get attached to the tape. (c) Then it is pressed against a substrate (SiO₂/Si for example). (d) Repeating the process in order to get few to single layers.

Once the mechanical exfoliation procedure is carried out, the challenge is to locate the single layers on the substrate that, due to their low thickness, could be invisible even in the microscope. However, studies showed that the contrast of the films can be increased when the appropriate thickness of SiO₂ on Si is chosen[5–7]. It was found that an optimal thickness of 250-300 nm of SiO₂ on Si is needed to visualize thin films of TMDCs in the microscope, showing differing contrast colour depending on the TMDC layers thickness down to a single layer. To illustrate this, in fig. 2.2 an optical image of a MoSe₂ single layer surrounded by thicker flakes is presented. Note how the film colour varies from flake to flake. For more accuracy of the thickness, once the target films are localized, atomic force microscopy (AFM) is commonly used. More complex techniques to establish the film thickness have been employed. One example is Raman spectroscopy [8–10] where the separation of Raman modes depends

on the film thickness. However, this is not entirely reliable as Raman spectroscopy is sensitive to strain and charging. Another spectroscopy technique is photoluminescence when a transition from indirect to direct band gap in the material occurs, as it will be discussed later in this chapter.

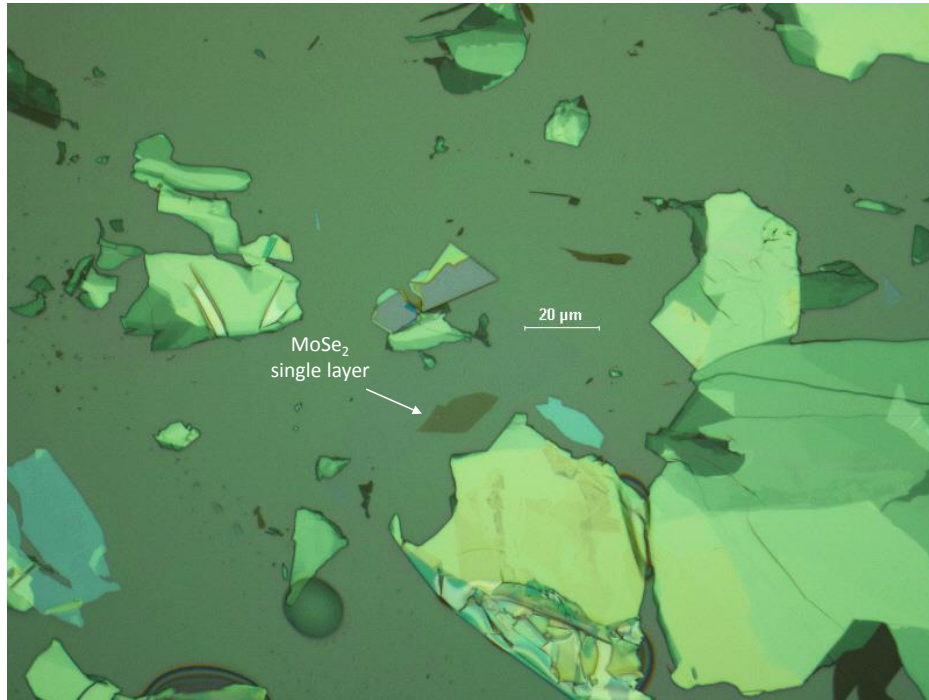


Fig. 2.2 Optical image of a mechanical exfoliated MoSe₂ single layer surrounded by thicker flakes on a SiO₂/Si substrate. The film colour indicates difference in thickness.

2.1.2 Heterostructure fabrication

In chapters 5 and 6 an optical study of light-emitting diodes based on quantum wells (QW) made of van der Waals heterostructures will be presented. These heterostructures were produced via multiple 'peel' and 'lift' transfer processes described below [11, 12].

An example of a single quantum well structure is hBN/Gr_B/2hBN/MoS₂/2hBN/Gr_T/hBN. This requires stacking of single or few layers of the materials. Fig. 2.3A-F and Fig.

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2.4 show the schematic for the processes. For the 'peel' process a flake is mechanically exfoliated (Fig. 2.3A), as described in section 2.1.1, onto a poly(methyl methacrylate) (PMMA) double layer. Then the bottom polymer is dissolved releasing the membrane which then floats on top of the liquid as is schematically shown in fig. 2.3B,C. The PMMA membrane is then inverted and aligned onto the target crystal, Fig. 2.3D. The two crystals are brought into contact (Fig. 2.3E) and heated until the PMMA adheres to the target substrate. Once the flake has adequately stuck to the target crystal the PMMA membrane is brought back, fig. 2.3F. The flake due to the strong van der Waals interaction peels from the PMMA onto the target flake.

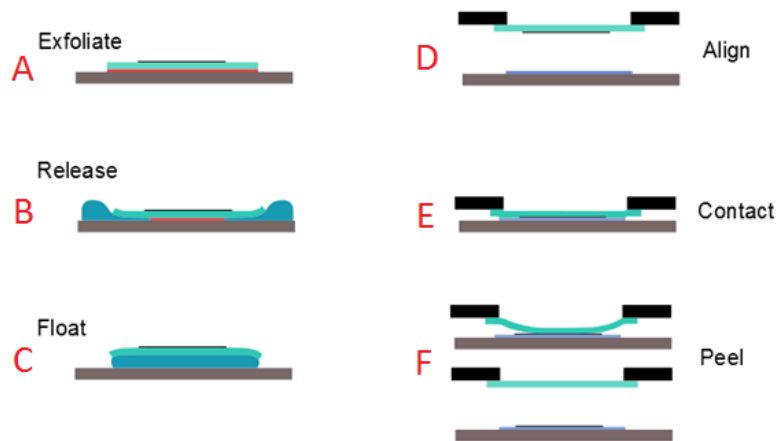


Fig. 2.3 Schematic procedure for the 'peel' process. (A) A flake is mechanical exfoliated from bulk on PMMA, (B) and released from this polymer (C) which then floats on liquid. (D) The flake is inverted and aligned onto the target crystal, (E) and then brought into contact. (F) The flake gets stacked in the substrate when peeling the PMMA.

For the 'lift' process the membrane is produced in the same way as shown in Fig. 2.3. Instead of peeling the flake onto the target crystal, a large flake on the membrane is used to collect a smaller flake on the substrate. The flake to be lifted is exfoliated onto a second SiO_2/Si wafer.

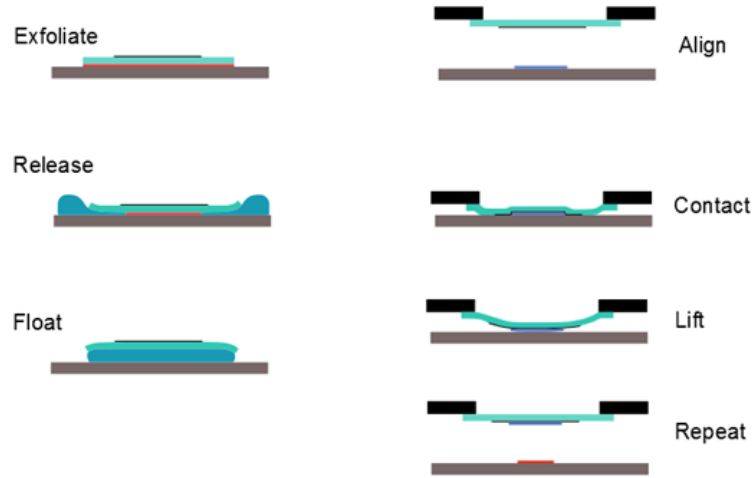


Fig. 2.4 Schematic procedure for the 'lift' process. The septa are similar to the ones described in 2.3. Instead of peeling the flake onto the target crystal, a large flake on the membrane is used to collect a smaller flake on the substrate. The flake to be lifted is exfoliated onto a second SiO_2/Si wafer.

In this work multiple QW heterostructures are also studied, such as the double QW $\text{hBN}/\text{Gr}_B/2\text{hBN}/\text{MoS}_2/2\text{hBN}/\text{MoS}_2/2\text{hBN}/\text{Gr}_T/\text{hBN}$. Fig. 2.5 shows the fabrication route for the multiple QW (MQW) structure. Firstly a graphene flake is peeled from a PMMA membrane to an hBN crystal on the Si/SiO_2 substrate, Fig. 2.5A. After this a thin hBN tunnel barrier is peeled from the PMMA membrane onto the hBN- Gr_B structure, Fig. 2.5B. A thin hBN spacer carrying a single layer TMDC crystal (lifted from a second substrate) is then peeled from the membrane, thus completing the first well, Fig. 2.5C. This process can be repeated as shown in Fig. 2.5D to produce a double QW and even further to produce multiple QW structures.

2.2 Optical techniques

In the next four experimental chapters, the main subject will be the study of optical properties of the materials previously discussed. Two essential physical phenomena in

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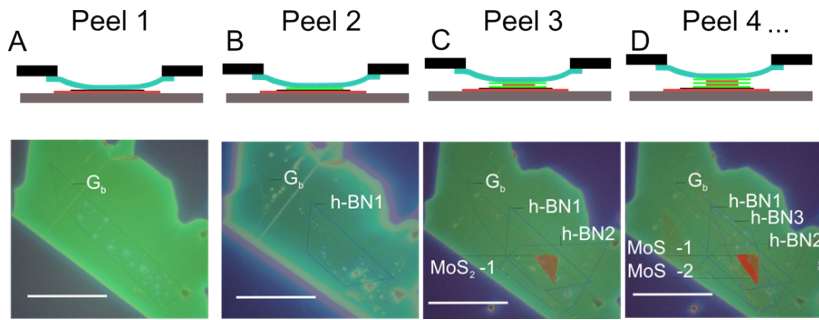


Fig. 2.5 (A-D) Schematic and differential interference contrast microscope images with semi-transparent dark field images overlaid to highlight flake edges, for the multiple QW structures. Scale bar is $50 \mu\text{m}$.

these chapters will be photoluminescence and electroluminescence. In the next sections a description and experimental techniques employed to measure these phenomena will be discussed.

2.2.1 Luminescence in solids

Luminescence is the emission of light in materials that is not a result of heating (called incandescence). This emission can be caused by external excitations with light or an electrical current[13].

Fig. 2.6 shows an schematic of the luminescence process in a solid. We can consider two bands with a relatively large gap between them. Electrons (holes) are injected into the conduction (valence) band and relax to the lowest available level. The photon is emitted when an electron from the conduction band undergoes a transition into an empty (hole) state in the ground valence band. These empty state are generated by the injection of holes.

The spontaneous emission rate for a two level system where N is the population

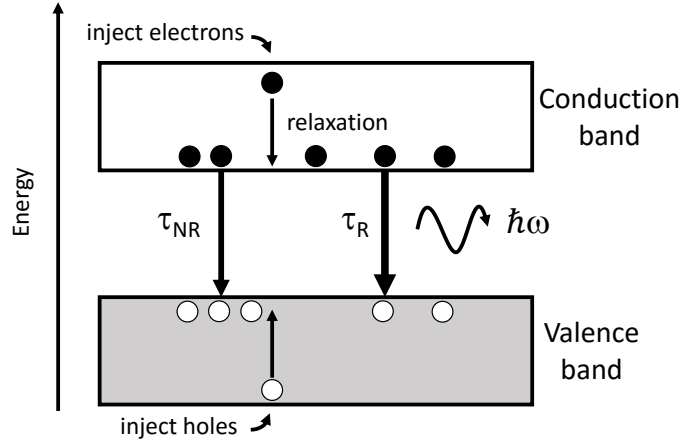


Fig. 2.6 Schematic of luminescence process in a solid. Electrons (holes) are injected into the conduction (valence) band before recombination that produces the emission of a photon.

in the upper level at time t :

$$\left(\frac{dN}{dt}\right)_{\text{radiative}} = -AN \quad (2.1)$$

This spontaneous emission is ruled by the Einstein coefficient A and the rate equation can be solved to give:

$$N(t) = N(0)\exp(-At) = N(0)\exp(-t/\tau_R) \quad (2.2)$$

where $\tau_R = A^{-1}$ is the radiative lifetime of the transition. A is proportional to the other Einstein coefficient B that determines the absorption probability:

$$A = \frac{8\pi h\nu^3}{c^3}B \quad (2.3)$$

Hence, if the transitions have large absorption coefficients, they will have high emission probability and short radiative lifetime. However, according to eq. 2.1, even if there is high absorption but low population N in the upper level, emission intensity

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may be low, for example, due to strong non-radiative processes

The electrons relax to within $\sim k_B/T$ of the conduction band. The holes follow a similar series of relaxations. Thus light is only emitted within a narrow energy range, which is further modified by strong excitonic effects.

An important factor that needs to be taken into account is that electrons excited in the conduction band can relax into the valence band by emitting phonons or be trapped by defects. As no photons are emitted in this process, this is called non-radiative relaxation. Eq. 2.1 needs to be rewritten by adding a new non-radiative lifetime τ_{NR} :

$$\left(\frac{dN}{dt}\right)_{total} = -\frac{N}{\tau_R} - \frac{N}{\tau_{NR}} = -N\left(\frac{1}{\tau_R} + \frac{1}{\tau_{NR}}\right) \quad (2.4)$$

so the total τ will be:

$$\frac{1}{\tau} = \frac{1}{\tau_R} + \frac{1}{\tau_{NR}} \quad (2.5)$$

Let's define the term η_R as the luminescence efficiency given by the ratio of the radiative emission rate to the total de-excitation rate:

$$\eta_R = \frac{N/\tau_R}{N(1/\tau_R + 1/\tau_{NR})} = \frac{1}{1 + \tau_R/\tau_{NR}} \quad (2.6)$$

If $\tau_R \ll \tau_{NR}$, $\eta_R \approx 1$, the maximum possible light is emitted. If $\tau_R \gg \tau_{NR}$, $\eta_R \approx 0$, light emission is very inefficient. Therefore radiative lifetime should be much shorter than non-radiative lifetime in order to achieve efficient luminescence.

For the luminescence process described above, if the excitation takes place with light, it is called photoluminescence. If it is with electrical current, electroluminescence. We will describe both in the following subsections together with the spectroscopy techniques used to measure them.

Photoluminescence (PL) is the emission of a photon created due to a recombination of an electron-hole pair that is optically created in the first step (see fig. 2.7). This occurs when the excitation photon energy is greater than the band gap (E_g) of the material, and an electron is then excited from the valence to the conduction band. Electrons and holes relax to the bottom of their bands and recombine leading to the creation of a photon with energy around the band gap E_g [13] (excitonic effects are neglected here). This case, illustrated in fig. 2.7a occurs in materials with direct band gap, where the bottom of the conduction band and the top of the valence band are observed for the same k -vector in the Brillouin zone. Fig. 2.7b shows an opposite case, where they occur for different k -vectors, this leads to an indirect band gap where a phonon assisted transition is needed for the radiative recombination process and, therefore, the efficiency is low. For this reason, as it will be discussed in the next section, commercial applications for luminescence devices are made of direct band gap semiconductors.

It was mentioned above that PL intensity depends on the absorption of photons. This process can be described as a function of thickness in a finite material as the intensity of the light $I(h)$ travelling across in the form $I(h) = I_0 e^{-\alpha h}$, where I_0 is the intensity of the light before enters the material, α is the absorption coefficient and h the thickness. In chapter 3 PL intensity will be described as a function of film thickness by using this equation.

A common term to calculate the efficiency of the PL process is the Quantum Yield (QY) defined as:

$$QY \equiv \frac{\# \text{ of photons emitted}}{\# \text{ of photons absorbed}} \quad (2.7)$$

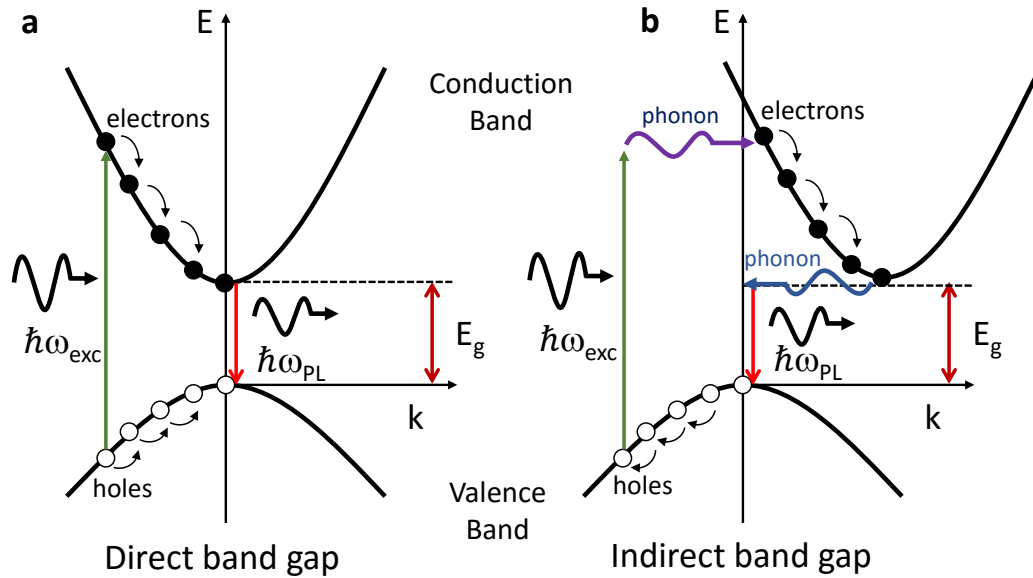


Fig. 2.7 Diagram of photoluminescence processes in semiconductor materials. (a) In a direct band gap material the top of the valence band coincide with the bottom of the conduction band. When a photon with an energy $\hbar\omega_{exc}$ excites electrons and holes, after relaxation of electrons (holes) to the bottom (top) of the band, a photon with energy $E_g = \hbar\omega_{PL}$ is emitted. (b) For indirect band gap materials, the bottom of the conduction band and top of the valence band are at different k -vectors. Here, a phonon assisted transition is needed, after relaxation a photon and phonon are emitted.

Photoluminescence measurements

In order to detect PL we use a set-up schematically shown in fig. 2.8. A diode laser used as the excitation source, typically with a wavelength of 532 nm. The laser beam (green solid line) is guided with a mirror to a 50-50 beam-splitter, 50% of the light goes to a power meter and the rest to a 50x microscope objective with a 25 mm working distance that focuses the beam onto the target sample achieving a laser spot of ~ 1 -2 μm diameter with a power range from ~ 10 nW to 25 mW. The light emitted by the sample (red dashed line) is collected with the same objective (NA=0.55) and guided to a slit of a 0.5 m spectrometer, where a 4 cm focal length lens is used for focussing light on the entrance slit of the spectrometer. Additional devices such as a white light

source and a CMOS camera in the setup are necessary for visualizing the sample. The spectrometer is equipped with 3 different gratings: 300, 600 and 1200 lines/mm for achieving different spectral resolution and a nitrogen cooled CCD camera with a chip of 100x1240 pixels. The sample is placed in vacuum in a cold finger inside a continuous flow He cryostat that allows to cool down to ~ 6 K and achieve controllable temperatures up to room-T with a heater. Heavy duty XY stages are used to move the entire cryostat with an accuracy of $\sim 1 \mu\text{m}$ in order to align different parts of the sample under the excitation laser.

Electroluminescence

Electroluminescence (EL) is the generation of light in response to an electric current passing through a material[13]. As in the case of PL, EL is generated in an active region due to a radiative recombination of electrons and holes.

The first observation of the phenomenon of EL was on silicon carbide (SiC) in 1907 by H. J. Round in the Marconi Labs[14]. The first light-emitting diode (LED) was developed using the same material a few years later in 1923 [15]. Since then, LEDs have been rapidly developing in a large variety of ways and nowadays have many applications such as smart displays, efficient lighting, traffic lights, car lights, etc. This made solid-state LEDs a key component of today's technology [16]. Different materials and structures have been used for design and fabrication such as nitrides (GaN, InGaN)[17], porous silicon[18], polymers[19], carbon nanotubes[20], quantum dots[21], III-V materials (GaAs, AlGaAs, GaAsP, GaP)[22] and more recently graphene[23] and other 2D materials [24–26].

In fig. 2.9A-B the band diagram of the most common LEDs can be observed. It consists on a $p - n$ junction with heavily doped p and n regions to generate holes in the p -region and electrons in the n -region. In (A) when $V_b = 0$, a depletion region is

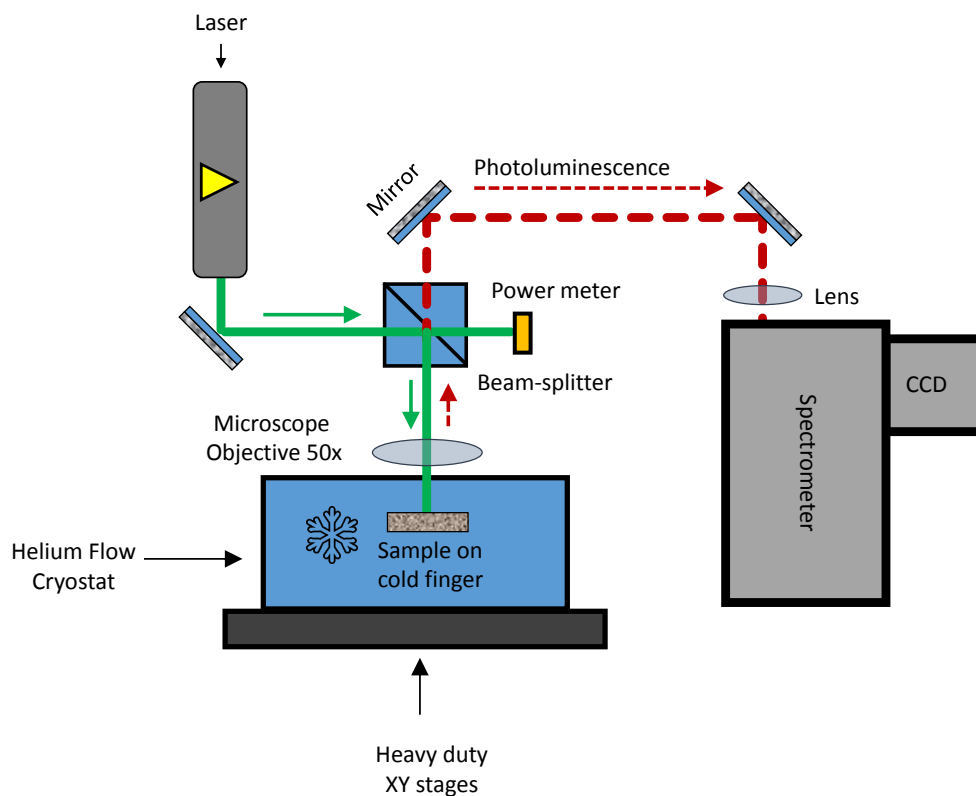


Fig. 2.8 Schematic diagram of the experimental cw μ PL setup. A laser beam (green solid line) is directed and focused to the target sample with a 50x (NA=0.55) objective in a continuous flow He cryostat. PL emitted (red dashed line) is collected and guided to a spectrometer and a CCD camera. The laser power is monitored with a power meter after the beam-splitter. The cryostat is suitable for electrical measurements and is also used for EL.

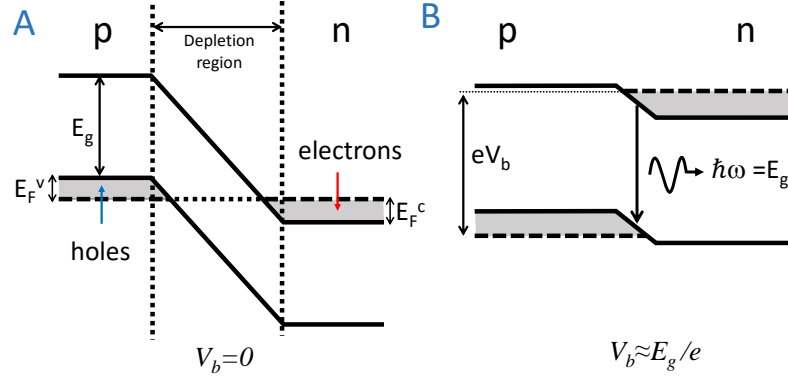


Fig. 2.9 Band diagram of a LED that consists on a $p - n$ junction. In (a) $V_b = 0$, no light is emitted due to absence of electrons and holes in the depletion region. (b) $V_b \approx E_g/e$, light is emitted the spatial overlap of e and h populations.

formed in between the two regions so light cannot be emitted. When $V_b \approx E_g/e$ (fig. 2.9B), current starts to flow through the device. A region is formed at the junction where both electrons and holes are present. Analogous to PL, light is emitted when the electrons recombine with holes in that region [13]. The emission wavelength will depend on E_g and on the binding of the $e - h$ pairs due to excitonic effects.

In this work, electroluminescence from LEDs made of 2D materials was measured using the μ PL setup showed in fig. 2.8. The cryostat is equipped with wires that are able to connect the contacts in the sample to a voltage source (Keithley 2400) so when the laser is blocked, the setup allows to measure EL by injecting carriers with the source. Further experiments of PL or EL bias dependence with either the laser or electrical excitation are possible to perform with this setup.

Quantum efficiency

The conversion of electrical energy into light is the main function of an LED. So it is necessary to calculate the efficiency of a device, in other words, the proportion of the injected carriers converted into photons[27].

The internal quantum efficiency, or luminescence efficiency (described in sec. 2.2.1),

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is the ratio of the radiative electron-hole recombination to the total (radiative and non-radiative) recombination coefficient (eq. 2.6). This parameter is important for a semiconductor material because it determines the efficiency of light generation. However, for a LED we need to take into account the injected current and compare it with the number of photons emitted [28, 29].

The internal photon flux Φ (photons per second), generated within a volume V of the semiconductor, is directly proportional to the carrier-pair injection rate R (pairs/cm³-s). In the steady-state, the injected carrier concentration $\Delta n = R\tau$ (recombination rate = injection rate), where τ is the total recombination lifetime ($1/\tau = 1/\tau_R + 1/\tau_{NR}$). The injection of RV carrier pairs per second therefore leads to the generation of a photon flux:

$$\Phi = \eta_R RV = \eta_R i / e \quad (2.8)$$

where e is the electron charge and i the injection current.

The efficiency with which the internal photons can be extracted from the LED is known as the extraction efficiency η_e . The output photon flux Φ_o is related to the internal photon flux Φ :

$$\Phi_o = \eta_e \Phi = \eta_e (\eta_R i / e) \quad (2.9)$$

where the extraction efficiency η_e indicates how much of the internal photon flux Φ is transmitted outside the device.

We can define a single quantum efficiency based on both η_e and η_R as external quantum efficiency (EQE):

$$\eta_{ext} \equiv \eta_e \eta_R \quad (2.10)$$

So the output photon flux will be:

$$\Phi_o = \eta_{ext}i/e \quad (2.11)$$

We can think of η_{ext} as the ratio of the output photon flux Φ_o to the injected electron flux i/e .

Finally, we can deduce the LED optical power P :

$$P = h\nu\Phi_o = \eta_{ext}h\nu i/e \quad (2.12)$$

With this definition of EQE, an estimate of the efficiency of the devices described in chapters 5 and 6 will be carried out. Additional losses in the system used to measure EL that need to be taken into account in order to perform this calculation are discussed in Appendix B.

Another aspect to consider is the light extraction efficiency of an LED. Photons are created inside of a material rather than in air, where we would ultimately like them to go. Different techniques are employed to maximize the light extraction on the device depending on the chip refraction index. One way to improve the situation is to immerse the chip in a surrounding index that is higher than air. Typically, the material used is silicone, with an index of roughly 1.47. A dome larger than the chip is typically used, since it results in very little refraction (ray deviation) and therefore does not cause further issues. Additionally, a silver film is deposited below the active materials to reflect the light emitted [30].

There are two primary types of LED package architectures that have emerged over the years: dome and flat cast. For the most part, the LED manufacturers supply dome or flat cast LED sources, and optical designers choose from an array of off-the-shelf output optics or design their own optic implementations. Typically, when the most important specification in your application is total lumens, dome out-coupling optics

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provide a better solution. [30].

In fig. 2.10 it shown the domed system geometry on the left and the relative source flux exiting the dome on the first pass to the right. A relatively modest dome radius of 1 mm allows $>99.9\%$ of the chip flux to exit.

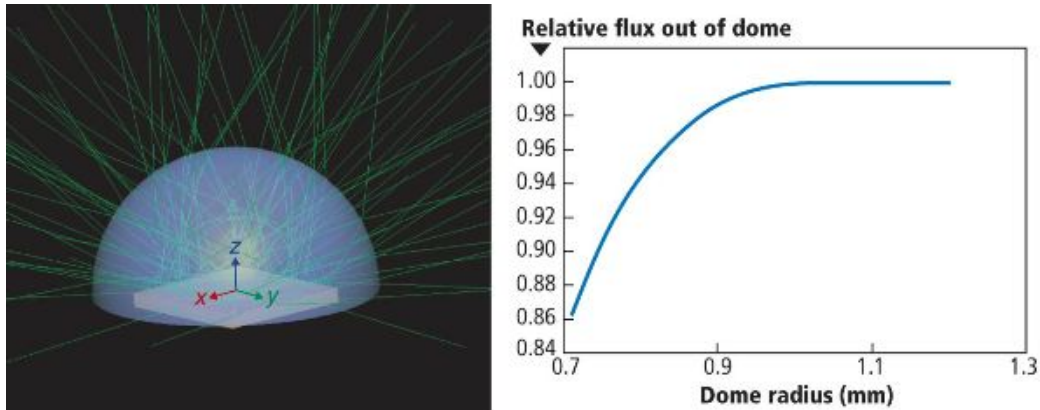


Fig. 2.10 Light flux on a domed LED system (left side). Relative flux out of dome plotted as a function of dome radius (right side) [30]

2.2.2 Time-resolved spectroscopy

Time-resolved spectroscopy is the technique to study dynamic processes that occur in materials after illumination. It is commonly used to study with temporal and spectral resolution the luminescence of excited materials such as semiconductors [31].

As discussed in previous subsections, semiconductor materials can emit light by different ways such as EL and PL. When the excitation needed in the process is not continuous but pulsed, study of the dynamics of the generated light can be performed. This allows the measurement of the overall lifetime τ described in section 2.2.1, that is given by $1/\tau = 1/\tau_R + 1/\tau_{NR}$.

An essential instrument to measure PL with temporal resolution is the Streak Camera. This tool is able to transform a temporal profile of a light pulse into a

spatial profile. It is achieved when the light pulse enters the camera and a deflection of the photons is rapidly executed in such a way that the photons that arrive first are detected in a different place on the detector than those that arrive later in the detector. When the streak camera is combined with a spectrometer, both temporal and spectral resolution can be achieved.

The experimental setup used in this work to measure time-resolved PL is presented in fig. 2.11. It is very similar to the standard μ PL system of fig. 2.8. The excitation source is one of the main differences. A pulsed Ti:Sapphire laser tuned to 808 nm (red solid line) is directed into a second harmonic generator (SHG) in order to obtain a higher photon energy beam (blue solid line) at 404 nm. Then this light is guided with a 50:50 beam splitter to the sample focused with a 20x objective lens and to a power meter. The PL emitted is collected and directed with the same objective lens to the spectrometer. The light diffracted by the spectrometer is studied with a CCD camera to acquire the spectrum and with a streak camera that allows to acquire a time-resolved PL signal.

2.2.3 Raman spectroscopy

When light is scattered by atoms in a crystal lattice, most photons are scattered elastically with the same wavelength as the incident light. This process is known as elastic or Rayleigh scattering. However, a small amount of light is scattered following an inelastic process that results in a difference in wavelength [32]. The process leading to this inelastic scattering is called the Raman effect and occurs with a change in vibrational, rotational or electronic energy of atoms in the crystal [33].

Raman scattering was discovered by Sir C. V. Raman who won the Nobel prize for his work. A diagram of the different processes is shown in fig. 4.4. If the material being studied is illuminated by monochromatic light, like a laser, the spectrum of the

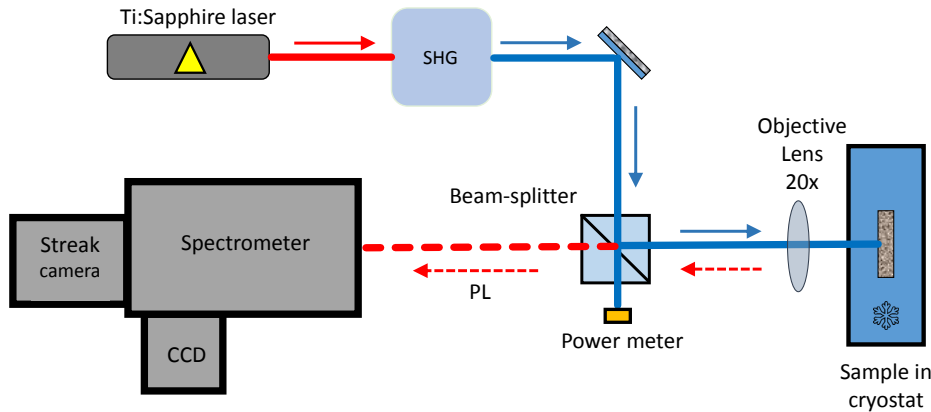


Fig. 2.11 Schematic diagram of the experimental time-resolved μ PL setup. A laser beam (red solid line) enters to a second harmonic generator, then the resulting beam (blue solid line) is guided and focused to the target sample in a continuous He flow cryostat. PL emitted (red dashed line) is collected and directed to a spectrometer equipped with a CCD and a streak camera. Additional devices in the setup allow the measurement of laser power, movement of cryostat and visualisation of sample.

scattered light consists of a strong line (the laser line, green arrow) with the same frequency as the incident illumination together with lines with weaker intensities on either side shifted from the strong line by frequencies from a few to about 3500 cm^{-1} . The lines with frequency lower (higher) than the exciting laser are called Stokes (anti-Stokes) lines, red and purple arrows respectively. Each line corresponds to a different vibrational mode of the crystal.

Raman spectroscopy is, nowadays, an important technique used to identify molecules with the vibrational and rotational modes [34] and it has been an essential technique to investigate two-dimensional materials [8, 9, 35, 36]. In this work it was used to study the stability of two-dimensional films in chapter 4. The equipment employed was an InVia Renishaw System with a 532 nm laser for excitation at room temperature.

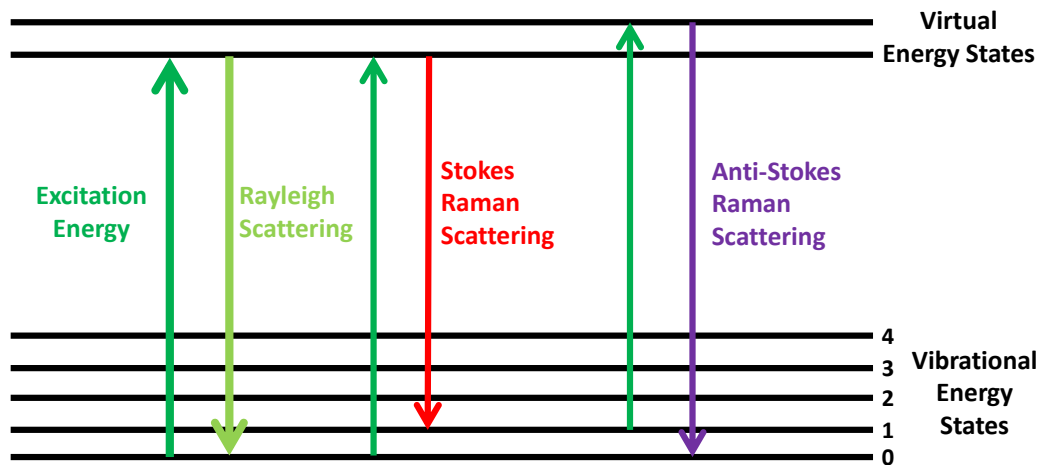


Fig. 2.12 Energy diagram showing the states and processes in Raman signal: Rayleigh scattering in light green arrow, Stokes Raman scattering in red and anti-Stokes scattering in purple. Excitation line is shown in green.

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Chapter 3

Photoluminescence of two-dimensional GaTe and GaSe thin films

3.1 Introduction

In section 1.2.3 a description of the properties of III-VI semiconductor materials was discussed. In this chapter we present a photoluminescence study of thin films of two of these compounds: GaTe and GaSe.

GaTe and GaSe are layered crystals with strong covalent in-plane inter-atomic bonding (with some ionic contributions [1, 2]) and weaker predominantly van der Waals inter-plane bonding [1, 3–6]. Single tetralayer having hexagonal in-plane structure consists of two Ga atoms and two Se or Te atoms: Se-Ga-Ga-Se and Te-Ga-Ga-Te [5], see fig. 1.4. The bulk lattices are built by stacking tetralayers, which can occur in several ways[1, 3, 4]. For the wider studied GaSe, several types of stacking exist leading to different polytypes[1, 3, 4]. For GaTe having a monoclinic crystal lattice [2, 7] the polytypic behavior has not been observed [2, 6]. This may lead to lower

probability of stacking faults in GaTe resulting in a clearer observation (compared to GaSe) of the excitonic features in optical experiments [2, 6].

Here we study optical properties of GaSe and GaTe thin films as a function of the film thickness. Continuous-wave (cw) and time-resolved low-temperature micro-PL for a wide range of films from 200 nm to sub-nm (one tetralayer) thicknesses is measured. PL intensity is used to monitor the quantum yield (QY), which falls dramatically for thin films: integrated cw PL intensity drops by up to $\approx 10^4 - 10^5$ when the film thickness is reduced from 200 to 10 nm. Apart from the PL intensity decrease, no trends in spectral and temporal properties have been observed as a function of the film thickness. A similar observation of reduced PL for thin films was previously reported for InSe and was explained as transition to a band-structure with an indirect bandgap, the conclusion also based on the observed PL blue-shift with the decreasing film thickness [8]. Such indirect bandgap behavior is also theoretically predicted for single monolayers of GaSe and GaTe [5]. However, no size-quantization effects as in InSe are observed in our work for GaSe and GaTe. Furthermore neither shows any trends as a function of the film thickness observed in time-resolved PL. Based on the evidence from both cw and time-resolved spectroscopy, we develop a model that shows that the PL reduction can be explained by non-radiative carrier escape to surface states. Our explanation does not require introduction of the direct-to-indirect band-gap transition. Following the cw PL data analysis, we identify a critical film thickness of about 30-40 nm, below which the non-radiative carrier escape changes its character. In our work we use thin films encapsulated in Si_3N_4 . The importance of surface states predicted by our results emphasizes the need for development of novel surface passivation for III-VI films, possibly involving oxygen-free dielectrics such as boron nitride [9].

3.2 Experimental procedure

3.2.1 Fabrication of GaTe and GaSe samples

Single crystals of GaSe and GaTe were grown by high-pressure vertical zone melting in graphite crucibles under Ar pressure. The detailed description of crystal growth processes and properties of GaSe and GaTe can be found in Refs.[10–13]. The gallium mono-chalcogenides used in this work were synthesized from high-purity materials: Ga and Te - 99.9999 %, Se - 99.9995 %. The crystals used are high-resistivity semiconductors with low free carrier absorption, which has been confirmed in infra-red transmission measurements. GaSe and GaTe have n-type and p-type conductivity, respectively. This is a typical observation: selenides are usually of n-type conductivity, whereas tellurides often can have conductivity of both types, even within one ingot. Such behaviour is attributed to deviations of the crystal composition from stoichiometry, usual for metal chalcogenides. In our case the n-type conductivity clearly indicates some excess of Ga (donor) in GaSe, whereas the p-type conductivity indicates a slight excess of Te (acceptor) in the GaTe.

The III-VI thin films studied in this work were fabricated using the mechanical exfoliation method from bulk described in section 2.1.1. The films were deposited on Si/SiO₂ substrates. Within the first 15 minutes after the exfoliation/deposition procedure, the films were placed in a plasma-enhanced chemical vapor deposition (PECVD) reactor and a 15 nm Si₃N₄ layer was deposited with the sample maintained at a temperature of 300 °C. This process leads to a complete encapsulation of the films, protecting them from interaction with oxygen and water present in the atmosphere. Although Si₃N₄ molecules could be grown in between multilayer films, this technique shows no substantial effects in the PL signal. Films with a wide variety of thicknesses were obtained from single unit cell (single monolayer, ML) shown in Fig.3.1 to 200

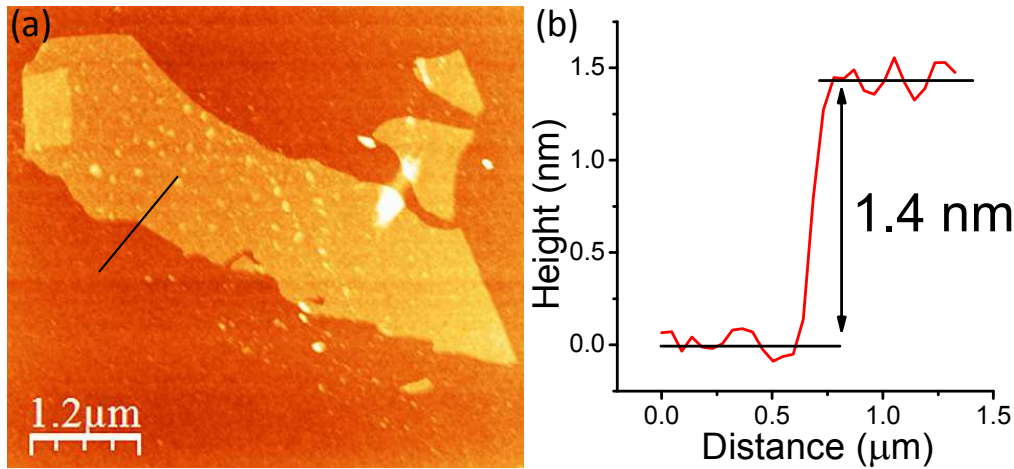


Fig. 3.1 (a) AFM image of a GaSe thin film having a single unit cell thickness. This film is obtained by mechanical exfoliation. (b) Cross-sectional plot along the black line in (a).

nm. The thicknesses of the films were determined using atomic force microscopy.

3.2.2 Optical characterization methods

Optical characterization of the GaTe and GaSe thin films was carried out using the low-temperature micro-photoluminescence (μ PL) technique described in section 2.2.1. The sample was placed on a cold finger in a continuous flow He cryostat at a temperature of 10 K. A microscope objective was placed outside the cryostat and was used to focus the laser beam on the sample (in a $\approx 2\mu\text{m}$ spot) and to collect the photoluminescence (PL) from the films. In continuous-wave (cw) experiments, PL was detected with a 0.5 m spectrometer and a liquid nitrogen cooled charge coupled device. For cw PL excitation a laser emitting at 532 nm (2.33 eV) was used. Ultra-fast spectroscopy experiments was obtained with the setup described in section 2.2.2. The excitation of GaTe and GaSe layers was performed using frequency-doubled titanium-sapphire (wavelength of 415 nm) focused on the sample in $\approx 10\mu\text{m}$ spot. Time-resolved PL (TRPL) was detected using a streak camera. The temporal resolution of the experimental setup

for the time-resolved measurements was 10 ps.

3.3 Experimental results

3.3.1 Low-temperature cw PL results: GaTe

Fig.3.2(a) shows typical PL spectra measured for Si₃N₄-capped GaTe thin films at $T = 10\text{K}$ (cw laser power $P=2\text{ mW}$ in Fig.3.2(a)). In this figure, in all films but the one with the thickness $h_{film}=8\text{ nm}$, a narrow feature is observed around 1.75 eV. It is observed at an energy where free exciton (FE) PL is expected. We will therefore refer to such features in GaTe (and GaSe) films as a 'free exciton' peak as opposed to the low energy broad PL bands corresponding to excitons bound to impurities/defects and observed in the range of 1.6-1.7 eV for GaTe. In reality the free excitons may also experience disordered potential and degree of localization as evidenced from a relatively broad line of 10-15 meV (varying from sample-to-sample). A different behaviour of the FE peak compared to the bound excitons has been verified in temperature and power-dependent measurements and is further confirmed in time-resolved studies discussed below. Although all spectra in Fig.3.2(a) show these typical features, no clear trend is observed in the relative intensities of the impurity/defect band and the FE peak as a function of h_{film} . The FE peak is usually pronounced in thick films of around 100 nm and above. In thin films as in the 8 nm film in the figure, the FE peak could only be observed under high power pulsed excitation, when the impurity/defect states saturate.

No size-quantization effects have been observed in GaTe thin films in contrast to InSe in Ref.[8] where the PL blue-shift was observed for thin layers. The sharpest PL feature in GaTe spectra, the FE line, has the peak energy varying from film to film in the range 1.74-1.76 eV. There are also no clear trends for the PL peak energies of the

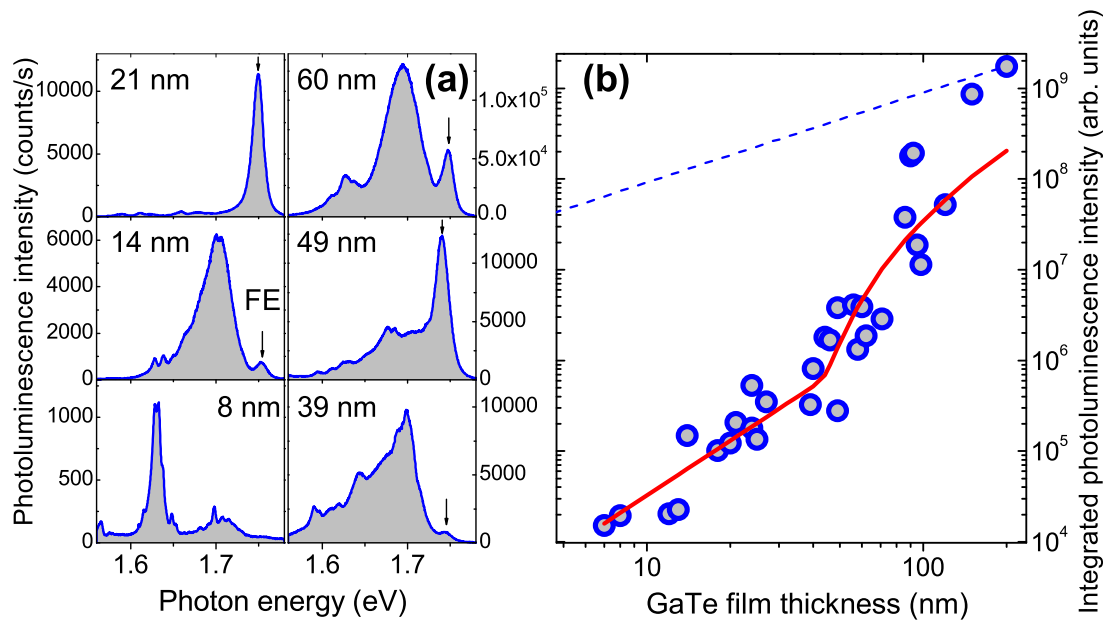


Fig. 3.2 Low-temperature cw PL results for GaTe thin films. (a) PL spectra for films of various thicknesses. h_{film} is marked on the plots. On the plot for the film with $h_{film}=49$ nm FE marks the free exciton peak observed at similar energies for all films on the figure except the one with $h_{film}=8$ nm. (b) Symbols show experimentally measured integrated PL (for $T=10$ K) for the cw laser excitation power of 2 mW at 532 nm (2.33 eV). Dashed line shows expected variation of PL following the change in the absorption of the thin film assuming constant quantum efficiency. Solid line shows the results of calculations using the model discussed in text.

localized exciton band as a function of the film thickness.

A pronounced feature of PL measured from different films is a dramatic decrease of PL intensity with the decreasing thickness of the material [see Fig.3.2(b)]: about 10^5 (10^4) decrease is observed between 200 (100) and 7 nm. The strongest PL reduction by 3 orders of magnitude is detected between 200 and 40 nm. For $h_{film} < 40$ nm, the PL intensity reduction slows down and decreases less than 100 times when h_{film} is varied between 40 and 7 nm. The dotted curve in the graph shows the expected PL intensity behaviour assuming thickness-independent quantum efficiency, i.e. when reduction in PL is caused solely by the reduced absorption and reduced number of e-h pairs created by the laser. The curve is described by the expression $I_{PL} = I_{GaTe}[1 - \exp(-\alpha_{GaTe}h_{film})]$, where I_{GaTe} is the PL intensity for films with $h_{film} \approx 200$ nm. The absorption coefficient $\alpha_{GaTe} = 5000 \text{ cm}^{-1}$ is used according to Ref.[14]. A discrepancy by a few orders of magnitude between the experiment and the calculated curve in a wide range of film thicknesses is evident on the graph.

3.3.2 Low-temperature cw PL results: GaSe

Fig.3.3(a) shows typical PL spectra measured for Si_3N_4 -capped films of GaSe of several thicknesses between 8 and 70 nm (the cw laser power of 2 mW is used). The PL signal is observed in a range from 1.95 to 2.05 eV, which is below the emission energy of the free exciton, reported to be around 2.10 eV for some high purity GaSe samples (see e.g. Ref.[15]). The detected PL in our samples thus comes from impurity/defect states. The observed localized states may originate from the non-stoichiometric composition of the bulk material.

In Fig.3.3(a) it is observed that PL spectra of thin films < 20 nm usually consist of multiple pronounced lines (a feature similar to GaTe in Fig.3.2(a)), whereas PL spectra tend to exhibit a single pronounced peak for thicker films. PL linewidths

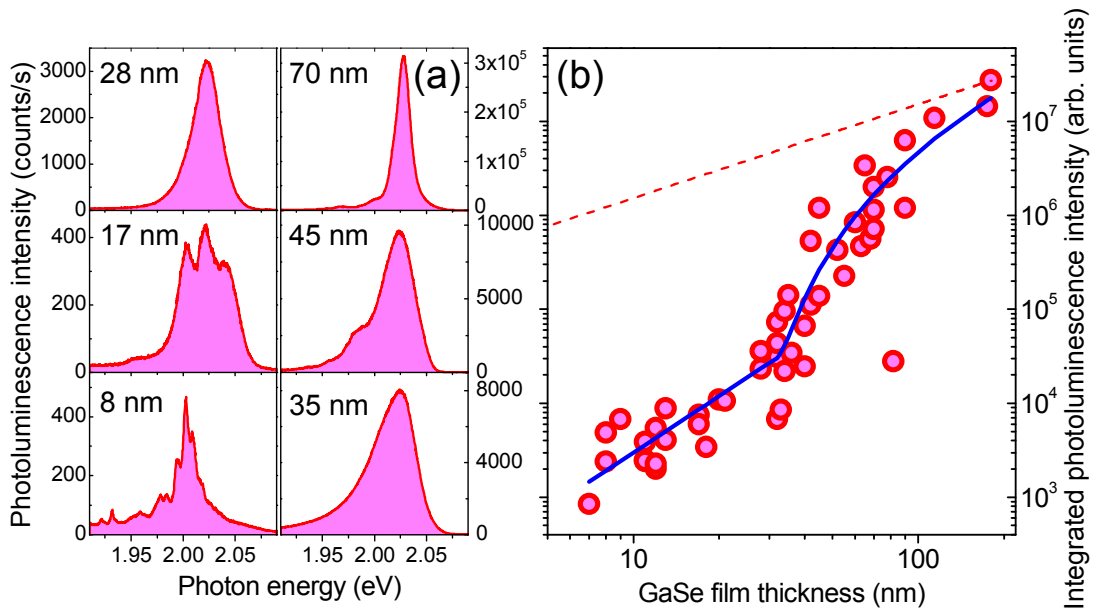


Fig. 3.3 Low-temperature cw PL results for GaSe thin films. (a) PL spectra for films of various thicknesses. (b) Symbols show experimentally measured integrated PL (for $T=10\text{K}$) for the cw laser excitation power of 2 mW at 532 nm (2.33 eV). Dashed line shows expected variation of PL following the change in the absorption of the thin film assuming constant quantum efficiency. Solid line shows the results of calculations using the model discussed in text.

vary between 15 and 60 meV. From the data measured on more than 50 films, we observe that PL peak energies have a very wide distribution in the range 1.99-2.06 eV. Similarly to GaTe, there is no evidence for size-quantization effects as a function of the film thickness.

In some GaSe films, when excited with low power ($<1 \mu\text{W}$), sharp PL lines with linewidths from 1 to 5 meV are observed (see fig. 3.4a-f). These lines are similar to the sharp features in the PL spectrum for the 8 nm film in Fig.3.3(a) and occur in the whole energy range of GaSe PL. This 'quantum dot-like' emissions can be originated from bound excitons due to impurities in the crystal [16]. A detailed study of such defect states is outside the scope of this thesis.

Further evidence for exciton localization in thin films is a pronounced Stokes blue-shift observed when the laser excitation density is increased as shown in Fig.3.5. The inset in Fig.3.5(a) shows that the PL peak shifts by ≈ 20 meV as the cw laser power is changed from 0.01 to 2 mW. This is a typical behaviour observed in all GaSe films independent on the film thickness: at high power, saturation of some of the PL features is observed accompanied in most cases with a blue-shift of PL of around 10-20 meV. This is a typical behaviour observed for localized exciton states in semiconductors, the effect also similar to the state-filling phenomenon in semiconductor quantum dots [17]. In some GaSe films, if the optical pumping is further increased, for example, by using pulsed excitation, a relatively broad free exciton feature can be observed, as shown in Fig.3.5(b). This behaviour is in agreement with that observed previously in GaTe and GaSe under pulsed excitation, and is related to saturation of the localized states having relatively slow recombination rates [18].

Similarly to GaTe films, a significant decrease of PL intensity with the decreasing thickness of the GaSe films is observed [see Fig.3.3(b)] by about 2×10^4 between 200 and 7 nm. As for the GaTe films in Fig.3.2(b), the strongest PL reduction by 3 orders

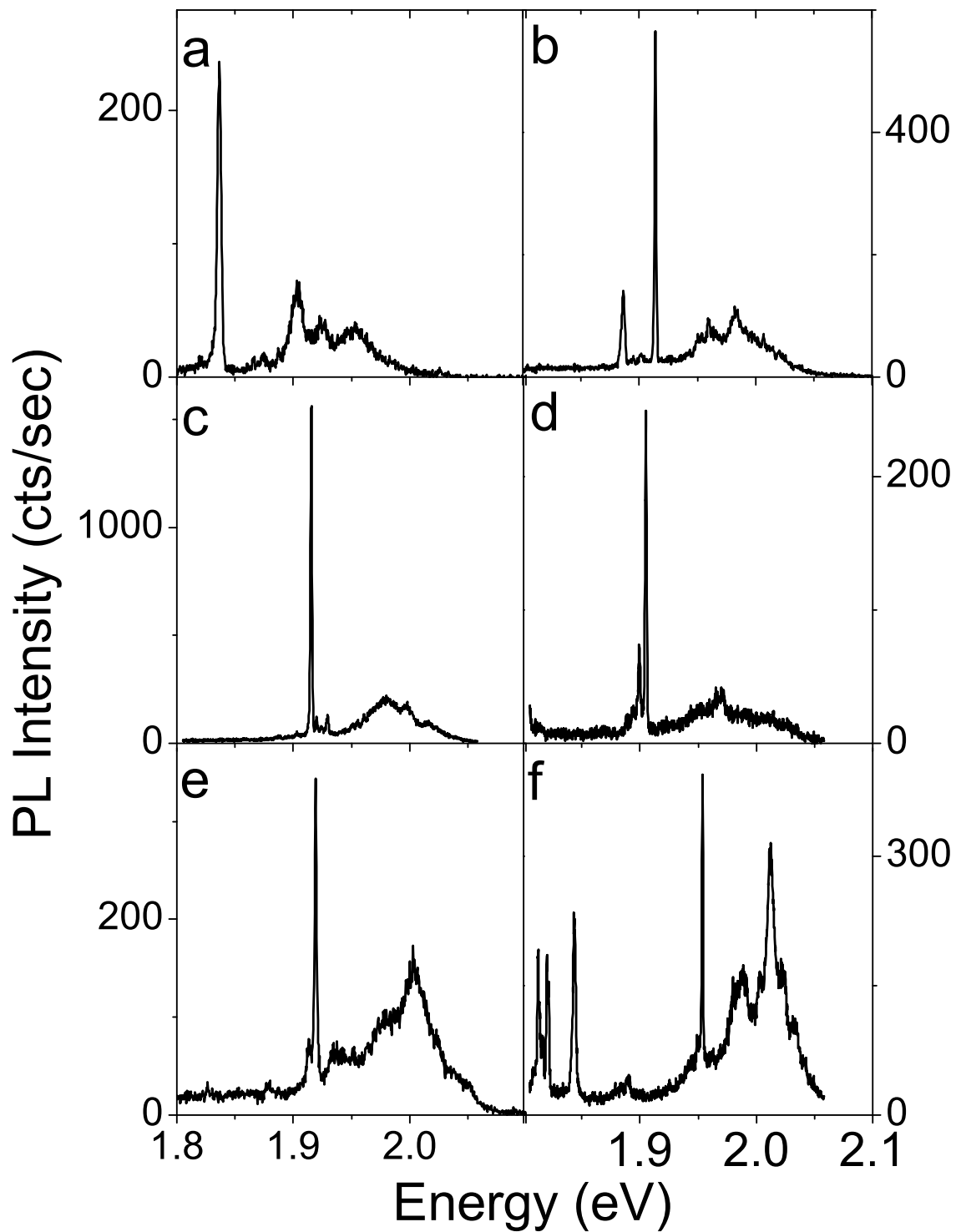


Fig. 3.4 (a-f) GaSe narrow emission lines at low excitation power (500 nW) on thick films (~ 80 -100 nm). The linewidth of these peaks fluctuate from 1 to 5 meV.

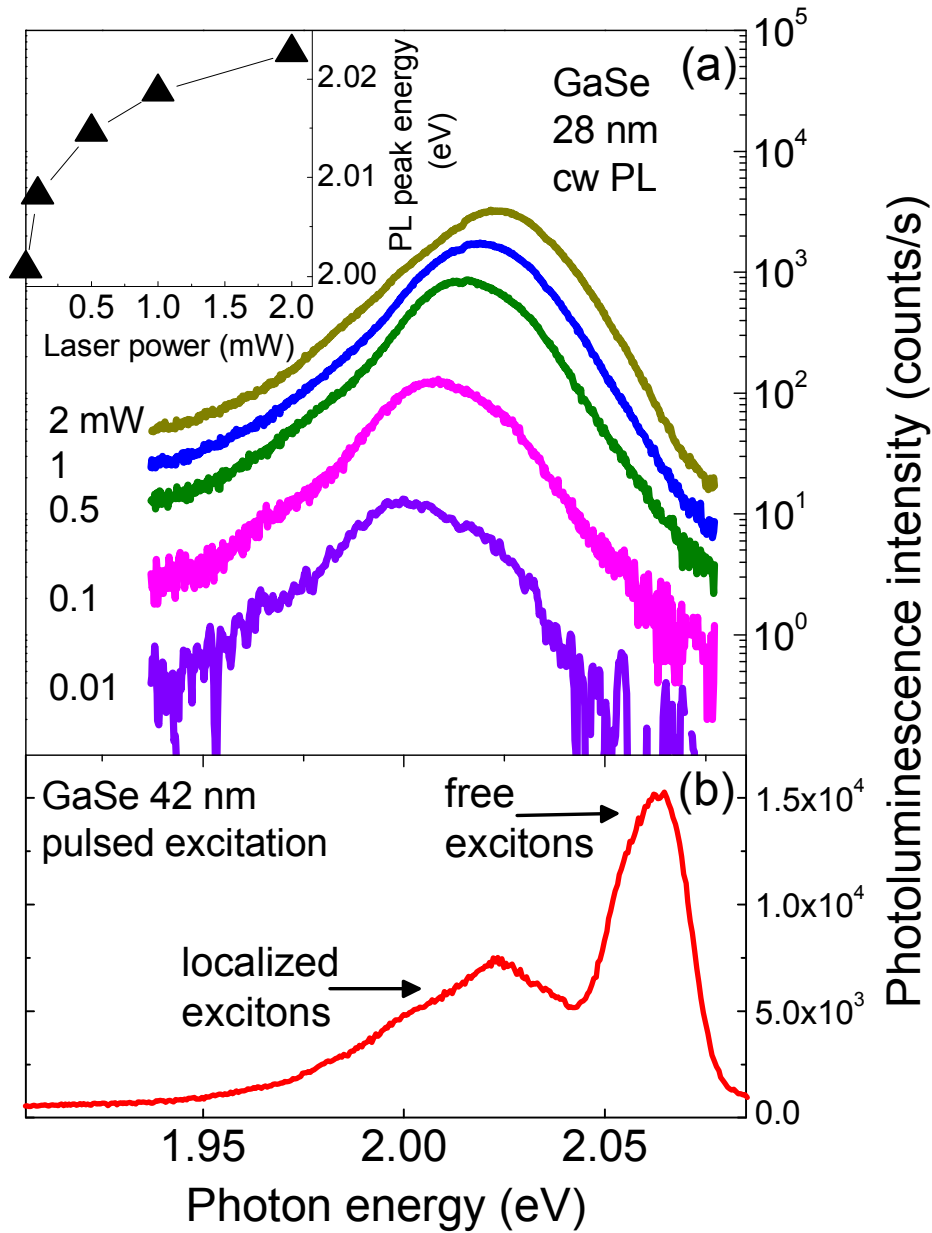


Fig. 3.5 (a) Observation of the Stokes shift of localized exciton states in cw PL of a 28 nm thick GaSe films. Excitation with a cw laser at 532 nm (2.33 eV) is used at $T=10\text{K}$. (a) Observation of the free exciton feature in PL of a 42 nm thick GaSe films. The time-integrated PL is shown measured for pulsed laser excitation at 420 nm (2.95 eV) with a power of 1 mW. A free exciton peak having a linewidth of ≈ 25 meV is observed at 2.06 eV.

of magnitude is detected between 200 and 30 nm. For $h_{film} < 30$ nm, the PL intensity reduction slows down and is about 30 when h_{film} is varied between 30 and 7 nm. Similar to GaTe in films with $h_{film} < 7$ nm PL is completely suppressed. Similarly to Fig.3.2, we show a curve that describes PL reduction due to the reduced absorption only calculated as $I_{PL} = I_{GaSe}[1 - \exp(-\alpha_{GaSe}h_{film})]$, where I_{GaSe} is the PL intensity for films with $h_{film}=150$ nm and the absorption coefficient $\alpha=1000$ cm⁻¹ [14]. As for GaTe, a significant discrepancy by a few orders of magnitude between the experimental results and the calculated curve is clear in a wide range of film thicknesses.

3.3.3 Time-resolved PL measurements

In order to shed further light on the results of cw PL and also provide further experimental foundation for our theoretical model, time-resolved PL experiments have been carried out. We find that TRPL measurements show no trend in life-times as a function of the film thickness in both GaTe and GaSe as discussed below. Fig.3.6 shows typical TRPL data obtained at $T \approx 10$ K.

For GaTe, the difference in the origin of the PL features observed in Fig.3.2(a) is further evidenced in TRPL. Fig.3.6(a,b) shows data for a 27 nm thick film exhibiting a behavior typical for films with h_{film} in the range 20 to 200 nm. Fig.3.6(a) presents a streak-camera scan clearly showing two pronounced features at 1.76 eV and 1.71 eV corresponding to the free and localized excitons, respectively. The free exciton peak intensity decays considerably faster than that of the localized states, which does not change significantly on the time-scale of 130 ps shown in the figure. Fig.3.6(b) shows two decay curves measured at 1.76 eV (gray) and 1.71 eV (red). Fitting with single exponential decay functions is shown with blue curves and gives 10 ps for the free exciton and 150 ps for the localized states. The inset shows a PL decay curve measured at 1.73 eV on a larger time-scale, exhibiting an almost complete decay of the

signal at 1 ns. We find similar life-times for other films, however no clear dependence on the film thickness is observed: the free exciton PL decay time varies between 5 and 25 ps and that for the localized states between 100 and 200 ps. The lifetimes also weakly depend on the laser excitation power. We also note the rise times of ≈ 15 and 20 ps for the free and localized states, respectively, indicating fast carrier relaxation into light-emitting states.

Similar difference between the PL dynamics of the high and low energy part of the spectrum is observed for GaSe thin films in Fig.3.6(c),(d). Here a behaviour resembling the Stokes shift shown in Fig.3.5 is observed: as the carrier density decreases with time after the laser pulse, the PL intensity maximum progressively moves to lower energy. Fig.3.6(d) details the behaviour shown in Fig.3.6(c): two decay curves measured at 2.055 eV (gray) and 2.035 eV (red) are shown. In the center of the PL band at 2.035 eV, the non-exponential decay occurs with a characteristic time of 400 ps, which also shows a slow-decaying component. At around 2.055 eV, the initial PL time-dependence can be well fitted with a single-exponential decay with a lifetime of ≈ 40 ps, dominated most likely by carrier relaxation to lower energy. The complex behaviour in GaSe films occurs due to the partial saturation of the states at short times after the excitation pulse and fast relaxation to lower energy. We find similar behaviour for films with other thicknesses. Similarly to GaTe no clear dependence on the film thickness is observed and the rise times of ≈ 30 ps are found for the localized states.

3.3.4 Modeling

In order to describe the observed trend of photoluminescence (PL) intensity as a function of film thickness h_{film} in GaSe and GaTe thin films we have developed a simplified rate equation model described in detail in the Appendix A. As shown in Fig.3.7 we assume that the film is divided in three regions: (1) two regions of thickness

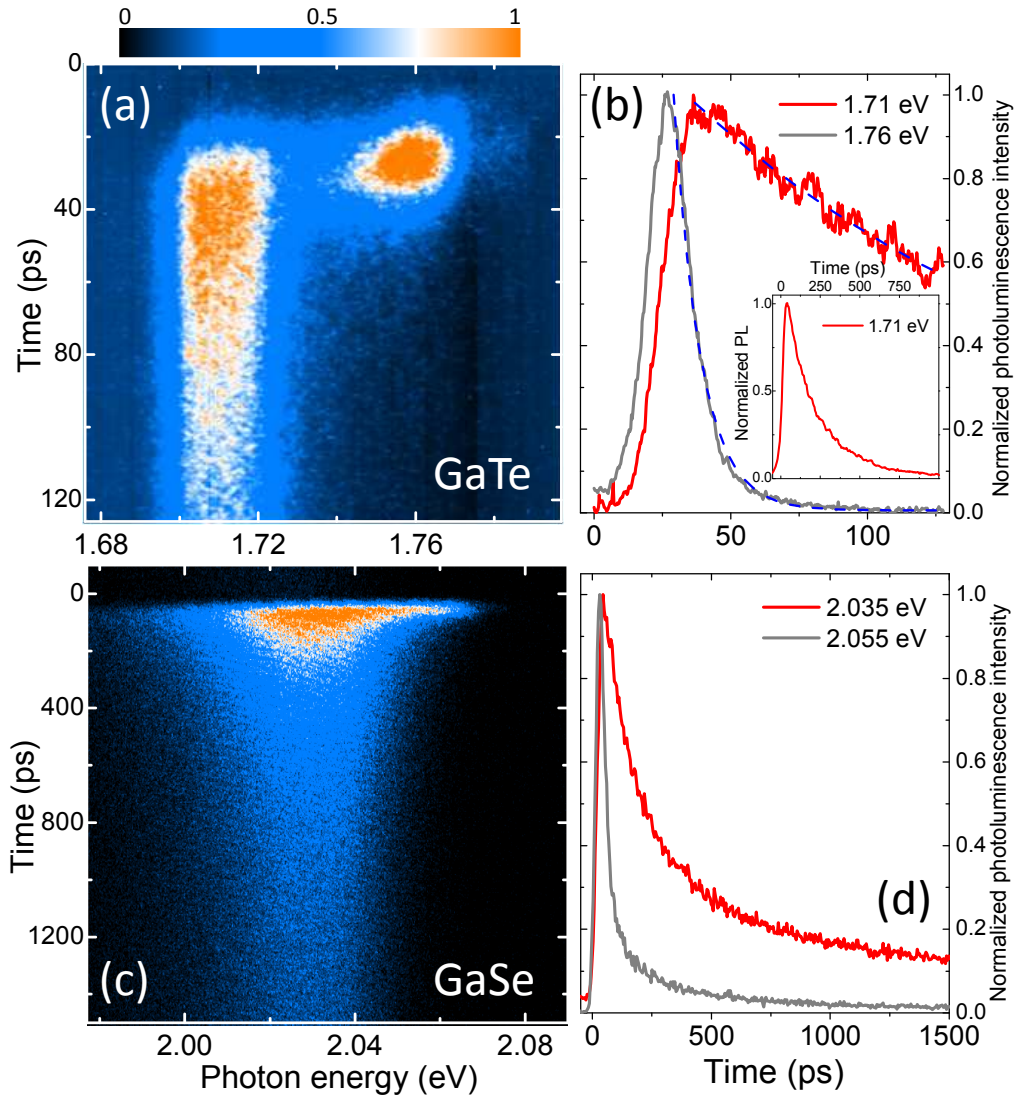


Fig. 3.6 Time-resolved PL data for GaTe (a,b) and GaSe (c,d) thin films measured at $T=10\text{K}$. Note different time-scales for the GaTe and GaSe data. (a) and (c) show streak-camera scans. (b) and (d) show PL traces measured at 1.76 (gray) and 1.71 eV (red) for GaTe and 2.055 (gray) and 2.035 eV (red) for GaSe. Blue lines in (b) show fitting with a single exponential functions as described in text.

h_0 near the film surfaces where fast carrier relaxation to surface traps occurs leading to non-radiative carrier escape; (2) a 'normal' region of thickness $h_{film} - 2h_0$ in the film's central part not containing the traps where the photo-excited carriers can escape into regions (1), where they undergo non-radiative decay. Here the h_0 value may be associated with a depletion depth or an average surface trap radius[19]. As shown in Fig.3.7 there are two possibilities: in Fig.3.7(a), where $h_{film} > 2h_0$ and both regions of type (1) and (2) exist; and in Fig.3.7(b), for films with $h_{film} \leq 2h_0$, where region (2) is not present. The model assumes that light absorption and PL occurs in both types of regions. In both regions, e-h pairs relax with the time τ_{rel} into the 'PL states' giving rise to radiative recombination. Non-radiative escape from the 'PL states' is neglected.

In region (1), in first approximation the average time it takes for the carrier/e-h pair to escape non-radiatively is proportional to half the thickness of region (1) (can be understood as the average time for the carrier to reach the surface or as the overlap of the wavefunction of the carrier and the trap). In region (2), the non-radiative escape time reflects the average time it takes for a carrier or an e-h pair to reach any of the regions (1). The underlying mechanism for this process may be depletion and band-bending expected at the film surface leading to charge separation and non-radiative decay [19]. We assume that once the carrier or e-h pair has reached region (1) it escapes non-radiatively. In first approximation, the average time it takes a carrier/e-h pair to reach region (1) is proportional to half the thickness of region (2). Thus we introduce non-radiative decay times in region (1) as $\tau_{nr1} = (h_0/2)/u_1$ for $h_{film} > 2h_0$, and $\tau_{nr1} = (h_{film}/4)/u_1$ for $h_{film} \leq 2h_0$. In region (2) it is $\tau_{nr2} = (h_{film}/2 - h_0)/u_2$. Here u_1 and u_2 have dimensions of m/s. In the case of region 2 where effectively we assume ballistic exciton (or electron/hole) transport preceding the non-radiative escape, u_2 can be interpreted as the average carrier velocity.

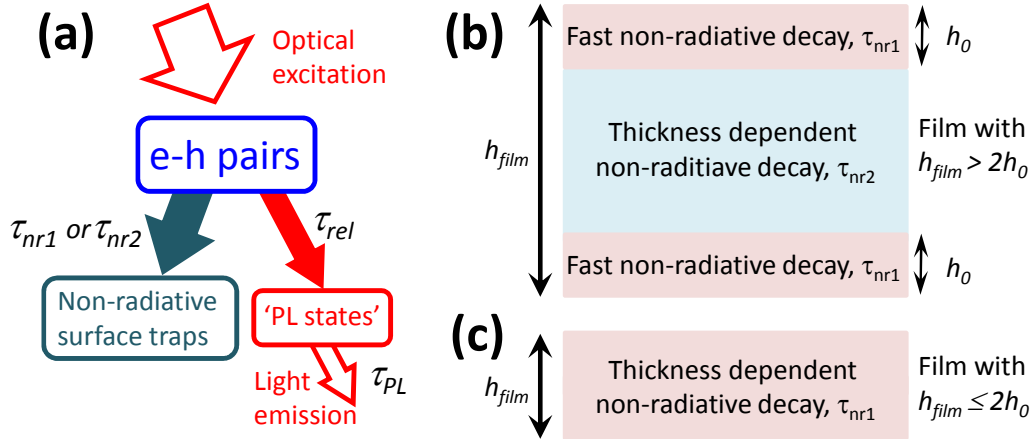


Fig. 3.7 Diagrams illustrating the theoretical model for the dependence of the cw PL intensity on the film thickness. (a) Processes included in the model (see text and supplementary information for more details): optical excitation of e-h pairs; relaxation into the non-radiative traps with a time τ_{nr1} or τ_{nr2} depending on the thickness of the film; relaxation with a time τ_{rel} into the light-emitting states denoted as 'PL states'; PL emission from these states with a time τ_{PL} . (b) and (c) shows two types of thin films with thicknesses above and below the critical thickness of $2h_0$. In (a) we show a relatively thick film with a thickness of $h_{film} > 2h_0$. This film has three regions: two regions of thickness h_0 near the film surfaces where fast carrier relaxation to surface traps occurs (shown in pink), and a 'normal' region of thickness $h_{film} - 2h_0$ in the middle of the film (blue) where non-radiative processes are weaker and occur through carrier escape into the surface regions. In (b) we show a thin film with $h_{film} \leq 2h_0$, where a region of one type only exists, where fast non-radiative carrier decay occurs.

3.3.5 Discussion

As detailed in Appendix A and observed in Fig.3.2(b) and Fig.3.3(b), we find that the proposed model provides a reasonable description of our data using four parameters (which are not completely independent as we find): the 'critical' thickness h_0 , a parameter describing the amount of light absorbed by the film, and products $\tau_{rel}u_1$ and $\tau_{rel}u_2$. In particular the fitting functions that we produce capture the change in the 'slope' of the data observed at around 30 nm for GaSe and 40 nm for GaTe, which we interpret as the thickness of the film where $h_{film} \approx 2h_0$, i.e. the thickness of region (2) turns to zero, and non-radiative carrier escape changes its character.

However, we find that the accuracy of the fitting is not sufficient to extend our analysis beyond determination of the order of magnitude of the products $\tau_{rel}u_1$ and $\tau_{rel}u_2$. We find that $\tau_{rel}u_1$ and $\tau_{rel}u_2$ are of the order of 1000 nm and 100 nm, respectively, for both GaSe and GaTe. The solid line shown in Fig.3.2(b) for GaTe films is obtained for $\tau_{rel}u_1 = 3800$ nm and $\tau_{rel}u_2 = 200$ nm, whereas the fitting in Fig.3.3(b) for GaSe films is done for $\tau_{rel}u_1 = 2900$ nm and $\tau_{rel}u_2 = 150$ nm. We note that the description of GaSe PL is more satisfactory, possibly because in GaTe there is a contribution from free exciton PL, so additional non-radiative escape channels and relaxation processes need to be taken into account. For thicker GaTe films, much stronger PL than predicted by the model is observed, which probably signifies suppression of additional non-radiative escape that free excitons experience in relatively thin films. It is also notable that in GaTe the PL lifetime for localized states is shorter than in GaSe, which may be due to non-radiative escape. Such processes are not included in the model. Another reason could be deviation from the effectively 'ballistic' transport that we assume leads to the carrier escape into regions (1), and its replacement for the larger thicknesses with a slower 'diffusion' process leading to slower non-radiative escape.

Assuming that the measured PL rise-times of ≈ 20 ps are close to τ_{rel} , we can estimate characteristic non-radiative times τ_{nr1} and τ_{nr2} for several limiting cases. For example for GaSe we obtain the following values using $\tau_{rel}u_1=1000$ nm and $\tau_{rel}u_2=100$ nm: for $h_{film}=10$ nm $\tau_{nr1}=0.05$ ps, for $h_{film}=30$ nm $\tau_{nr1}=0.15$ ps, for $h_{film}=100$ nm $\tau_{nr2}=7$ ps. For both h_{film} of 10 and 30 nm, the non-radiative decay occurs on a sub-picosecond time-scale. Here, the non-radiative escape is by a factor of the order of 100 faster than relaxation into the light emitting states, which is consistent with a low quantum yield of 10^{-3} reported previously for thin films of MoS₂[20]. For the middle region of a 100 nm film, the characteristic non-radiative escape time is comparable with the relaxation time into the states giving rise to PL, thus a much higher quantum

yield can be expected for films of this thickness.

We also note that in the case of region (2), u_2 should be of the order of a typical thermal velocity of an exciton, v_{th} . By using $T=10$ K and the exciton mass of 0.1 (or 0.2)[21] of the free electron mass we obtain $\tau_{rel}v_{th} \approx 1200$ (or 900) nm for $\tau_{rel} \approx 20$ ps, very similar to the order of magnitude predicted by the model for the product $\tau_{rel}u_2$.

3.3.6 Conclusions

We have investigated thin films of GaTe and GaSe prepared by mechanical exfoliation from bulk crystals, deposited on SiO₂ substrates and capped with a thin layer of Si₃N₄. The study of optical properties of the thin films has been conducted by means of low-temperature cw and time-resolved micro-PL techniques. The most pronounced property observed is a significant reduction of cw PL intensity by up to 10⁵ for thin films of thicknesses about 10 nm compared with the films of 150-200 nm. No measurable PL was observed in films thinner than 7 nm. Except for PL decrease, no other clear trends, including size-quantization effects and PL life-time modifications as a function of the film thickness were observed in cw and time-resolved PL. We argue that the reduction of quantum yield occurs due to the non-radiative processes associated with the surface states. The theoretical model that we develop differentiates between the fast non-radiative carrier escape to surface traps in the thin 15-20 nm layers near the film surface, and slower decay in the middle region of the film, where the carriers first diffuse to the surface layers and then decay. The model accounts for the change in the character of the PL decay for thin films with thicknesses less than 30-40 nm, where we expect only the fast direct relaxation to surface traps.

We do not refer in our model and interpretation to the direct-to-indirect bandgap transition found for thin InSe films in Ref.[8]. The observed strong PL decrease is not accompanied with any significant variation of the PL lifetime, which would most likely

accompany such a change in the band structure. This implies that the PL reduction in our case is not associated with this type of transition, and instead the reduction in the radiative recombination contribution is likely due to an increase in the non-radiative rate in thin films.

Strong non-radiative decay processes occur in the studied films despite complete encapsulation in Si_3N_4 and SiO_2 providing partial protection of the surface from chemical interactions with the ambient atmosphere. This emphasizes the need for development of novel surface passivation for III-VI films, possibly involving oxygen free substrates and encapsulation in additional layered materials such as boron nitride. Using this fabrication methods, the large family of III-VI materials may show weaker dependence of quantum yield on film thickness, which will allow them to play important role as building blocks in van der Waals heterostructures. It is likely that in the near future fabrication of such heterostructures will be carried out in oxygen and water free atmospheres and will include carefully designed passivating layers.

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Chapter 4

Photoluminescence and Raman investigation of stability of InSe and GaSe thin films

4.1 Introduction

An important issue for a variety of layered materials is stability of thin films in ambient conditions[3, 15–18]. This has particularly showed significance in recent studies of semiconducting black phosphorus [15], MoTe₂[19] and layered superconductors such as NbSe₂[20]. We show in this chapter that film stability issues have to be taken into account for III-VI films, in particular GaSe.

Both InSe and GaSe are layered crystals with strong covalent in-plane inter-atomic bonding and weaker van der Waals inter-plane bonding [21, 22] as discussed in 1.2.3. The single tetralayer having hexagonal in-plane structure consists of two Se atoms and two In or Ga atoms: Se-Ga-Ga-Se and Se-In-In-Se as shown in Fig.4.1(a) where the crystal structure of both materials is presented. InSe is a direct gap semiconducting material in its bulk form and has important applications in photo-voltaic devices

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[23]. More recently it has been studied in its layered form [10, 24, 25], where tuning of its band-gap by varying the film thickness has been demonstrated making this material promising for van der Waals heterostructures such as light emitting diodes [6]. GaSe is well known for its nonlinear properties[26] and in the past few years thin films of this material were used for fabrication of photo-detectors [27] and in optical microcavities, where control of light-matter interaction was demonstrated for two-dimensional films[28].

In this chapter we investigate the stability of mechanically exfoliated thin films of InSe and GaSe with thicknesses ranging from 9 to 75 nm using micro-photoluminescence (μ PL) and micro-Raman spectroscopy. We observe decrease of both the PL and Raman signal over a few days. Between the optical measurements the films are kept in air at room temperature. We interpret the decrease of both signals as gradual reduction of the effective film thickness due to interaction with oxygen and water in the atmosphere. This results in significant changes of the crystal properties, leading to modification and decrease of the Raman signal, and also weaker PL following occurrence of non-radiative centers at the layers adjacent to the films surfaces [11]. For GaSe, we estimate the rate of erosion of 0.14 ± 0.05 nm/hour based on our previous detailed measurements of thickness-dependent PL, where we observed a strong reduction of PL intensity with the decreasing film thickness[11] (see chapter 3). Here we also demonstrate that encapsulation of the films in SiO_2 or Si_xN_y allows to overcome the problem of film degradation and prolongs the life-time of the films by two orders of magnitude up to several months.

4.2 Samples and experimental methods

Bulk GaSe studied in this chapter was grown by high-pressure vertical zone melting in graphite crucibles under Ar pressure from high-purity materials: Ga - 99.9999 % and

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Se - 99.9995 % [29, 30]. InSe single crystals were grown by the Bridgman-Stockbarger method from a preliminarily synthesized ingot [31]. The conductivity of both materials is n-type, a typical observation in selenides.

InSe and GaSe films were exfoliated from bulk using a mechanical cleavage method discussed in detailed in section 2.1.1 and were deposited on thermally oxidized silicon substrates. This process generates a variety of films with different thicknesses. Encapsulated GaSe samples were fabricated with the same method and completed with additional deposition of a 10 nm capping layer of Si_xN_y or SiO_2 by plasma-enhanced chemical vapor deposition (PECVD). Atomic force microscopy (AFM) was used to measure the thickness of all studied films, including the capped structures. In Fig. 4.1(b) an AFM image of a multi-layer GaSe film is presented showing that the material thickness varies from ~ 20 nm to ~ 80 nm across the film (see the scale on the right). In 4.1(c) a group of InSe films with various thickness is presented in an optical microscope image. Here, film areas with different thicknesses are observed as different colors[32]. As evidenced from both images, by using a laser spot with a diameter of $\approx 1.5 - 2\mu\text{m}$, film areas with constant thickness can be reliably addressed in micro-PL and micro-Raman experiments.

Optical characterization of GaSe and InSe thin films was carried out using two different techniques: micro-PL (μPL) and micro-Raman spectroscopy. The μPL setup, described in section 2.2.1, was equipped with a 532 nm continuous wave (cw) semiconductor diode laser for excitation of the sample placed in a flow cryostat at $T=10\text{K}$. The laser spot on the sample was $\approx 2\mu\text{m}$. PL signal was collected with an $\text{NA}=0.55$ objective and detected with a 0.5 m single spectrometer and a liquid nitrogen cooled charge coupled device. Micro-Raman was measured at room temperature using an InVia Renishaw System equipped with a 532 nm laser for excitation.

We study PL signal in GaSe and InSe films following their exposure to air starting

Photoluminescence and Raman investigation of stability of InSe and GaSe thin films

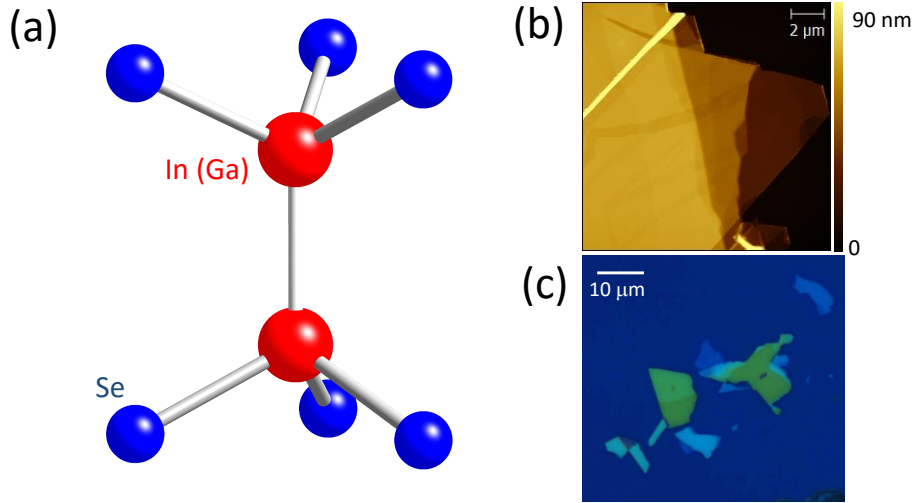


Fig. 4.1 (a) Crystal structure of InSe and GaSe. (b) AFM image of a multi-layer GaSe film. (c) Optical image of various isolated InSe thin films on SiO₂/Si substrates. Different colors in the image are associated with different film thicknesses.

from 1 hour after sample fabrication up to 100 hours. The time the sample is kept in the cryostat during the cool-down, the experiment itself and warm-up is not counted, as during this period the sample is kept in vacuum. After the measurement is complete and the sample returns to room temperature in the cryostat, it is taken out and stored in ambient conditions. The measurements in the cryostat were repeated after keeping the sample in air for 24 hours with a maximum total exposure of 100 hours. Raman experiments were carried out on the day of fabrication within a few hours after exfoliation and one week later.

4.3 Results

Fig.4.2 shows low-T μ PL spectra of (a) a 31 nm InSe and (b) a 52 nm GaSe films measured with 1 mW laser power. The spectra are recorded from the same area of both films after different exposure to air. The InSe film shows a typical excitonic peak at ~ 1.3 eV in agreement with previous reports [24, 25, 33]. PL intensity exhibits a

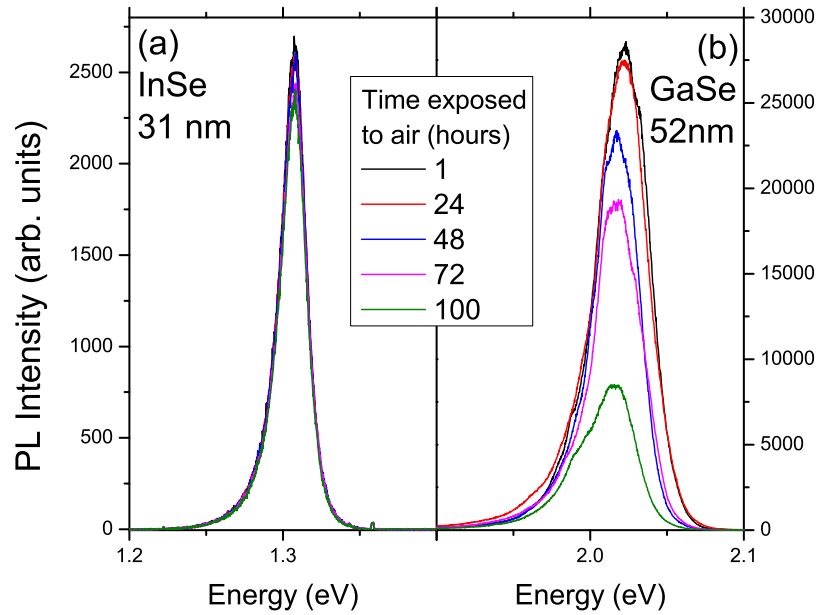


Fig. 4.2 Micro-photoluminescence spectra measured at low temperature of 10 K for (a) a 31 nm InSe film and (b) a 52 nm GaSe film after exposure to air from 1 to 100 hours.

negligible decrease from measurement to measurement that can be attributed to the re-alignment accuracy. For GaSe, a peak at ~ 2.02 eV is observed that we relate to impurity/defect states, which typically show bright PL in this material [11]. In the material studied in this work, the free exciton feature is only observed under pulsed excitation when the impurity states are saturated [11]. In contrast to the InSe film in Fig.4.2(a), for GaSe, PL signal changes its spectral shape and decreases significantly after the film has been exposed to air for several days.

Similar PL experiment was performed for 10 thin films of different thicknesses for each material. Figure 4.3 shows a comparison of integrated PL intensity of these InSe and GaSe thin films normalized by their intensity during the first measurement (1 hour after exfoliation). In order to reveal typical trends, we averaged over two groups of films showing similar behaviour: for group 1 we selected relatively thick films, 20-60 nm for InSe and 48-75 nm for GaSe; for group 2 we included thin films of 9-12 nm for

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InSe and 10-25 nm for GaSe. For group 1 InSe films (blue squares) the PL intensity decreases by only about 12 % after 100 hours of exposure to air. For group 2 InSe films (blue dots) during the same period of time it decreases by a factor of two more, 25 %. For the GaSe films (red squares), we find that after just 24 hours exposed to air the intensity reduction is 15 % for group 1 thick films and 60 % for the thin films (red dots). Such rapid reduction of PL intensity is observed for both types of GaSe films, with the strongest effect for the case of the thin films where PL completely vanishes after 100 hours of exposure to air, whereas for the thick GaSe films, PL falls to 30 % of its initial intensity.

Similar behaviour of the signal intensities from the two materials is observed in Raman spectroscopy. In Fig. 4.4 Raman spectra of a 16 nm InSe and a 24 nm GaSe films are presented measured at room temperature. Typical modes reported previously for bulk GaSe and InSe [25, 34–37] are observed for these thin films. For InSe (top panel), the prominent peaks at 117.5, 179.5, 201.7 and 228.6 cm^{-1} correspond to the vibrational modes A'_1 , E'' , $E'(TO)$ and A'_1 respectively. In the case of GaSe (bottom panel) the modes A'_1 , $E'(TO)$, $E'(LO)$ and A'_1 at 135.6, 215.6, 308.7 cm^{-1} respectively, are observed. These Raman modes were clearly observed in both samples on the day when the samples were fabricated. The measurements were then repeated one week later, during which time both samples were stored at room temperature in air. As seen in the figure, the Raman intensity for InSe remains unchanged. In contrast, the Raman spectra for GaSe show significant variation with time: (i) the intensity drops by a factor of 10 and (ii) different relative strengths of the Raman modes are observed. These changes indicate significant modification of physical properties of GaSe films after exposure to air for 1 week.

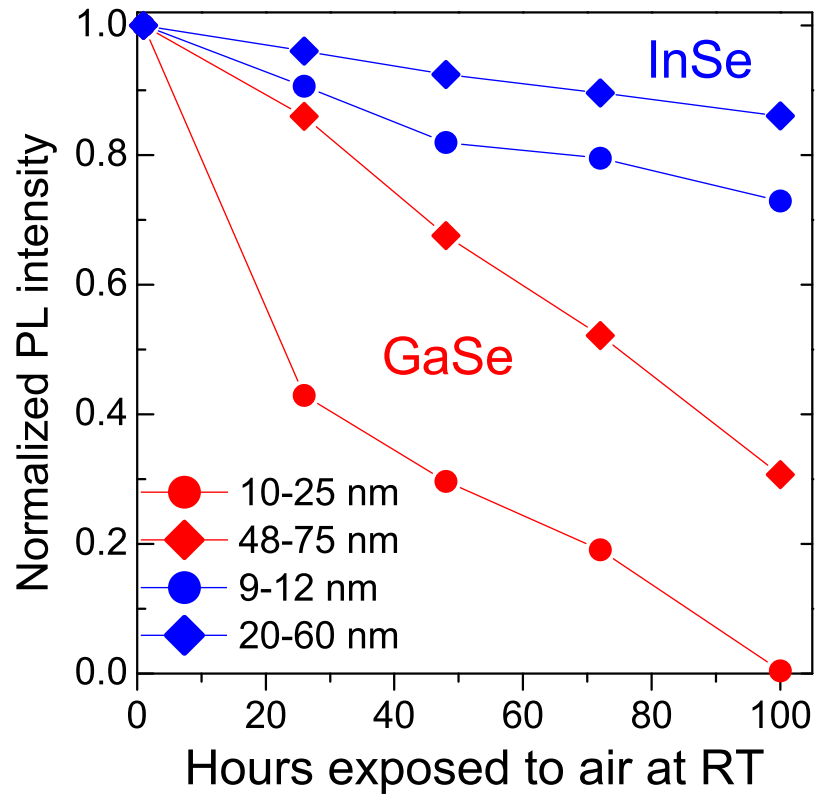


Fig. 4.3 (a) PL intensity measured at $T=10$ K as a function of time t_{exp} the InSe and GaSe thin films were exposed to air at room temperature. For InSe, the results are averaged and normalized by the intensity for $t_{exp}=1$ hour for two groups of five films with thicknesses in the range 9-12 nm (blue dots) and 20-60 nm (blue squares). For GaSe the same procedure is carried out for films with thicknesses in the range of 10-25 nm (red dots) and 48-75 nm (red squares).

Photoluminescence and Raman investigation of stability of InSe and GaSe thin films

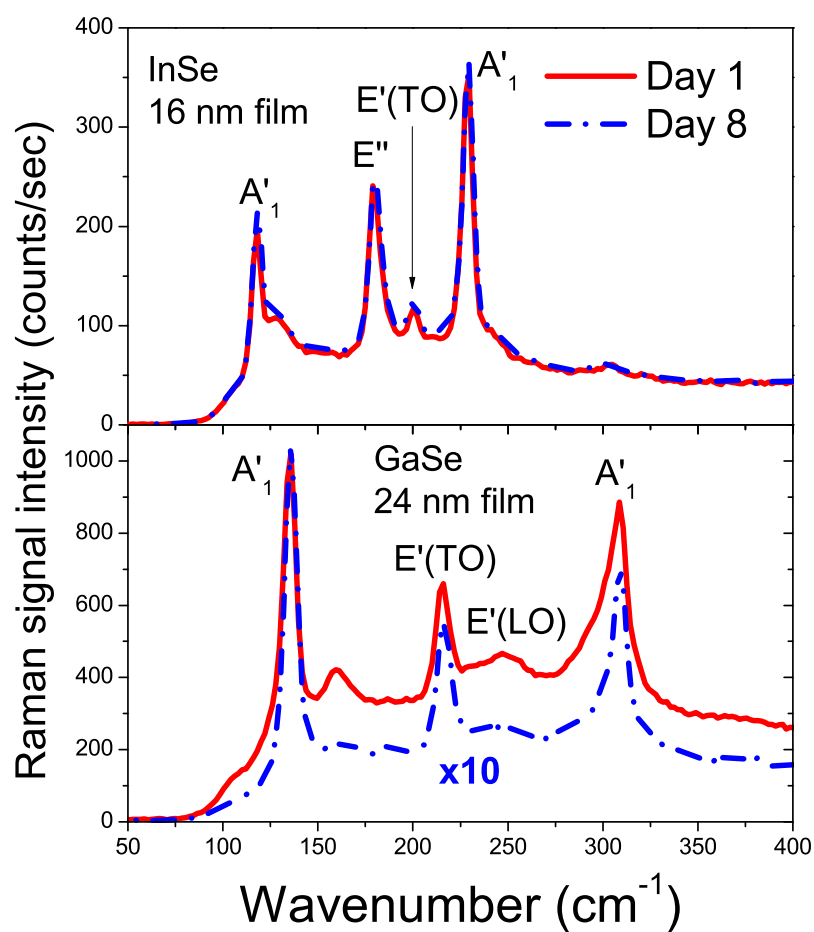


Fig. 4.4 Room temperature Raman spectra for a 16 nm InSe (top) and a 24 nm GaSe (bottom) film. The spectra were measured right after fabrication (solid line) and one week later (dashed line).

4.4 Discussion

The significant decrease of PL and Raman signals for GaSe can be attributed to the interaction of the film layers with water and oxygen in the atmosphere, similar to recent reports on black phosphorus [15, 16]. Such processes occur in the layers adjacent to the surface(s) of the film, leading to a significant change of the material properties. In a simplified way, this can be described as reduction of the effective film thickness with time when it is exposed to air, as it can be reasonably assumed that only the unaffected material would give rise to typical PL and Raman signals, whereas the eroded GaSe would show very low PL intensity (no PL emission at new wavelengths compared with the original samples were found).

Using our previous results for the detailed thickness-dependence of PL intensity in GaSe thin films presented in chapter 3, we can estimate the time-dependent changes of the effective film thickness in our present experiments. First, the PL dependence on exposure time for each film was analysed and effective thicknesses were obtained for each exposure time. This was carried out by matching the observed changes in PL intensities with those previously measured for a large number of films of different thicknesses, where we observed a strong reduction of PL intensity with the decreasing film thickness[11]. By performing this procedure, we deduced the erosion rate for each of the ten films, and finally obtained an average rate for all films of 0.14 ± 0.05 nm/hour.

In contrast, for InSe we suggest that the crystal erosion occurs just for a small number of atomic layers at the surface and then possibly saturates causing insignificant impact on PL and Raman intensities.

The fact that GaSe thin films are sensitive to the environmental effects highlights that additional techniques are required in order to protect this material from interaction with atmosphere. Alternative methods such as fabrication of samples in an

Photoluminescence and Raman investigation of stability of InSe and GaSe thin films

inert atmosphere as well as encapsulation of the films with dielectric materials like boron nitride (hBN) [38], SiO₂ or Si_xN_y[39, 40] is required. For example, thin layers of black phosphorus exposed to air have been reported to degrade over a few minutes under environmental influences such as humidity, oxygen and exposure to visible light [15, 41]. However, such processes have been reported to be suppressed if films are exfoliated in an oxygen-free atmosphere [16, 38].

In this work, we fabricated a large number of GaSe films of different thicknesses capped with 10 nm Si_xN_y or SiO₂. The films were initially produced by mechanical exfoliation in air, after which PECVD deposition was carried out. In Fig.4.5 μ PL spectra of a 32 nm GaSe film capped with Si_xN_y are presented. The spectrum shown in black was acquired one hour after fabrication, and the spectrum shown with a red curve was measured one month later. The difference between the intensity and lineshape of the two spectra is negligible, indicating significant slowing of the degradation processes. Similar degradation slow down was observed for films of any thickness for either Si_xN_y and SiO₂ capping.

4.5 Conclusions

Our results show that chemical interactions in the normal atmosphere lead to erosion of GaSe layers adjacent to the film surface with an average rate of 0.14 ± 0.05 nm/hour, leading to significant decay over time of PL and Raman intensity. Such degradation is significantly weaker for InSe thin films. We relate this behaviour of GaSe with surface oxidation and interaction with water in the atmosphere. Our findings imply that InSe will be the least demanding III-VI layered material for the use in devices based on van der Waals heterostructures, as it requires minimum protection from atmosphere. GaSe thin films are in contrast relatively unstable under ambient conditions. However, the stability of these films can be significantly improved using encapsulation with

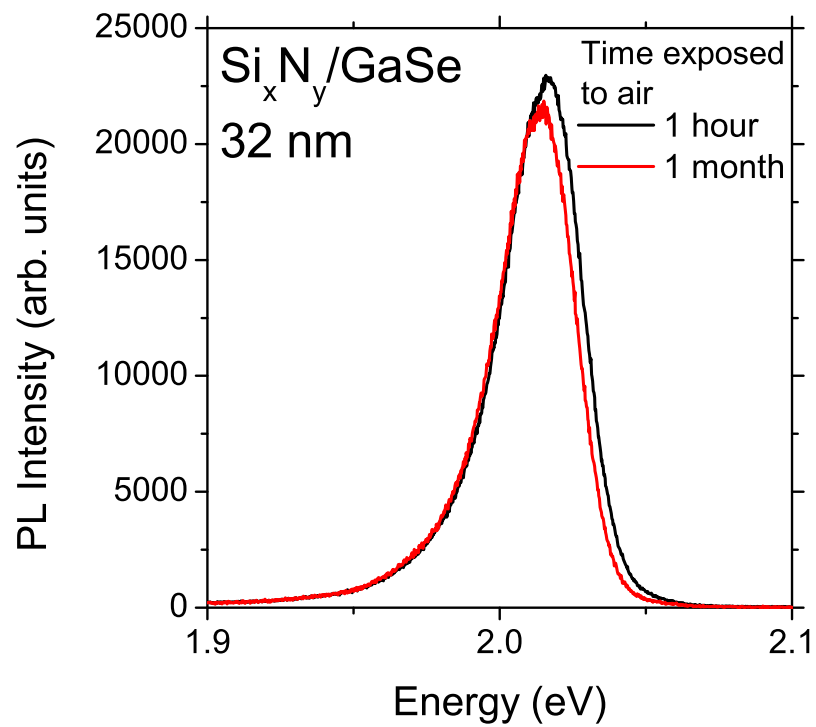


Fig. 4.5 PL spectra measured at $T=10$ K for a 32 nm thick GaSe film capped by a 10 nm PECVD layer of Si_xN_y . The film was measured one hour (black) and one month (red) after fabrication showing negligible signal decay.

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dielectrics such as Si_xN_y or SiO_2 , which slow down the degradation rate by more than two orders of magnitude.

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Chapter 5

Light-emitting diodes by band-structure engineering in van der Waals heterostructures

5.1 Introduction

The advent of graphene and related 2D materials [1, 2] has recently led to a new technology: heterostructures based on these atomically thin crystals [3]. As described in section 1.2.5, many different devices and applications have been developed. In this chapter we take the complexity and functionality of such van der Waals heterostructures to the next level by introducing quantum wells (QWs) engineered with one atomic plane precision. We describe light emitting diodes (LEDs) made by stacking up metallic graphene, insulating BN and various semiconducting monolayers into complex but carefully designed sequences. Our first devices already exhibit extrinsic quantum efficiency of nearly 10% and the emission can be tuned over a wide range of wavelengths by appropriately choosing and combining 2D semiconductors (monolayers of TMDCs).

Light-emitting diodes by band-structure engineering in van der Waals heterostructures

Two-dimensional materials, that started with graphene, now include a large variety of compounds[4]. However the real diversity can be achieved when combining these materials to make heterostructures. If we take into account their different band-structures, distinct potential landscapes for electrons to live in can be created.

Such band-structure engineering has previously been exploited to create LEDs and lasers based on semiconductor heterostructures grown by molecular beam epitaxy[5]. Here we demonstrate that using graphene as a transparent conductive coating, hexagonal boron nitride (hBN) as tunnel barriers and different TMDCs[6] as the materials for QWs, we can create efficient LEDs. In our devices, electrons and holes are injected to a layer of TMDC from the two graphene electrodes. Because of the long lifetime of the quasiparticles in the QWs (determined by the height and thickness of the neighbouring hBN barriers), electrons and holes recombine, emitting a photon. The emission wavelength can be fine-tuned by the appropriate selection of TMDC and quantum efficiency (QE) can be enhanced by using multiple QWs (MQWs).

We chose TMDC because of wide choice of such materials and the fact that monolayers of many TMDC are direct band gap semiconductors[7–11]. Until this work, electroluminescence (EL) in TMDC devices has been reported only for lateral monolayer devices and attributed to thermally assisted processes arising from impact ionization across a Schottky barrier[12] and formation of p-n junctions[11, 13, 14]. The use of vertical heterostructures allows us to improve the performance of LED's in many respects: reduced contact resistance, higher current densities allowing brighter LEDs, luminescence from the whole device area (Figs. 5.4A-B) and wider choice of TMDC and their combinations allowed in designing such heterostructures. The same technology can be extended to create other QW-based devices such as indirect excitonic devices[15], LEDs based on several different QWs and lasers.

5.2 Samples and experimental methods

Figure 5.1 shows schematically the architecture of (a) single quantum well (SQW) and (b) MQW structures. In the SQW a TMDC single layer is sandwiched by hBN, graphene (for electrical injection) and again hBN for encapsulation. In the case of MQW the two or more TMDC layer are separated by hBN (for tunnel barriers) and then the structure is completed with the sandwich as in SQW. A peel/lift van der Waals technique[16] was used to produce the devices (see section 2.1.2 for further details on device fabrication). Electrical contacts connected to the top and bottom graphene electrodes are patterned using electron-beam lithography followed by evaporation of 5 nm Cr/60 nm Au. Different QW structures comprising single and multiple layers of TMDC flakes from three different materials, MoS₂, WS₂ and WSe₂ were produced by Sheffield's collaborator Dr. F. Withers.

Cross sectional bright field scanning transmission electron microscope (STEM) images of the SQW and MQW devices, provided by Dr. S. Haigh (Manchester), demonstrate that the heterostructures are atomically flat and free from interlayer contamination [17]; Fig. 5.1b,d. The large atomic numbers for TMDC allow the semiconductor crystals to be clearly identified due to strong electron-beam scattering (dark contrast observed in Fig. 5.1b-d). Other layers were identified by energy dispersive X-ray spectroscopy. The large intensity variation partially obscures the lattice contrast between adjacent layers but, despite this, the hBN lattice fringes can clearly be seen in Figs. 5.1b-d. The different contrast of the four MoS₂ monolayers in the MQW of Fig. 5.1d is attributed to their different crystallographic orientations, (confirmed by rotating the sample around the heterostructure's vertical direction which changes the relative intensity of different layers).

Optical characterization of the devices was carried out using micro-photoluminescence and electroluminescence spectroscopy within a helium flow cryostat at 6 K. The setup

Light-emitting diodes by band-structure engineering in van der Waals heterostructures

is described in sections 2.2.1 and 2.2.1. A 532 nm laser was used for PL measurements, focused to a spot size of $\sim 1\text{-}2\ \mu\text{m}$ using a 50x objective. EL was produced with a voltage source (Keithley 2400) with different bias voltages depending on device characteristics.

5.3 Results

In fig. 5.2(A-C) the band diagrams of a SQW based on MoS₂ (that can be extended to any TMDC) for the case of zero applied bias (A), intermediate bias V_b (B) and high bias (C) are shown. It shows how electrons and holes tunnel through the hBN barrier when V_b reach the onset bias. When this happens, excitons are created and able to recombine to create photons. Here the charge neutrality point of graphene lies 1.5 eV above the valence band of h-BN and 0.5 eV below the conduction bands of MoS₂ as found earlier in tunnelling experiments [18]. The devices often demonstrate some small asymmetry with respect to the reversal of V_b , which we associate with unintentional chemical doping being slightly different for Gr_B and Gr_T. In nearly all devices, EL could be observed both for positive and negative V_b , though with somewhat different intensity.

To gain insight into the operation of the SQW devices current-voltage characteristics and photoluminescence (PL) and EL spectra were measured. Three different SQW LEDs were characterised using MoS₂, WS₂ and WSe₂. The concerned heterostructures are shown schematically in Fig. 5.1(a).

5.3.1 MoS₂ single quantum well LED

In Fig. 5.3(a-c) the spectral characterisation of the MoS₂ SQW device (named SL MoS₂-1) at T=6 K is presented. Fig.5.3(a) shows the PL bias (V_b) dependence from

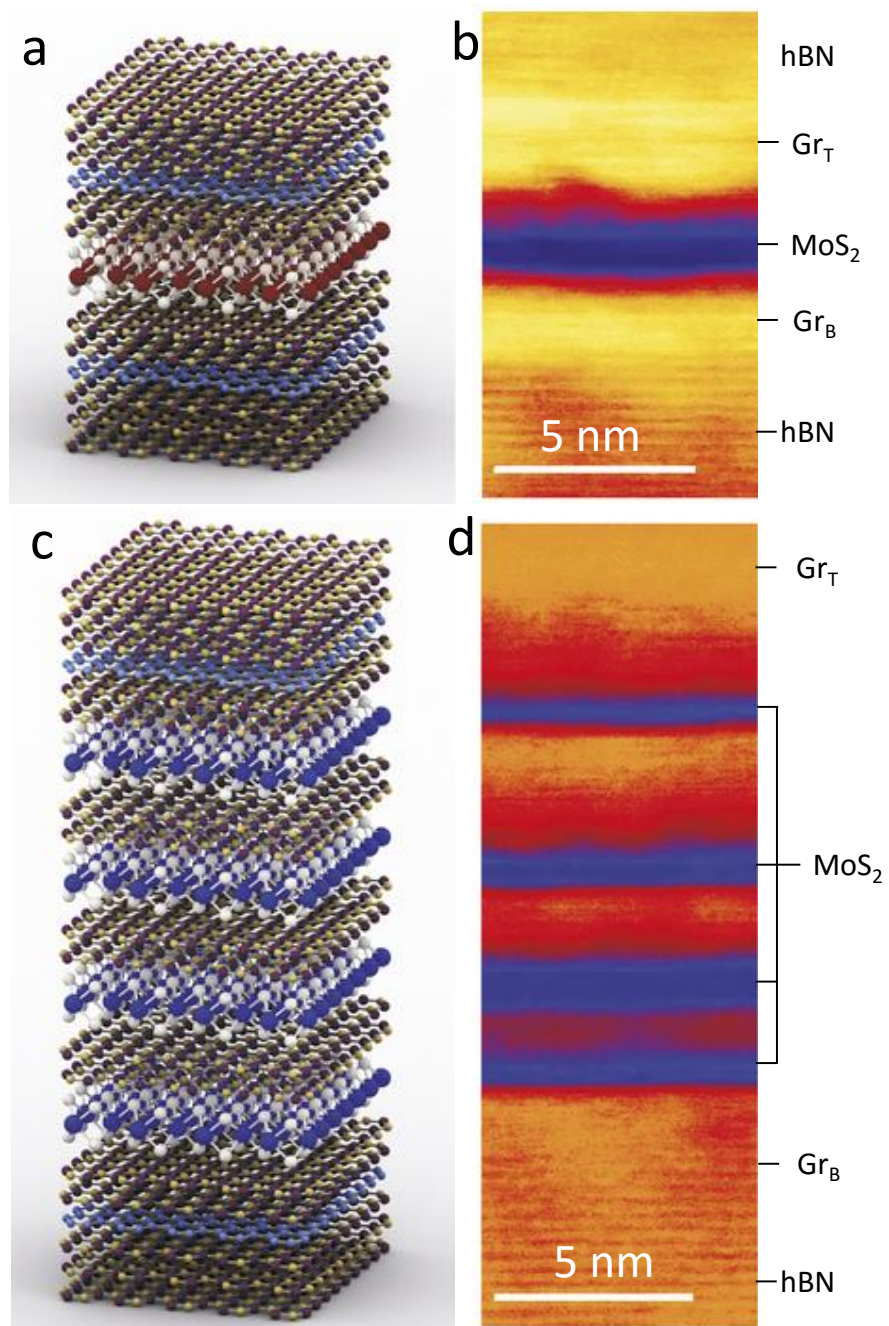


Fig. 5.1 Schematics of the (a) SQW heterostructure $\text{hBN}/\text{Gr}_B/2\text{hBN}/\text{MoS}_2/2\text{hBN}/\text{Gr}_T/\text{hBN}$. (b) Cross-sectional bright field STEM image of the type of heterostructures presented in (a). (c) MQW heterostructure $\text{hBN}/\text{Gr}_B/2\text{hBN}/\text{MoS}_2/2\text{hBN}/\text{MoS}_2/2\text{hBN}/\text{MoS}_2/2\text{hBN}/\text{MoS}_2/\text{Gr}_T/\text{hBN}$. (d) Cross-sectional bright field STEM image of the type of heterostructures presented in (c).

Light-emitting diodes by band-structure engineering in van der Waals heterostructures

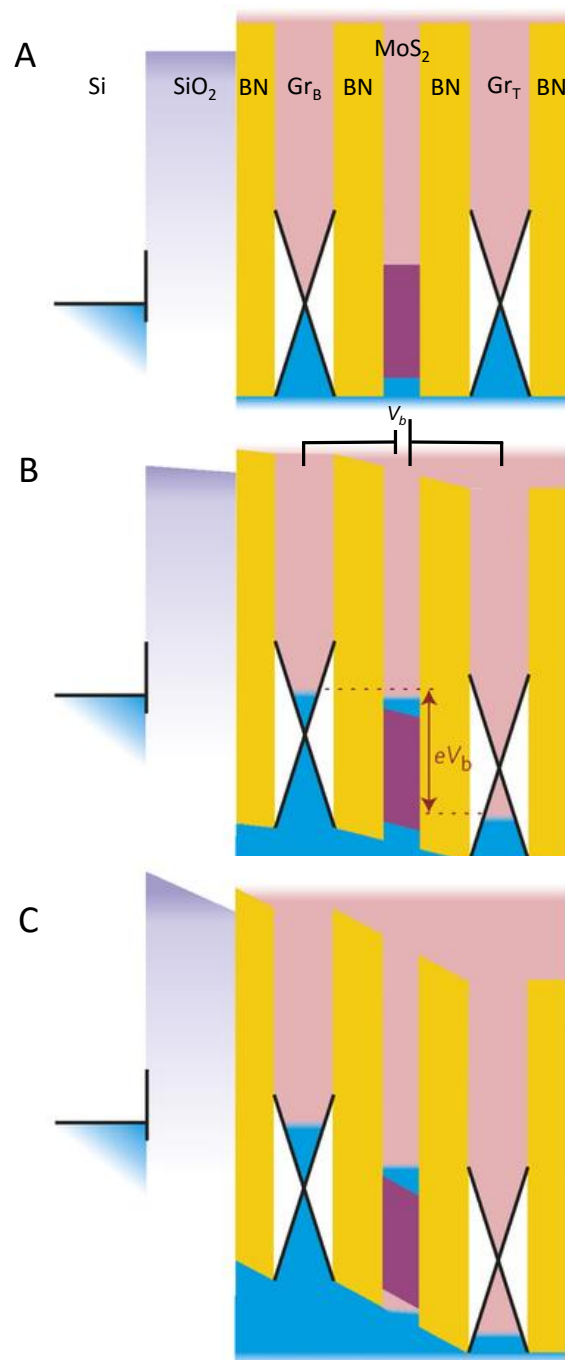


Fig. 5.2 Band diagrams for the case of zero applied bias (A), intermediate applied bias V_b (B) and high bias (C) for the heterostructure presented in Fig. 5.1(a), electrons and holes recombine to produce EL with energy $\hbar\omega_{EL}$.

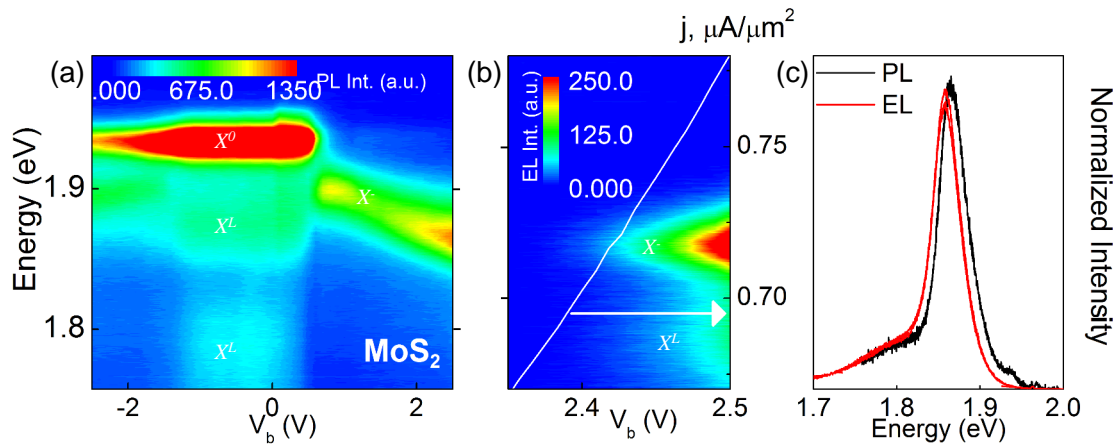


Fig. 5.3 Optical and transport characterization of a MoS₂ SQW device (SL MoS₂-1). (a) PL bias dependence of MoS₂ LED. (b) EL bias dependence of MoS₂ LED. (c) Comparison of EL and PL spectra for MoS₂ LED at 2.3 V.

-2.2 to 2.2 V. At low V_b , the PL is dominated by the peak at 1.93 eV, which we interpret as the neutral A exciton, X^0 [8]. We attribute the two weaker and broader peaks at 1.87 and 1.79 eV to bound excitons, X^L [19, 20]. For $-0.8 \text{ V} < V_b < 0.6 \text{ V}$ no significant change occurs in the spectrum. However, at $V_b = 0.65 \text{ V}$, the PL spectrum changes abruptly with another peak emerging at a lower energy of 1.90 eV. We explain this transition as being due to the fact that at this voltage the Fermi level in Gr_B rises above the conduction band in MoS₂, allowing injection of electrons into the QW (Fig. 5.3(c)). This leads not only to an increase in tunnelling conductivity but, also, to accumulation of electrons in MoS₂ and formation of negatively charged excitons, X^- [13]. The X^- peak is positioned at a lower energy compared to the X^0 peak due to the binding energy, E_B , of X^- . In the case of MoS₂ it is estimated as $\sim 36 \text{ meV}$ near the onset of X^- . As the bias is increased, the energy of the X^- shifts to lower values which we attribute to the quantum confined Stark effect or increase in the Fermi energy in MoS₂ [8].

In contrast to PL, EL starts only at V_b above a certain threshold (Fig. 5.3b). Such

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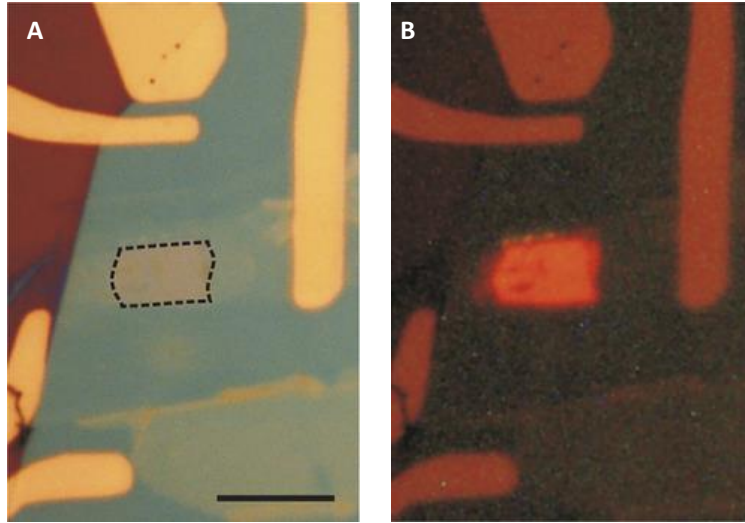


Fig. 5.4 Optical image of a SQW device based on MoS₂ (SL MoS₂-1). (A) Device with no voltage applied, the dashed line outlines the heterostructure area. Scale bar, 10 μm . (B) The same device emitting light with a voltage of 2.5 V.

behaviour is associated with the Fermi level of Gr_T being brought below the edge of the valence band so that holes can be injected to MoS₂ from Gr_T (in addition to electrons already injected from Gr_B) as sketched in Fig. 5.2C. This creates conditions for exciton formation inside the QW and their radiative recombination.

We find that the EL frequency is close to that of PL at V_b 2.4 V (Figs. 5.3a-c), which allows us to attribute the EL to radiative recombination of X^- . A comparison of both EL and PL spectra is shown in fig. 5.3c with $V_b=2.4$ V. A real image of this device is presented in fig. 5.4, (A) shows the device with no voltage applied. Here dashed line highlight the overlapping area of the layers. In (B) this area emits red light due to a voltage $V_b=2.5$ V applied through the graphene contacts. The black scale bar corresponds to 10 μm .

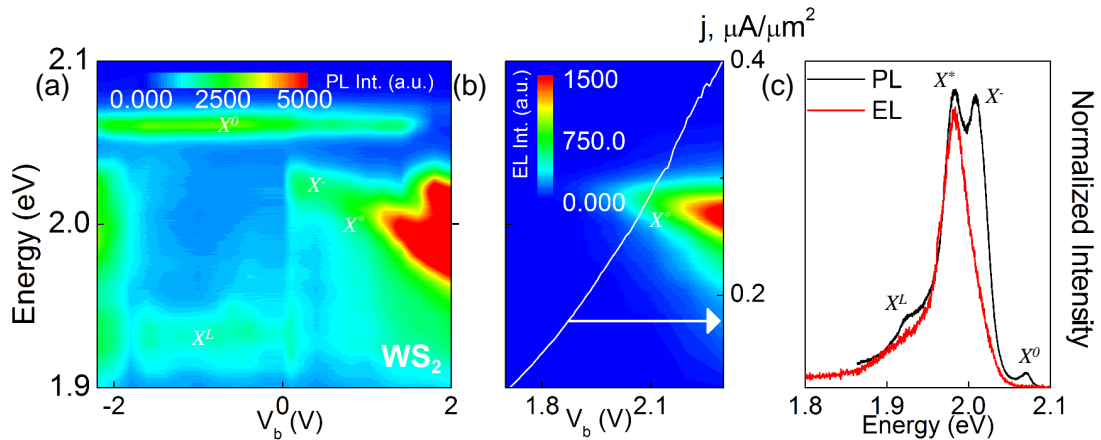


Fig. 5.5 Optical and transport characterization of a WS_2 SQW device (SL WS_2 -1). (a) PL bias dependence of WS_2 LED. (b) EL bias dependence of WS_2 LED. (c) Comparison of EL and PL spectra for WS_2 LED at 2 V.

5.3.2 WS_2 single quantum well LED

Similar behaviour is observed for a WS_2 QW device (SL WS_2 -1), see Figs. 5.5a-c. PL spectra show a neutral exciton X^0 peak [21] at 2.06 eV from -2.1 to 1.8 V (fig. 5.5a). At positive bias V_b a trion peak X^- is tuned from 2.01 eV to lower energies down to 1.95 eV. At negative bias an additional peak at lower energy (1.91 eV) is observed, this emission is associated with the bound excitons [19]. EL arises from ~ 1.9 V from X^* , as can be seen in fig. 5.5b. White line in this figure shows the current density as a function of V_b . The difference in EL and PL spectra is notable in this material at 2V (fig 5.5c), showing two pronounced peaks in PL and only one peak and a shoulder at lower energy beyond X^0 and X^- peaks in EL, the origin of other spectral lines cannot be established at present with certainty. It is likely that X^* peak correspond to emission from impurity-bound excitons, similarly to X^L .

5.3.3 Multiple quantum well LED devices

As described above, MQW devices were fabricated. The same optical characterisation as for SQW devices was carried out.

Figs. 5.6 and 5.7 show results for two such MQW structures with three (3MQW MoS₂-1) and four MoS₂ QWs (4MQW MoS₂-1), respectively. The current increases with V_b in a step-like manner, which is attributed to sequential switching of tunnelling through individual MoS₂ QW (see Figs. 5.6 and 5.7).

In the three-QW device PL is qualitatively similar to that of SQW devices (Fig. 5.6a) but the X_0 peak is replaced with X^- peak at a $V_b=-0.4$ V. The X^0 peak reappears again at $V_b>1.2$ V. This can be explained by charge redistribution between different QWs. The EL first becomes observable at $V_b>3.9$ V and j of 1.8 nA/ μm^2 ; Fig. 5.6d. This current density is nearly 2 orders of magnitude smaller than the threshold current required to see EL in similar SQWs. Figure 5.6e shows EL and PL spectra where it can be seen that both X^0 (width 24 meV) and X^- (width 78 meV) peaks appear in EL at energies of 1.94 and 1.86 eV, respectively, similar to their positions in SQWs. The width of the X^- exciton line, however, has been significantly broadened compared to the SQWs.

For the four-QW device, PL spectra are dominated by X_0 and X^- (Fig. 5.7b). A broad peak at 1.6 eV appears at negative bias from 2.5 V originating from the localized excitons X^L . EL is seen from 0.9 nA/ μm^2 (blue curve in Fig. 5.7d), less than in the 3 QWs device with a photon energy of 1.8 eV originated from only X^- . Two more individual spectra with higher injection currents are shown in Fig. 5.7d in green and red. Comparison of both EL and PL at 7 V is observed in Fig. 5.7e.

The described technology of making designer MQWs offers the possibility of combining various semiconductor TMDCs in one device. Figs. 5.8(a-c) describe an LED made from WSe₂ and MoS₂ QWs: Si/SiO₂/hBN/Gr_B/3hBN/WSe₂/3hBN/MoS₂/3hBN/Gr_T/hBN

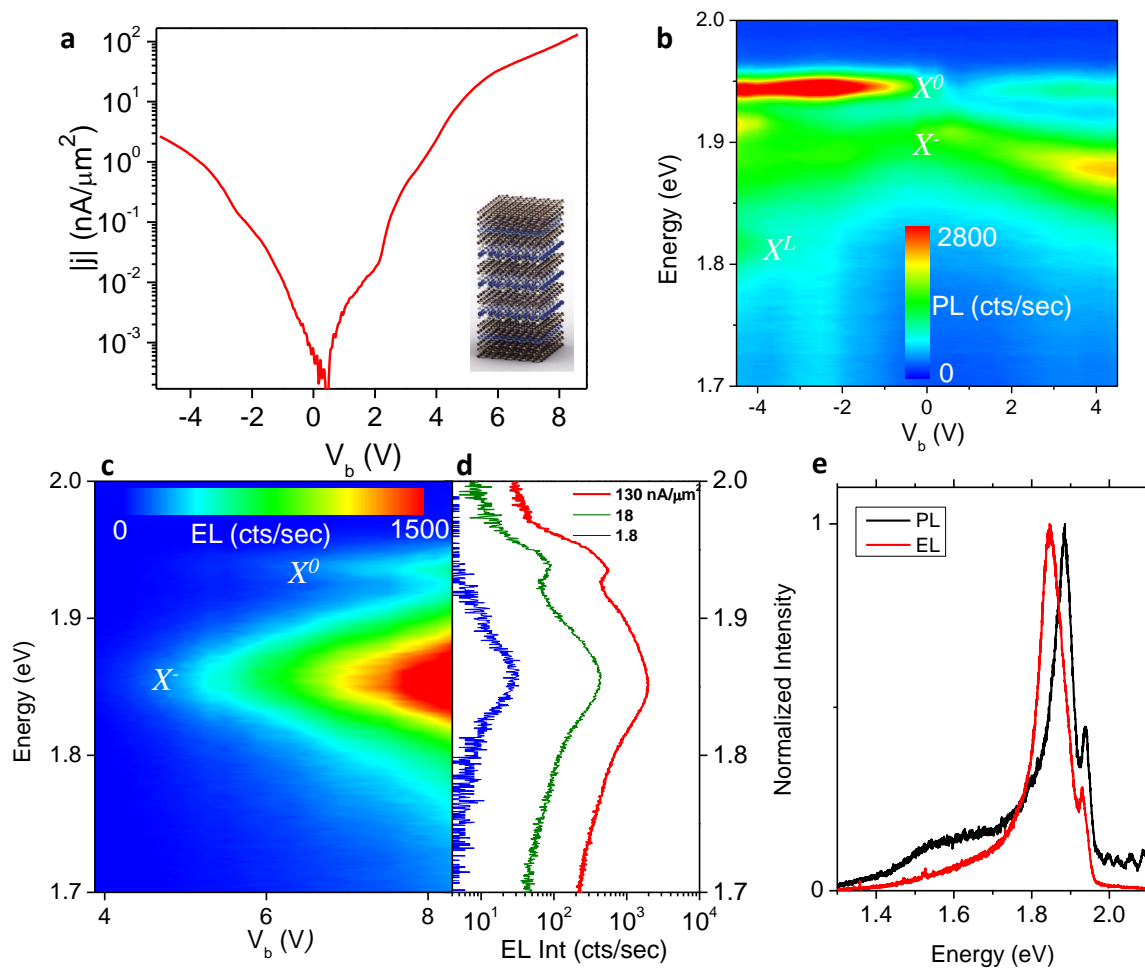


Fig. 5.6 (a) Modulus of the current density through a tree-QW structure based on MoS₂ (3MQW MoS₂-1). (b,c) Maps of PL and EL spectra for this device. (d) Individual EL spectra plotted on logarithmic scale to show the onset of EL at 1.8 nA/μm² (blue curve). (f) Comparison of EL and PL spectra.

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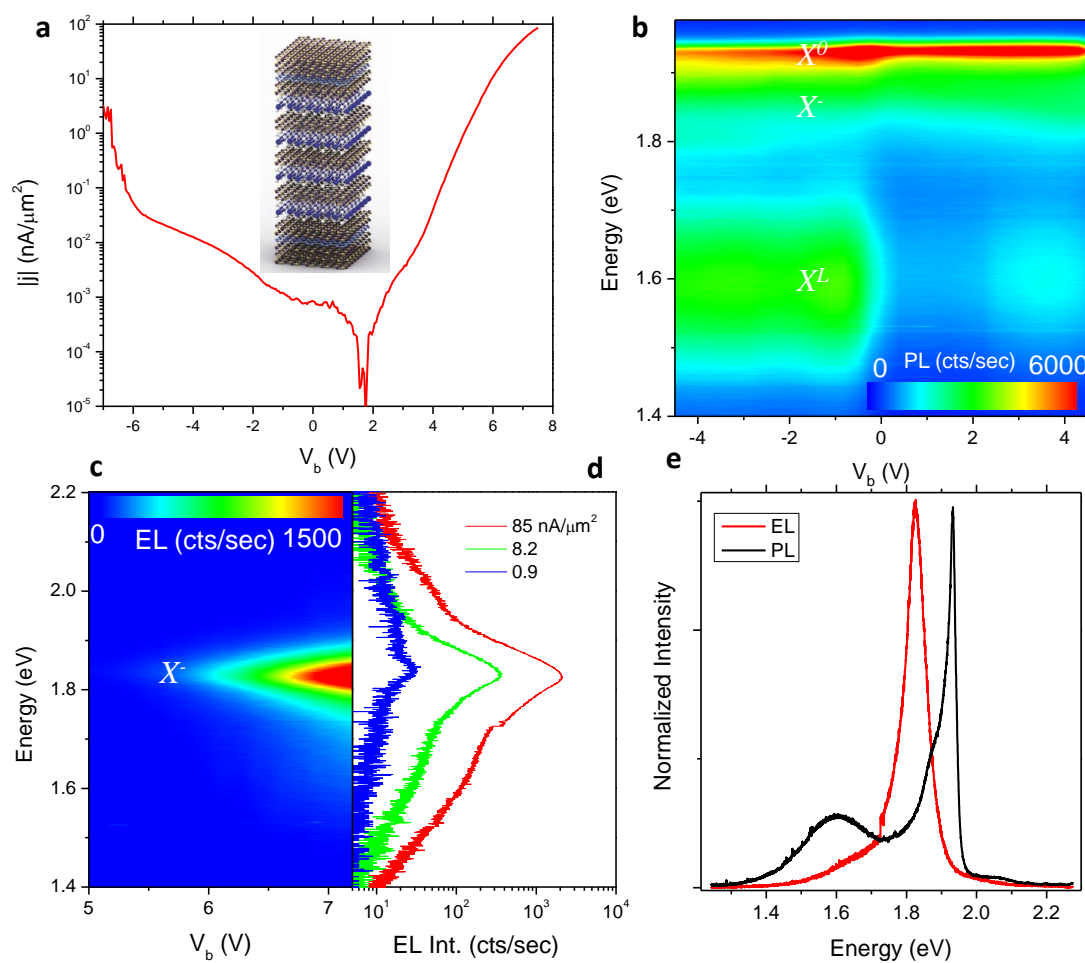


Fig. 5.7 (a) Modulus of the current density through a four-QW structure based on MoS₂ (4MQW MoS₂-1). (b,c) Maps of PL and EL spectra for this device. (d) Individual EL spectra plotted on logarithmic scale to show the onset of EL at 0.9 nA/ μm^2 (blue curve). (e) Comparison of EL and PL spectra.

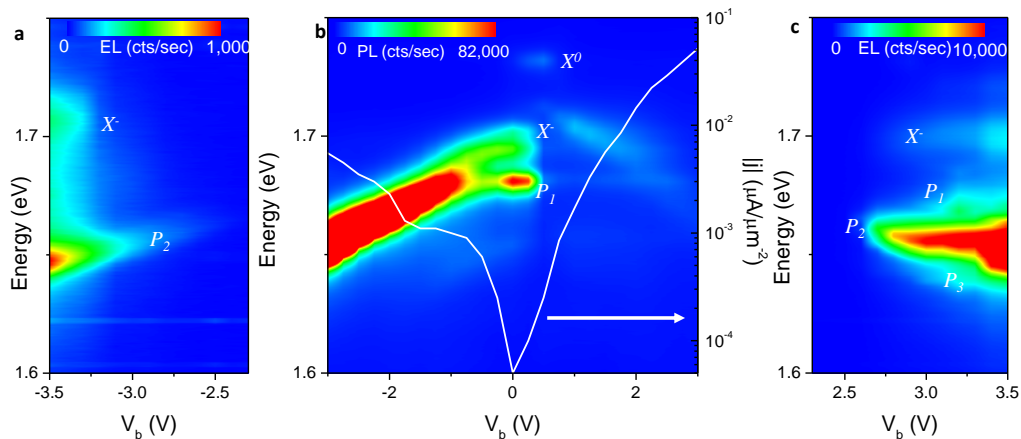


Fig. 5.8 Multiple QW well: hBN/Gr_B/3hBN/WSe₂/3hBN/MoS₂/3hBN/Gr_T/hBN (WSe₂/MoS₂ QW-1). (a) EL intensity map for negative bias. (b) PL intensity map bias dependence. (c) EL intensity map for positive bias.

(WSe₂/MoS₂ QW-1). EL and PL occur here in the low energy part of the spectra and can be associated with excitonic states proposed by You *et al.* [22] in WSe₂. EL is shown in 5.8(a and c) and originates from both X^- and P_2 for negative bias and from X^- , P_1 , P_2 and P_3 for positive bias. In the case of PL (5.8b)) it is much brighter at negative bias showing a notable shift of emission from 1.7 eV to lower energies for X^- and, less intense, P_1 . At positive bias a relatively weak emission from X^- is seen. X^0 is observed with very low intensity at 1.72 eV and $V_b \approx 0$ V. In comparison with SQW devices, the combinational device in Fig. 5.8 exhibits more than an order of magnitude stronger both PL and EL. We associate this with charge transfer between the MoS₂ and WSe₂ layers such that electron-hole pairs are created in both layers but transfer to and recombine in the material with the smaller band-gap [23]. Such process is expected to depend strongly on band alignment, which is controlled by bias and gate voltages. This explains the complex, asymmetric V_b dependence of PL and EL in Fig. 5.8b.

5.4 Discussion

EL is seen from all nine devices characterised showing different intensities and emission wavelengths depending on the active material. An important parameter for any light emission device is the QE that is defined as the number of emitted photons N to the number of injected electron-hole pairs, as discussed in section 2.2.1. To perform an accurate estimation of EQE, a detailed study of the losses in the system was performed in order to obtain a relation between the number of counts taken with the CCD and the current. This study is presented in Appendix B.

For the SQW structures we estimate η up to 1%, this value by itself is ten times larger than that of planar p-n diodes [11, 13, 14] and 100 times better than EL from Schottky barrier devices based on single atomic layer TMDCs [12]. Our rough estimations show that the EQE for PL is lower than that for EL. Relatively low EQE found in PL indicates that the crystal quality itself requires improvement and that even higher EQE in EL may then be achieved [21]. Further improvements to η could be expected, if fabrication procedures are developed to reduce the exposure of the crystals to oxygen and atmospheric water, which may lead to their partial oxidation.

Another route to improve the QE is to increase the time quasiparticles spend in the QWs. To this end, we have employed multiple QWs stacked in series (see figs. 5.6 and 5.7), which increases the overall thickness of the tunnel barrier and enhances the probability for injected carriers to recombine radiatively. For these devices, η reaches values of $\sim 8.4\%$ for four QW MoS₂ device (4MQW MoS₂-1) and 5.4% for MoS₂/WSe₂ LED (MoS₂/WSe₂ QW-2), figs. 5.8. These values of QE are comparable to modern-day organic LED lighting and the concept is compatible with the popular idea of flexible and transparent electronics due to the thickness of 10–40 atoms.

A summary of all devices studied showing EL peak position, linewidth and quantum efficiency at highest injection current at 6 K is presented in table 5.1. Linewidths

Device	EL Peak Position (eV)	FWHM (meV)	QE (%)
SL MoS ₂ - 1	1.86	30	0.30
SL MoS ₂ - 2	1.81	90	0.51
SL WS ₂	2.00	23	1.32
3MQW MoS ₂	1.94	78	6.00
4MQW MoS ₂	1.83	65	8.40
WSe ₂ /MoS ₂ QW-1	1.66	14	4.80
WSe ₂ /MoS ₂ QW-2	1.68	21	5.40

Table 5.1 The main peak position, width and quantum efficiency at highest injection current T=6 K for the electroluminescence spectra for different devices based on TMDC materials.

as narrow as 14 meV were observed in MoS₂/WSe₂ devices.

5.5 Conclusions

In summary, we have demonstrated band-structure engineering with one atomic layer precision by creating QW heterostructures from various 2D crystals including several TMDC, hBN and graphene. Our LEDs based on a single QW already exhibit quantum efficiency of the order of 1% and linewidths down to 14 meV, despite the relatively poor quality of currently available TMDC layers. This QE can be improved significantly by using multiple QWs, and LED devices consisting of 3 to 4 QWs show the QE up to 8.4% at low temperature. Combining different 2D semiconductor materials allows fine tuning of the emission spectra, and the demonstrated concept is compatible with the popular idea of flexible and transparent electronics. The range of functionalities for the demonstrated heterostructures is expected to grow further with the rapid progress in technology of chemical vapour deposition growth that will allow scaling up of production of similar devices.

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Chapter 6

Temperature-dependent behaviour of MoSe₂ and WSe₂ van der Waals heterostructure LEDs

6.1 Introduction

Recently molybdenum and tungsten dichalcogenides (MoX₂ and WX₂), have attracted considerable attention following the discovery of the indirect-to-direct band gap transition [1, 2] and the coupling of the spin and valley degrees of freedom in atomically thin layers[3]. Both in WX₂ and MoX₂, electrons and holes form excitons exhibiting very high binding energies [4–6] and stability at room temperature [1–3, 7, 8], making them attractive for optoelectronic devices.

Further to this, an important band-structure property arising from the strong spin-orbit interaction in these compounds has been predicted by density functional theory [9, 10] (see Fig. 6.1A-B): in WX₂ (MoX₂) the lowest energy states in the conduction band and the highest energy states in the valence band have the opposite (same) spin orientations preventing (enabling) their recombination with emission of a photon. For

Temperature-dependent behaviour of MoSe₂ and WSe₂ van der Waals heterostructure LEDs

this reason, according to the spin-state ordering predicted by the theory, the lowest energy excitonic sub-band in WX₂ corresponds to dark excitons, separated from the bright exciton sub-band by the energy approximately equal to the electron spin-orbit splitting Δ_{SO} of the order of 30-40 meV [9–12]. In this chapter we will show that such band-structure properties of WX₂ strongly impact on the LED performance, similar to the structures presented in chapter 5, leading to significant enhancement of the room temperature external quantum efficiency of $\sim 10\%$ for the WSe₂ LEDs in the EL regime. A comparison with a MoSe₂ LED will be carried out, where a more common behaviour is observed: EL intensity falls by up to 100 times when the temperature is varied from 10 to 300 K leading to a reduction of the quantum efficiency.

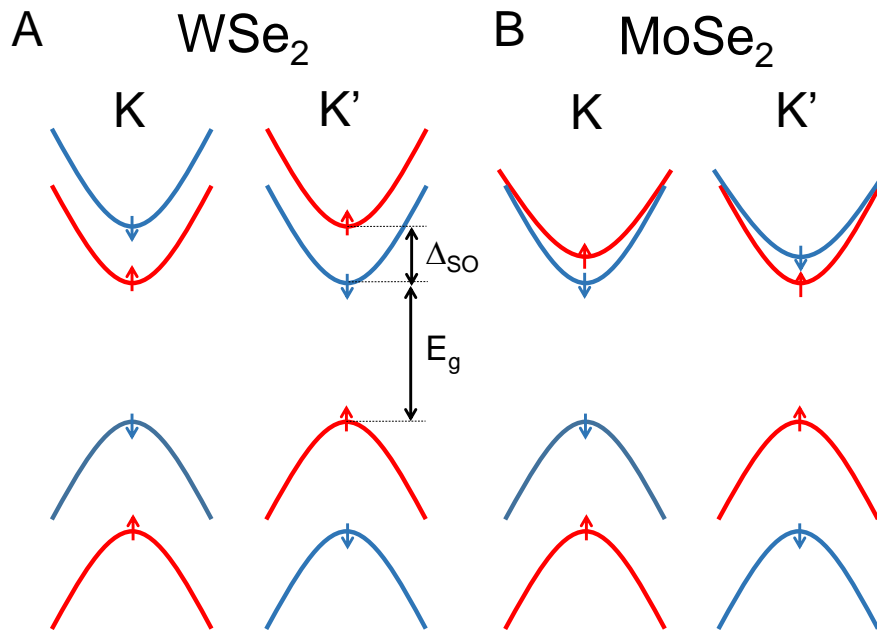


Fig. 6.1 Schematic representation of the band structure of (A) WSe₂ and (B) MoSe₂. Red and blue arrows denote spin orientation.

6.2 Samples and experimental methods

The LEDs used in this chapter are identical to the devices presented in chapter 5 fabricated with the lift/peel method described in section 2.1.2. In the present case the TMDCs used were single layers of WSe_2 and MoSe_2 . Fig. 6.2A shows schematically the structure: $\text{Gr}_B/3\text{hBN}/\text{WSe}_2/3\text{hBN}/\text{Gr}_T$. A bias voltage V_B is applied between the two graphene contacts, hBN layers act as tunnel barriers and the WSe_2 is the active semiconductor layer.

Optical characterisation of the LEDs was carried out using μPL setup shown in fig. 2.8 with a 532 nm diode laser for excitation and a helium flow cryostat that allows temperature dependent measurements from 6 to 300 K. EL was obtained applying bias voltage with a source-meter device (Keithley 2400).

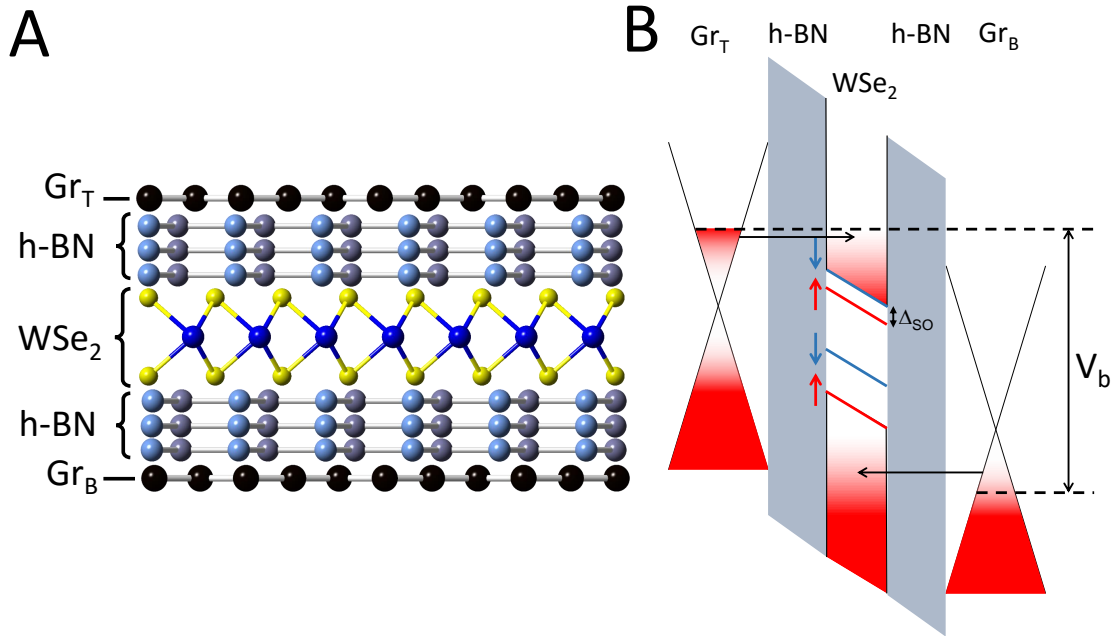


Fig. 6.2 (A) Schematic of the WSe_2 LED architecture. (B) Band alignment at high bias of a WSe_2 LED.

6.3 Results

Fig. 6.2B shows the band structure of an LED based on WSe₂. By applying bias voltage, V_b , between the two graphene electrodes, it is possible to set conditions for electron (hole) injection into the conduction (valence) band of the WSe₂ layer. The hBN barriers act to control the lifetime of the charge carriers inside the TMDC layer leading to increased probability of radiative recombination: for a long enough dwell time of the charge carriers controlled by the thickness of hBN, excitons forming in the WSe₂ layer can recombine with light emission. The same effect occurs in a MoSe₂ based LED. Full optical characterisation including EL and PL at different temperatures was performed for both structures in order to make a comparison and analyse the results influenced by the spin-orbit effects described above.

6.3.1 MoSe₂ device

As the temperature dependence of the LEDs is the key study of this work, it is necessary to describe the behaviour of the excitons in the devices. For this reason a temperature dependence of PL was carried out. This is presented in fig. 6.3A-D. In 6.3A, a PL spectrum at 5 K can be observed, the two well resolved peaks correspond to X^0 (1.65 eV) and X^- (1.62 eV, binding energy $E_c=30$ meV) with linewidths of 10.4 meV and 8.6 meV respectively (calculated with the lorentzian fits shown in blue and red solid lines, respectively). In 6.3B, a normalized temperature dependence from 5 to 300 K is shown. X^- line dominates up to 55 K when it starts vanishing while X^0 remains up to room temperature. The peak positions of both lines are plotted in 6.3C as a function of temperature in order to see the band gap (E_g) dependence with T (red circles for X^0 and black squares for X^-). Solid lines are fittings done using the

Bose-Einstein model [13]:

$$E_g(T) = E_g(0) - \frac{2\alpha_B}{e^{\frac{\Theta_E}{T}} - 1} \quad (6.1)$$

where $E_g(0)$ is the band gap at base temperature, α_B represents the strength of the exciton-phonon interaction and Θ_E corresponds to the average phonon temperature. The values obtained were $E_g(0)=1.648$ eV, $\alpha_B=45$ meV and $\Theta_E=250$ K for X^0 and $E_g(0)=1.618$ eV, $\alpha_B=50$ meV and $\Theta_E=250$ K for X^- . In 6.3D, Arrhenius fits using the following equation:

$$\frac{I(T)}{I(T = 5K)} = \frac{1}{1 + e^{-E_1/k_B T}} \quad (6.2)$$

for PL intensities of both lines are shown, this allows to calculate a activation energies E_1 of 20 and 25 eV for X^0 and X^- , respectively. This plot clearly shows how X^- (black squares) decays faster with temperature than X^0 (red circles).

Once knowing the excitonic behaviour of MoSe₂ through PL, electrically driven experiments on LEDs were performed. This is summarized in fig. 6.4A-F. PL bias-dependence at T=6 K is shown in fig. 6.4B, X^0 and X^- are observed for $|V_b| < 1.5$ V at 1.63 meV and 1.61 meV, respectively. X^0 gets weaker for $|V_b| > 1.5$ V. The current density is plotted with a white line in fig. 6.4B). Fig. 6.4A shows EL negative bias-dependence, where strong emission appears when $V_b < -1.85$ V from X^- . A different behaviour is seen for positive bias (fig. 6.4C) where EL, from both X^0 and X^- , is 1000 times weaker possibly due to asymmetry in the thickness of hBN barriers. An EL spectrum for $V_b=2$ V is plotted in fig. 6.4E showing features similar to those in the PL regime. Temperature dependent EL measurements were also performed: in fig. 6.4D spectra taken at 6 K (blue), 150 K (green) and 300 K (red) for $V_b=-2$ V are presented showing a significant EL reduction with T. For positive bias, EL rapidly decreases to zero when $T \simeq 40$ K. The Arrhenius fit in 6.4F for EL at -2V shows an

Temperature-dependent behaviour of MoSe₂ and WSe₂ van der Waals heterostructure LEDs

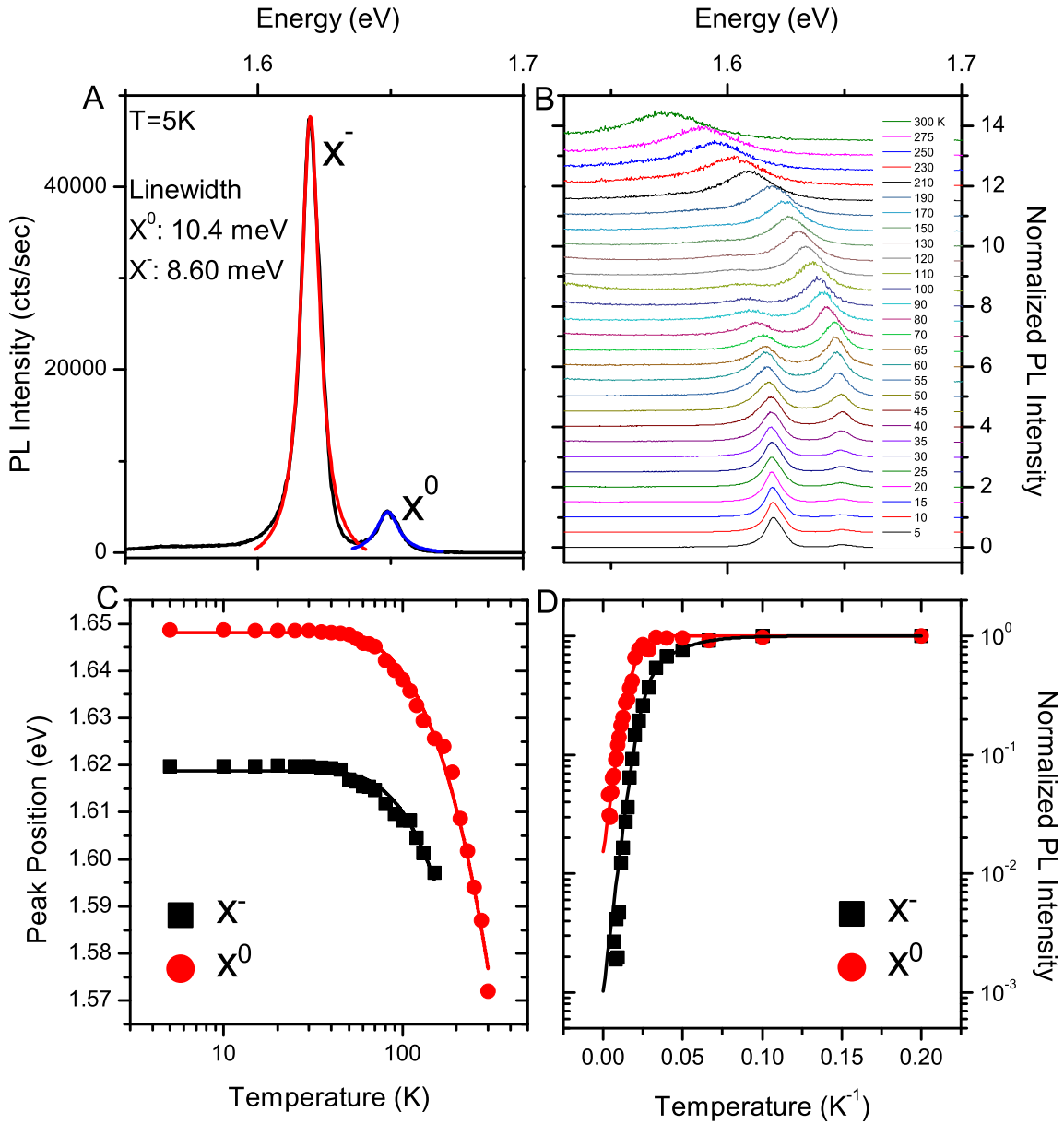


Fig. 6.3 (A) Typical PL spectrum of MoSe₂ LED taken at $T=5\text{ K}$ with $10\ \mu\text{W}$ showing two characteristic peaks attributed to a neutral (X^0) and charged (X^-) exciton with a measured linewidth of 10.4 meV and 8.6 meV respectively obtained from the Lorentzian fittings (red and blue solid lines). (B) Normalized PL temperature dependence of MoSe₂ LED from 5 to 300 K. (C) Peak positions of X^0 (red) and X^- (black) as a function of temperature. Solid lines indicate fittings with Bose-Einstein model (eq. 6.1). (D) Arrhenius plot of the PL yield for both X^0 (red) and X^- (black), fittings with eq. 6.2 are shown in red and solid lines to calculate activation energies E_1 of 20 and 25 meV for X^0 and X^- , respectively.

activation energy of 30 meV.

6.3.2 WSe₂ device

Similar to the experiments presented for a MoSe₂ LED, PL was first studied in WSe₂. Fig. 6.5A shows a PL spectra taken at T=6 K and excitation power of 50 μ W. Using the notation adopted in You et al [14], the peaks observed are neutral exciton X^0 (1.733 eV, blue line) and trion X^- (1.70 eV, red) as well as a number of additional features, with the most pronounced peaks P_1 (1.675 eV) and P_2 (1.652 eV). Linewidths obtained with lorentzian fits are shown in the graph. In fig. 6.5B, normalized PL temperature dependence from 6 to 300 K can be observed. X^0 is visible at all temperatures while the three additional peaks eventually disappear with P_1 and P_2 becoming negligible above 80 K. The band gap temperature dependence for X^0 and X^- is plotted in fig. 6.5C. By doing the same analysis as in MoSe₂ using eq. 6.1, the following values are calculated: $E_g(0)=1.734$ eV, $\alpha_B=70$ meV and $\Theta_E=270$ K for X^0 and $E_g(0)=1.706$ eV, $\alpha_B=60$ meV and $\Theta_E=270$ K for X^- . The Arrhenius plot in 6.5D was obtained with integrated PL intensities over the whole spectra for all temperatures to calculate an activation energy of ~ 40 meV. Strikingly PL intensity increases with temperature, opposite to the behaviour observed in MoSe₂.

Fig. 6.6A,C shows typical light emission behaviour of a WSe₂ LED at low T=6K. 6.6A,C show that for biases $|V_b|>2$ V, strong EL is observed. The peaks in EL spectra are clearly identifiable, using the description of PL features as shown in figs. 6.6A,C. For negative bias, EL originates from X^- and P_1 while for positive bias the four peaks X^0 , X^- , P_1 and P_2 are observed. Similar to PL, when T increases, the neutral exciton line dominates. This can be corroborated in fig. 6.6D where EL spectra for $V_b=2.3$ V at 6K (black), 20 K (red), 40 K (green) and 80 K (blue) are plotted. It is worth noting that, although features P_1 , P_2 are always present in all the LEDs measured,

Temperature-dependent behaviour of MoSe₂ and WSe₂ van der Waals heterostructure LEDs

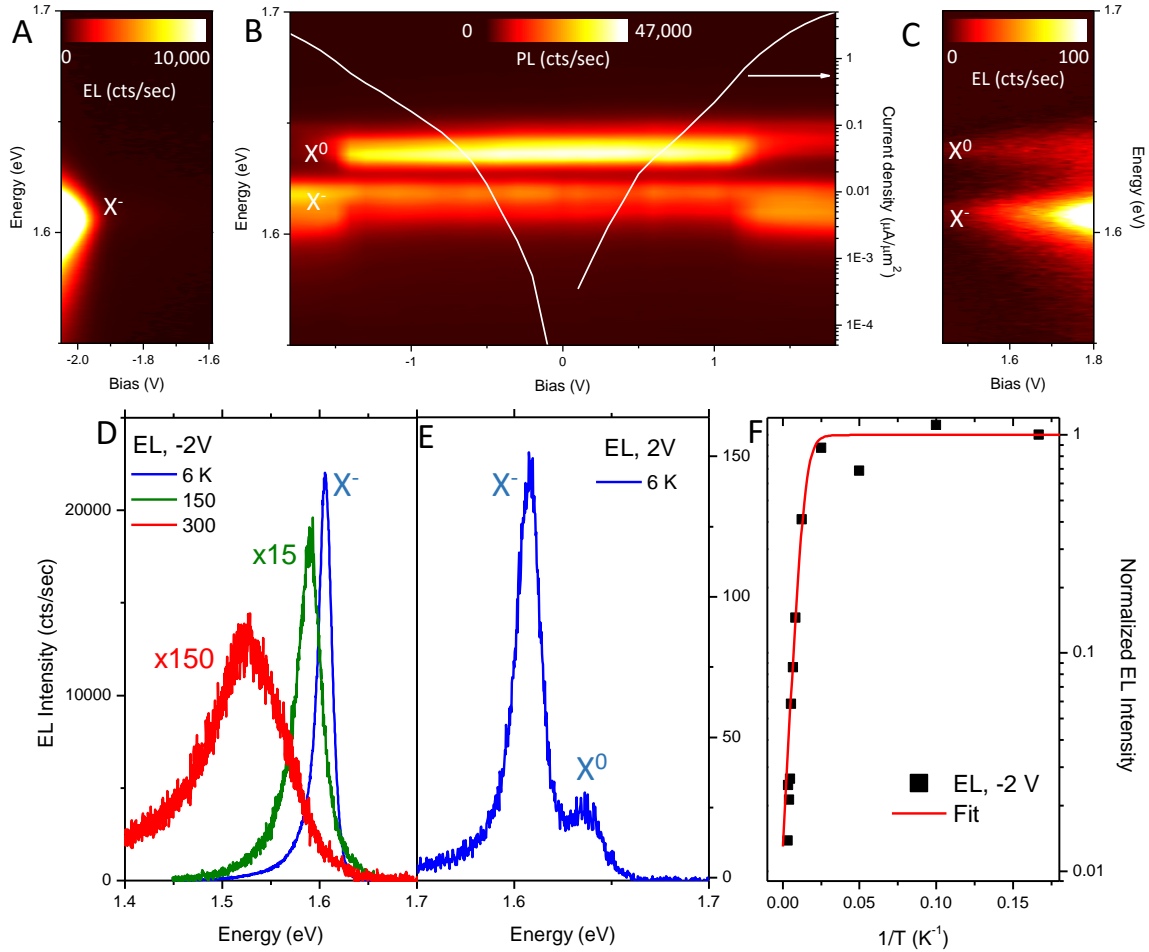


Fig. 6.4 Contour maps of the EL spectra from a MoSe₂ LED at T=6 K for negative (A) and positive (C) bias voltage. (B) PL contour map as a function of V_b , measured at an excitation power of $20 \mu\text{W}$ and excitation energy of 2.33 eV. White curve in (B) is the j -V curve. (D) Low temperature electroluminescence spectra collected for a MoSe₂ LED at several temperatures at a fixed excitation bias voltage. (E) Electroluminescence spectra taken at 6 K with $V_b=2$ V for a MoSe₂ LED. (F) Arrhenius plot of the EL yield with temperature for the MoSe₂ LED, fit shown with a red solid line using eq. 6.2 gives calculated activation energy E_1 of ~ 30 meV.

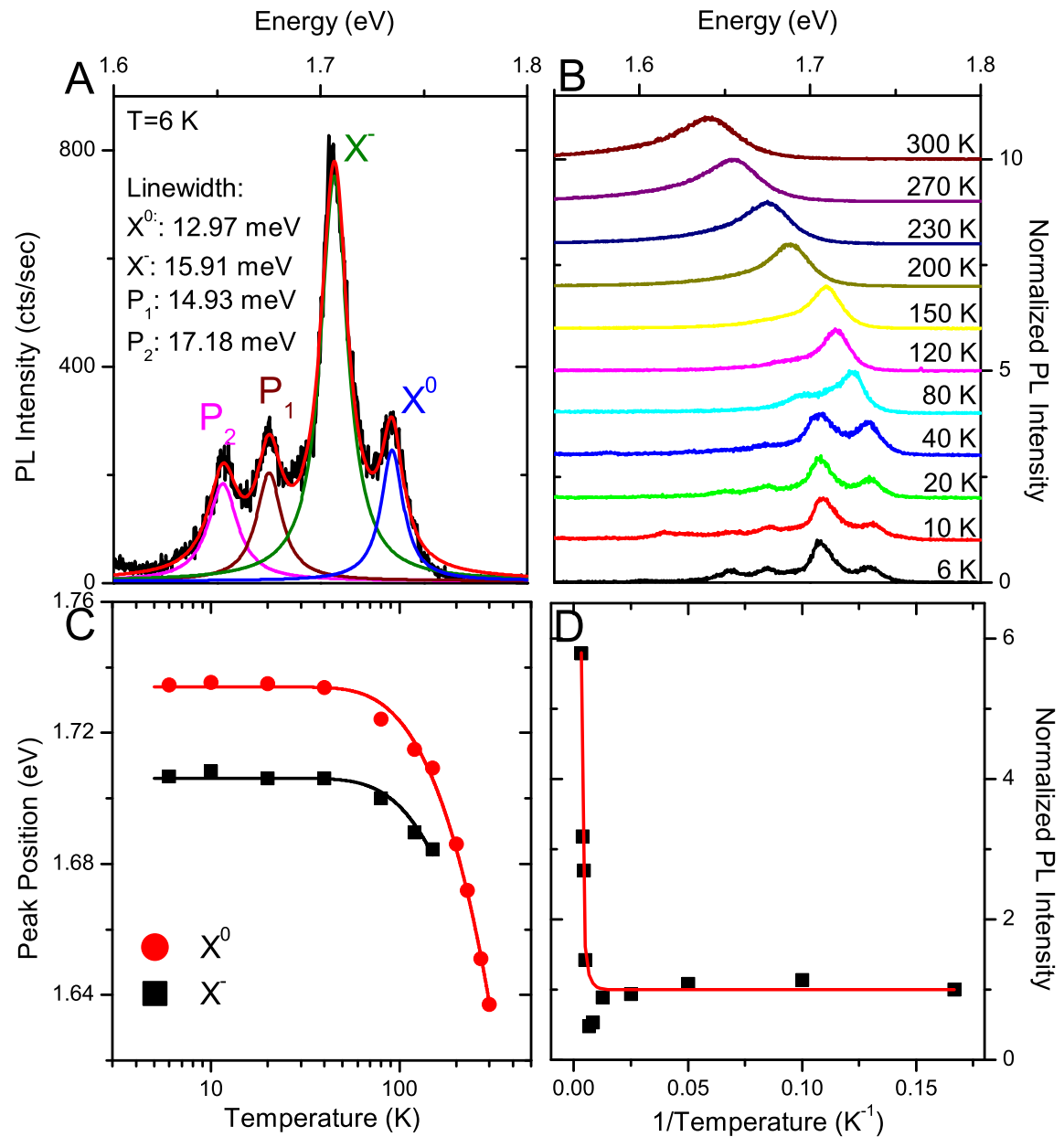


Fig. 6.5 (A) Typical PL spectrum of WSe₂ LED taken at T=6 K with 50 μ W showing four characteristic peaks attributed to a neutral (X^0) and charged (X^-) excitons and two additional feature (P_1 and P_2) with a measured linewidth of 12.97, 15.91, 14.93 and 17.18 meV respectively calculated with lorentzian fittings (blue, green, brown and pink solid lines). (B) Normalized PL temperature dependence of WSe₂ LED from 6 to 300 K. (C) Peak positions of X^0 (red) and X^- (black) as a function of temperature. Solid lines indicate fittings with Bose-Einstein model (eq. 6.1). (D) Arrhenius plot of the WSe₂ PL yield to calculate, using the eq.6.2, an activation energy E_1 of ~ 40 meV.

Temperature-dependent behaviour of MoSe₂ and WSe₂ van der Waals heterostructure LEDs

their intensities relative to X^0 and X^- vary from device to device. For example, in 6.6E EL obtained at three different temperatures with a different device (WSe₂-LED2) ($V_b=2.3$ V) shows only emission from X^- (1.698 eV at 6 K) up to room temperature with an increase of ~ 400 times compared to low-T. This kind of device is made with thin h-BN barriers (2 single layers) and no X^0 is observed. Finally, EL enhancement is studied in fig. 6.6F with an Arrhenius fit. Like in PL, this result contrasts with the T-dependent behaviour of EL from MoSe₂.

6.4 Discussion

MoSe₂ LED exhibits a typical behaviour with temperature: PL and EL intensity decrease when increasing temperature up to 300 K. WSe₂ shows the opposite behaviour with an enhancement of both PL and EL intensities at room temperature. This comparison is illustrated in Fig. 6.7. WSe₂ EL, shown in red squares, increases with a factor of 200 from 5 to 300 K while MoSe₂ EL (blue squares) decreases by a factor of ~ 100 .

The devices described in chapter 5 show a significant decrease in EL intensity by a factor exceeding 10 when the temperature was varied between 10 and 300 K. As an example, Fig. 6.8 shows EL temperature dependence of a MoS₂ LED. In 6.8A I-V curves from 6 to 300 K are plotted showing an increase in the current through the device with temperature of ~ 20 μ A at high bias ($|V_b|>2$ V). EL spectra in the same range of temperatures for $V_b=2.7$ V can be observed in Fig. 6.8B. An Arrhenius fit of EL yield is plotted in fig. 6.8C that allows to calculate an activation energy of 15.57 meV. The decrease in intensity corresponds to a reduction in EQE to ~ 0.1 % at room temperature.

The strong increase of the EL with temperature in WSe₂ corresponds to a rise of EQE. This makes van der Waals heterostructures with embedded WSe₂ monolayers

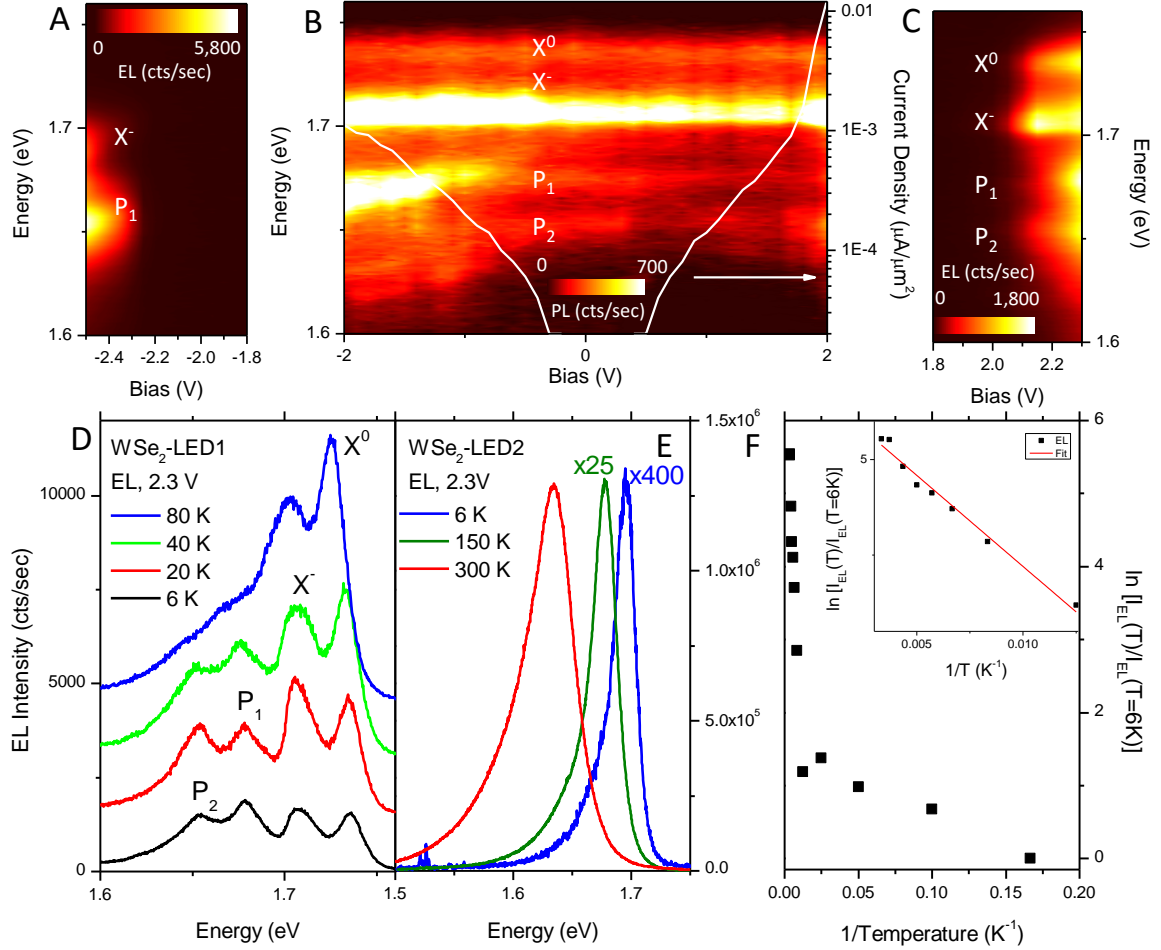


Fig. 6.6 Contour maps of the EL spectra from the WSe₂ LED1 at T= 6 K for negative (A) and positive (C) bias voltage. (B) PL contour map as a function of V_b , measured at an excitation power of 50 μ W and excitation energy of 2.33 eV. White curve in (B) is the j -V curve. (D) Electroluminescence spectra collected for the WSe₂-LED1 at several temperatures at a fixed excitation bias voltage. (E) Electroluminescence spectra taken at different temperatures for the WSe₂-LED2 with two layer hBN tunnel barriers. This sample demonstrates a 200x increase of the EL output when the temperature is increased from T= 6 K to T = 300 K. (F) Arrhenius plot of the EL yield with temperature for the WSe₂ device shown in (E) inset: zoom in around the high temperature region used for the linear fit to calculate an activation energy of \sim 40 meV.

Temperature-dependent behaviour of MoSe₂ and WSe₂ van der Waals heterostructure LEDs

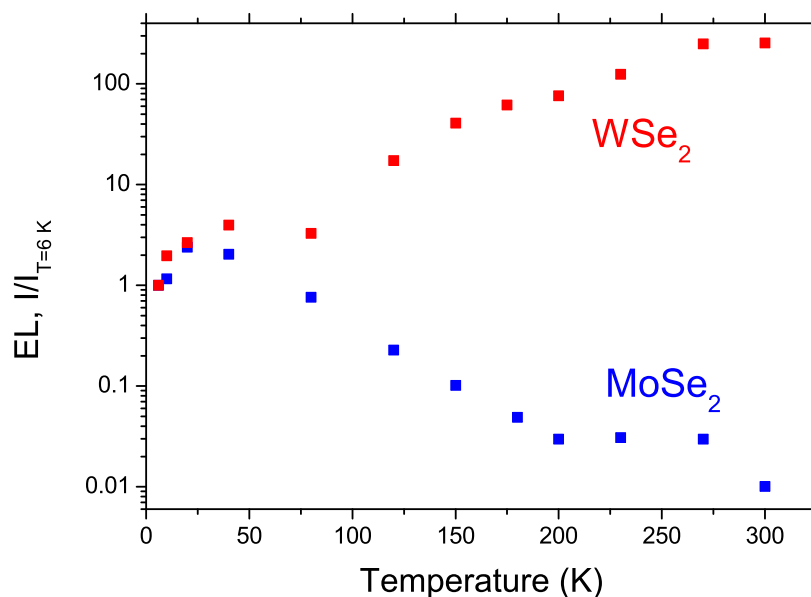


Fig. 6.7 Comparison between the temperature dependence of the electroluminescence intensity for the WSe₂-LED2 (blue) and MoSe₂ (red) showing opposite trends with increasing T. The intensities are normalised by those measured for each device at T=6 K.

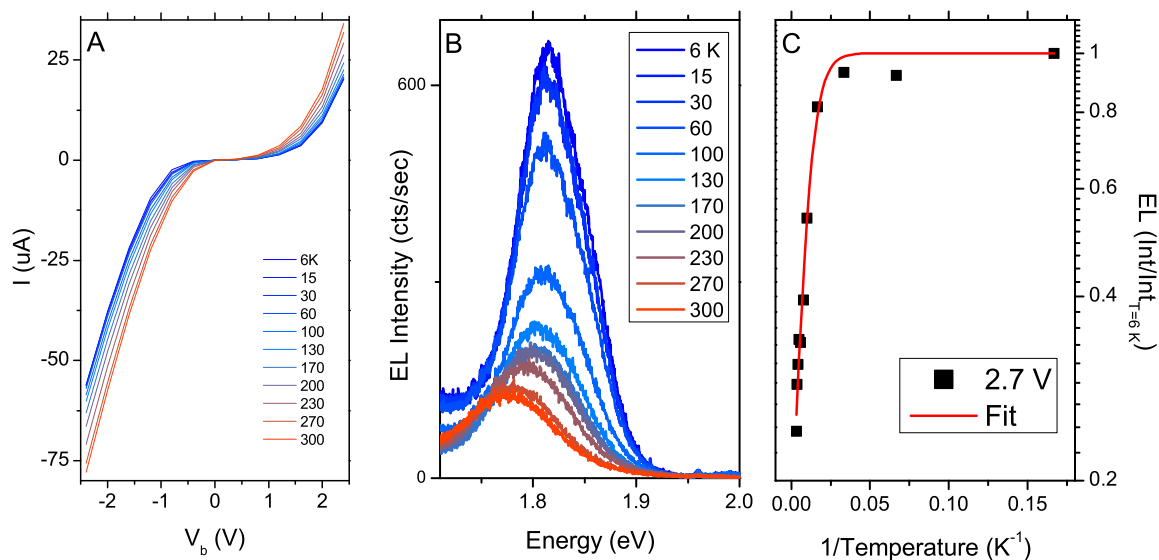


Fig. 6.8 MoS₂ EL Temperature dependence. (A) I-V curves taken from 6 to 300 K. (B) EL spectra with V_b=2.7 V, T=6 to 300 K. (C) Arrhenius plot for EL intensity of spectra in (B).

highly promising material for ultra-thin flexible LEDs. Fig. 6.9A shows EQE T-dependence for three WSe₂ devices. Here the EQE is calculated as described in section 2.2.1 and appendix B, similar to previous chapter. It is observed in Fig. 6.9A that the EQE shows the characteristic increase with temperature reaching 10% at T = 300 K, a factor 200-500 improvement in the room temperature performance as compared to the single-monolayer MoS₂ or MoSe₂ LEDs. In addition to this, in Fig. 6.9C we observe a monotonically increasing EQE as a function of bias voltage and injection current density to a maximum measured value of $j=103$ A/cm². Fig. 6.9B shows the EL spectra obtained from device 3 in Fig. 6.9A operated at room temperature at increasing injection current densities with a peak emission of more than 1.3 million counts per second.

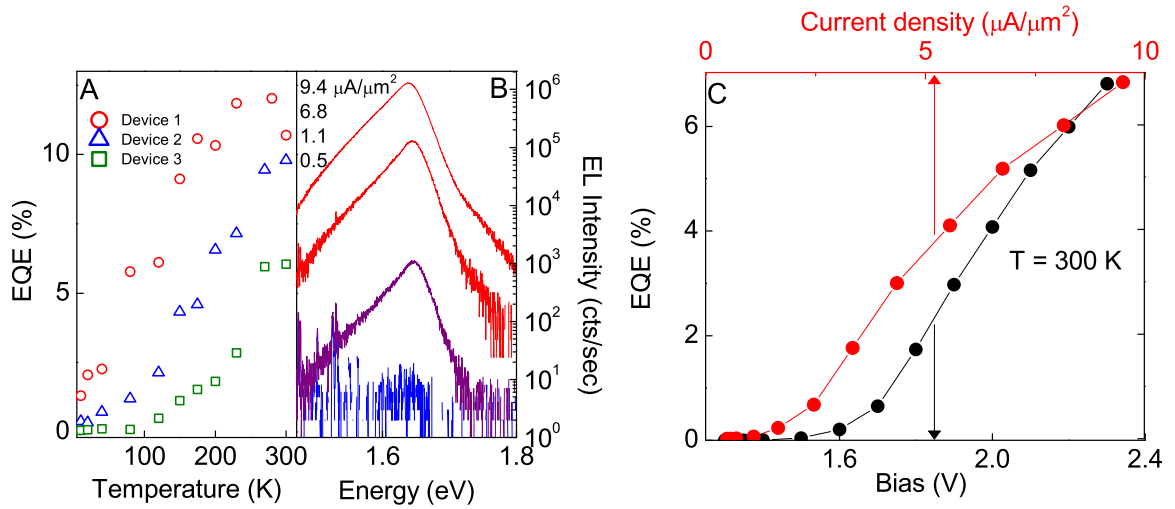


Fig. 6.9 (A) Temperature dependence of the quantum efficiency for three separate WSe₂ LEDs measured at bias voltages and injection currents of, Device 1 (2.8 V, $j=0.15$ $\mu\text{A}/\mu\text{m}^2$), Device 2 (2.8 V, $j=0.5$ $\mu\text{A}/\mu\text{m}^2$) and Device 3 (2.3 V, $j=8.8$ $\mu\text{A}/\mu\text{m}^2$). (B) Individual electroluminescence spectra plotted for four different injection current densities for Device 3. (C) The external quantum efficiency plotted against bias voltage and injection current density at T = 300 K for device 3. Note, the EQE monotonically increases even up to current densities of 10 $\mu\text{A}/\mu\text{m}^2$ or 1000 A/cm².

Indeed, a common drawback of commercial LED lighting is a suppression of the

Temperature-dependent behaviour of MoSe₂ and WSe₂ van der Waals heterostructure LEDs

EQE (so-called 'efficiency droop') at high injection currents caused by increased non-radiative processes [15, 16]. In contrast, the presented WSe₂-based van der Waals heterostructure LEDs become brighter at higher temperature, and their efficiency remains high and increases with current densities even at high j as shown in Fig. 6.9C.

We suggest that the mechanism of the unusual T-dependence of EL and PL in WSe₂ LEDs is related to the ordering of the spin states in the free carrier bands as illustrated in Fig. 6.1, leading to the specific ordering of the bright and dark exciton sub-bands in these materials [9, 10]. The discussion below will be focused on the behaviour of the neutral exciton, the PL and EL of which dominate at room T in the majority of LEDs described. The dark and bright exciton sub-bands are split by approximately $\Delta_{SO} \sim 30\text{-}40$ meV. The splitting is further modified by the electron-hole exchange interaction, which is different for the bright and dark states but is of the order of 1-2 meV [9–12] and is therefore neglected in the following discussion.

At low temperature the exciton population accumulates in the low energy dark exciton sub-band, which mostly decay via (i) non-radiative escape, and (ii) phonon- and electron-assisted light-emission processes. In processes (ii), momentum and energy conservation is fulfilled as a result of the exciton scattering with phonons or electrons. It is possible, that the origin of the features P_1 - P_2 is exactly through such phonon or electron assisted decay of the dark exciton, which would explain the strong intensity of those features at low T. At the same time, at low T the bright exciton sub-band is weakly populated as it is shifted to higher energy by $\sim \Delta_{SO}$ with respect to the dark exciton sub-band and therefore the intensity of the X^0 line is low.

As the temperature increases, the thermal equilibrium dictates the increase of the exciton population in the bright exciton sub-band in WSe₂. This population will be mostly contained in the high- k exciton 'reservoir' outside the small- k light-cone

states. Excitons from the reservoir will then be scattered into the light-cone where they recombine radiatively. The scattering will occur as a result of interaction with phonons and electrons, the latter is particularly significant in the EL regime.

The fact that at high T the intensity of the X^0 line increases both in PL and EL is the manifestation of such thermal activation behaviour. The rate of the X^0 line intensity increase with T depends on several factors, and will not only reflect the equilibrium Boltzmann distribution of excitons among the four exciton and trion sub-bands. In a simplified approximation, the light emission intensity will be proportional to the product of such equilibrium population in the bright exciton sub-band and the ratio of $\tau_{nr}/(\tau_{rel} + \tau_{nr})$, where τ_{nr} is a typical non-radiative decay time for the high-k reservoir excitons, and τ_{rel} a typical relaxation time of the high-k excitons into the light-cone states. Both time constants may be functions of temperature, and vary from sample to sample depending on the residual doping of the layers and device architecture (the latter for example defining the carrier dwell time in the active layer). Furthermore, the X^0 linewidth shows pronounced dependence on T, typically increasing from 10-20 meV at T=10K to over 40 meV at room T. Such homogeneous broadening would effectively relax the energy conservation required for the scattering into the light-cone states and would modify the way the reservoir states (as well as the dark excitons) contribute to PL and EL.

The factors considered above imply a non-trivial dependence of the PL and EL intensity on temperature in WSe₂. Indeed the overall increase of PL and EL in the range from 10 to 300 K varied between different devices. Still, the Arrhenius fit (Fig. 6.6F) to the exponential increase of the EL with increasing temperature on some of our devices yields the characteristic energy of ~ 40 meV, which is in reasonable agreement with the predicted Δ_{SO} [9]. On the other hand, the MoSe₂ films and LEDs, where the dark exciton sub-band is higher in energy than the bright exciton sub-band,

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always showed the opposite behaviour with the notable decrease of light emission with increasing T . While it is expected that the emission efficiency would decrease due to the transfer of some exciton population into the dark exciton sub-band and also into the high- k states, such a strong decrease by a factor of 100 and more may only be possible if the non-radiative escape time shortens at elevated temperature.

6.5 Conclusions

High-efficiency LEDs made from van der Waals heterostructures comprising a single atomic layer of WSe_2 as active light-emitting material, hBN tunnel barriers, and graphene electrodes for vertical current injection were fabricated. Such WSe_2 -based LEDs show unexpectedly enhanced performance at room temperature compared with the low- T operation. This enhancement is also in contrast to MoSe_2 and MoS_2 LEDs studied in this thesis, where both PL and EL decrease by a factor of 10 to 100 when the temperature is varied from 10 to 300 K. With the record external room temperature efficiencies of 10%, such single layer WSe_2 LEDs present significant promise for future development of flexible opto-electronic components. The efficiency can be boosted further by creating multiple quantum well devices as described in chapter 5 and by the fine tuning of the h-BN tunnel barrier thickness. One of the remaining challenges is scalable production of these components, only possible with well controlled wafer-scale growth techniques [17, 18].

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Chapter 7

Conclusions

In this thesis optical characterisation of two different groups of two dimensional semiconductor materials was carried out: III-VI compounds and transition metal dichalcogenides (TMDCs). With the latter, light emitting diodes based on van der Waals heterostructures were realised and characterised using photoluminescence (PL) spectroscopy. The key results for each topic were the following:

1. In chapter 3, low temperature cw and time resolved PL of two dimensional GaTe and GaSe two dimensional films was presented. It was observed that PL intensity decreases significantly when the film thickness is reduced for both materials and that films thinner than 7 nm show no detectable PL. We claim that the reduction of quantum yield is a consequence of non-radiative processes associated with surface states. A model was proposed to differentiate between the fast non-radiative carrier escape to surface traps in the thin 15–20 nm layers near the film surface, and slower decay in an intermediate region of the film, where the carriers first move to the surface layers and then decay. The model accounts for the change in the character of the PL decay for thin films with thicknesses less than 30–40 nm, where we expect only fast direct relaxation to

Conclusions

surface traps. No other trends with thickness were observed including notable changes in PL lifetime and blue shift which would be expected due to quantum confinement. Further protection of the films such as oxygen-free encapsulation is needed to hinder the fast PL intensity decay. This will allow III-VI materials to be considered as building blocks for van der Waals heterostructures.

2. In chapter 4, the stability of thin films of InSe and GaSe was studied using PL and Raman spectroscopy. We showed that GaSe films are affected by chemical interactions on the surface layers leading to film erosion with an average rate of 0.14 ± 0.05 nm/hour when they are exposed to ambient conditions. This was observed as a significant decrease over time of PL and Raman intensity. In the case of InSe, degradation under the same condition is much weaker. It demonstrates that this III-VI layered material is a good candidate to be used in van der Waals heterostructure. A reduction of the degradation rate by more than two orders of magnitude in GaSe films can be achieved by using encapsulation of the films with dielectrics such as Si_xN_y or SiO_2 .
3. In chapter 5, light emitting diodes based on van der Waals heterostructures fabricated by stacking graphene, insulating hexagonal boron nitride and various semiconducting TMDC monolayers were presented. Our devices made with a single quantum well (QW) exhibit an external quantum efficiency (EQE) of $\sim 1\%$ under electrical pumping. When increasing the number of quantum wells, the quantum efficiency increases up to $\sim 10\%$. The emission can be tuned over a wide range of wavelengths by appropriately choosing and combining 2D semiconductors such as MoS_2 , WS_2 and WSe_2 .
4. In chapter 6, light emitting diodes based on van der Waals heterostructures made from WSe_2 and MoSe_2 were presented. Here we show that the EQE in van der

WSe₂ LEDs comprising monolayers of WSe₂ unexpectedly grows with temperature. Room temperature EQE under electrical injection of 10% was demonstrated in single WSe₂ monolayer QW LEDs exceeding by more than 500 the previous best performance of MoX₂ QWs in ambient conditions. Such temperature activated behaviour is a consequence of the strong spin-orbit interaction and specific ordering of the spin states in tungsten dichalcogenides resulting in the lowest-energy dark exciton sub-band. A comparison with a MoSe₂-based LED was carried out to emphasize this behaviour. In MoSe₂ LEDs strong decrease of EQE was observed at elevated temperatures.

Appendix A

Model for PL intensity dependence on thickness of GaTe and GaSe thin films

In order to describe the observed trend of photoluminescence (PL) intensity as a function of film thickness h_{film} in GaSe and GaTe thin films we have developed a simplified rate equation model described below. It describes the behavior of photo-excited carriers in terms of populations of e-h pairs, i.e. electrons and holes are not treated separately for simplicity.

A.1 Modeling

As shown in Fig. 3.7(b) we assume that the film is divided into three regions: (1) two regions of thickness h_0 near the film surfaces where fast carrier relaxation to surface traps occurs leading to non-radiative carrier escape; (2) a 'normal' region of thickness $h_{film} - 2h_0$ in the central part of the film where carriers do not decay non-radiatively, but can escape into the regions (1), where they undergo non-radiative decay. Here the

Model for PL intensity dependence on thickness of GaTe and GaSe thin films

h_0 value may be associated with a depletion depth or an average surface trap radius[1]. It is assumed to be independent of the film thickness and may be different for films made of different materials.

In region (1), in first approximation the average time it takes for the carrier/e-h pair to escape non-radiatively is proportional to half the thickness of region (1) (can be understood as the average time for the carrier to reach the surface or as the overlap of the wavefunction of the carrier and the trap). This half-thickness is given by $h_0/2$ for the two regions adjacent to the surface for film thicknesses $h_{film} > 2h_0$, and is $h_{film}/4$ for $h_{film} \leq 2h_0$, when the two surface regions merge. Thus we introduce a non-radiative decay time in the regions of type (1) as $\tau_{nr1} = (h_0/2)/u_1$ for $h_{film} > 2h_0$, and $\tau_{nr1} = (h_{film}/4)/u_1$ for $h_{film} \leq 2h_0$. Here u_1 is a constant with the dimensions of meter per second.

In region (2), the non-radiative escape time reflects the average time it takes for a carrier or an e-h pair to reach any of the regions (1). The underlying mechanism for this process may be depletion and band-bending expected at the film surface leading to charge separation and non-radiative decay [1]. We assume that once the carrier or e-h pair has reached region (1) it escapes non-radiatively. In first approximation, the average time it takes a carrier/e-h pair to reach region (1) is proportional to half the thickness of region (2), $h_{film}/2 - h_0$. Thus we introduce a non-radiative decay time in region (2) as $\tau_{nr2} = (h_{film}/2 - h_0)/u_2$. Here u_2 is a constant with the dimensions of meter per second, effectively corresponding to the average carrier velocity in region (2). Notably, in films with $h_{film} \leq 2h_0$, region (2) does not exist, and the only non-radiative escape that we consider is due to the escape to traps as described above for region (1).

In equations below, we denote e-h pair populations in regions (1) and (2) as N_1 and N_2 , respectively, and the population of the 'PL states' as N_{loc} . Appropriately,

$N_2 = 0$ for $h_{film} \leq 2h_0$. In both regions, e-h pairs relax with the time τ_{rel} into the 'PL states' giving rise to radiative recombination. Non-radiative escape from the 'PL states' is neglected. This assumption is based on the fact that in most cases PL is observed from the tightly-bound localized states. N_{loc} states decay radiatively with a characteristic time τ_{PL} , so that the PL intensity $I_{PL} = N_{loc}/\tau_{PL}$. Saturation of 'PL states' is neglected as the data reported in Figs.3.2(b),3.3(b) is measured for the laser powers where saturation effects are unimportant. The light absorption coefficient, α is assumed to be the same in both types of regions (1) and (2). For the small thicknesses of the films that we consider, the e-h pair generation rate is equal to $P\alpha$ multiplied by the thickness of the regions (1) or (2) as given below. Here P is the e-h pair generation rate due to the laser excitation rescaled to account for the PL detection efficiency.

The following rate equations can thus be introduced for the case $h_{film} > 2h_0$:

$$\begin{aligned} dN_1/dt &= 2h_0P\alpha - N_1/\tau_{rel} - N_1/\tau_{nr1} \\ dN_2/dt &= (h_{film} - 2h_0)P\alpha - N_2/\tau_{rel} - N_2/\tau_{nr2} \\ dN_{loc}/dt &= N_1/\tau_{rel} + N_2/\tau_{rel} - N_{loc}/\tau_{PL} \end{aligned}$$

For the case $h_{film} \leq 2h_0$ we will have:

$$\begin{aligned} dN_1/dt &= Ph_{film}\alpha - N_1/\tau_{rel} - N_1/\tau_{nr1} \\ N_2 &= 0 \\ dN_{loc}/dt &= N_1/\tau_{rel} - N_{loc}/\tau_{PL} \end{aligned}$$

In the steady-state case we obtain the following expression for PL intensity for $h_{film} > 2h_0$:

$$I_{PL} = 2h_0P\alpha/(1 + \tau_{rel}/\tau_{nr1}) + (h_{film} - 2h_0)P\alpha/(1 + \tau_{rel}/\tau_{nr2}), \quad (\text{A.1})$$

Model for PL intensity dependence on thickness of GaTe and GaSe thin films

where $\tau_{nr1} = \tau_1 h_0/2$ and $\tau_{nr2} = \tau_2(h_{film}/2 - h_0)$, and for $h_{film} \leq 2h_0$

$$I_{PL} = h_{film} P\alpha / (1 + \tau_{rel}/\tau_{nr1}), \quad (\text{A.2})$$

where $\tau_{nr1} = (h_{film}/4)/u_1$.

By substituting these expressions for the non-radiative decay times directly we obtain for PL intensity for $h_{film} > 2h_0$:

$$I_{PL} = 2h_0^2 P\alpha / (h_0 + 2\tau_{rel}u_1) + (h_{film} - 2h_0)^2 P\alpha / (h_{film} - 2h_0 + 2\tau_{rel}u_2). \quad (\text{A.3})$$

For $h_{film} \leq 2h_0$ we get:

$$I_{PL} = h_{film}^2 P\alpha / (h_{film} + 4\tau_{rel}u_1), \quad (\text{A.4})$$

A.2 Results of the modeling

A.2.1 Results for GaSe thin films

According to Eqs. A.3, A.4, we have four parameters, with which we describe the observed dependence of I_{PL} on the film thickness: h_0 , the products $P\alpha$, $\tau_{rel}u_1$ and $\tau_{rel}u_2$.

Further analysis of Eqs.A.3, A.4 shows that these parameters are not completely independent. This becomes obvious if we note that for small h_{film} where we expect fast non-radiative process (i.e. large $\tau_{rel}u_1$) we get $h_{film} \ll 4\tau_{rel}u_1$, and Eq.A.4 reduces to a simple parabolic dependence: $I_{PL} \approx h_{film}^2 P\alpha / (4\tau_{rel}u_1)$. This function is independent of individual parameters $P\alpha$ and $\tau_{rel}u_1$, and depends on their ratio only.

For large h_{film} where we expect slower non-radiative process we get $(h_{film} - 2h_0)$

of the order of $2\tau_{rel}u_2$, and a simple approximation can be used for Eq.A.3 giving $I_{PL} \approx (h_{film} - 2h_0)P\alpha$. From here we find that $P\alpha$ should be of the order of $10^5 \text{ nm}^{-1}\text{s}^{-1}$.

The two functions described by Eqs.A.3, A.4 and especially their asymptotics have significantly different form, effectively with different 'slopes' as seen e.g. in Fig.3.3(b) for the case of GaSe. For GaSe data in Fig.3.3(b), it is relatively easy to define the position of a characteristic kink around $h_{film} = 30 \text{ nm}$, where the behavior changes. We use this observation to fix the value of h_0 to 15 nm .

In the next step, we have explored a range of fitting functions providing a reasonable agreement with the experiment. We conclude that our model can reliably predict the order of magnitude of the fitting parameters, but not the actual value. So we find that $\tau_{rel}u_1$ is of the order of a 1000 nm and $\tau_{rel}u_2$ is of the order of a 100 nm .

As an illustration for the GaSe data, in Fig.3.3(b) we show a fitting with $P\alpha = 3.5 \times 10^5 \text{ nm}^{-1}\text{s}^{-1}$, $\tau_{rel}u_1 = 2900 \text{ nm}$ and $\tau_{rel}u_2 = 150 \text{ nm}$. The figure also shows the variation of PL intensity as would be expected from the change in absorption only (dashed line). This would correspond to a thickness independent non-radiative processes occurring with the same rate in the whole volume of the film. This curve is described by the expression $I_{PL} = I_{GaSe}[1 - \exp(-\alpha_{GaSe}h_{film})]$, where I_{GaSe} is the PL intensity for films with $h_{film} \approx 200 \text{ nm}$ and the absorption coefficient $\alpha = 1000 \text{ cm}^{-1}$ [2].

A.2.2 Results for GaTe thin films

We used the same fitting procedure for GaTe with the results shown in Fig.3.2(b). Although the 'kink' in the thickness-dependence of I_{PL} for GaTe is not as pronounced as for GaSe, it is still observable at a slightly higher value of $2h_0 = 40 \text{ nm}$. The model describes adequately the data for h_{film} up to around 90 nm , but fails to reproduce

References

the very steep enhancement of PL for thicker films. This probably occurs due to the contribution of the free excitons to the PL signal. It is probable that there is an additional strong non-radiative escape mechanism for free excitons in thin films, which is overcome in thicker films with the result that the PL intensity for thick films grows faster than the model predicts. Another reason could be deviation from the effectively 'ballistic' transport that we assumed leads to the carrier escape into regions (1), and its replacement for the larger thicknesses with a slower 'diffusion' process.

In a similar way to GaSe, we confine ourselves to determination of the order of magnitude only for the main parameters, which are similar to GaSe: we find that $\tau_{rel}u_1$ is of the order of a 1000 nm and $\tau_{rel}u_2$ is of the order of a 100 nm. As an illustration for the GaTe data, in Fig.3.2(b) we show a fitting with $P\alpha = 5.0 \times 10^6 \text{ nm}^{-1}\text{s}^{-1}$, $\tau_{rel}/\tau_1 = 3800 \text{ nm}$ and $\tau_{rel}/\tau_2 = 200 \text{ nm}$. The figure also shows the variation of PL intensity as would be expected from the change in absorption only (dashed line). This curve is described by the expression $I_{PL} = I_{GaTe}[1 - \exp(-\alpha_{GaTe}h_{film})]$, where I_{GaTe} is the PL intensity for films with $h_{film} \approx 200 \text{ nm}$. The absorption coefficient $\alpha_{GaTe} = 5000 \text{ cm}^{-1}$ is used according to Ref. [2].

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Appendix B

Calculation of collection efficiency

In chapters 5 and 6 the external quantum efficiency was calculated following the description of section 2.2.1. In order to get the most accurate calculation it is necessary to take into account the losses on the system used to measure. This will depend on the device environment inside the sample, optics used to collect and guide the light emitted and the detector. On this appendix a detailed description of the calculation is presented

In section 2.2.1 the quantum efficiency was defined as the number of photons emitted per number of injected electron-hole pairs. In order to estimate the number of emitted photons we need to estimate our collection efficiency. The total loss is defined as:

$$\eta = \eta_{Lens}\eta_{optic}\eta_{system} \quad (\text{B.1})$$

η_{optic} is the loss of all the optical components in the optical circuit. It was measured directly using a 1.96 eV laser and a power meter to determine the loss at each component. We find $\eta_{optic} = 0.18$.

η_{system} - converts the number of photons arriving at the incoming slit of the detector

Calculation of collection efficiency

into the detector counts. It takes into account the loss of photons which pass through the slit, grating and onto the CCD and has been again measured directly by using the 1.96 eV laser and taking spectra of the laser for different powers in order to get a counts vs incident photons. For our system we get 4203 integrated cts/sec per 1 pW. If the system were 100% efficient we should count $N = P/h\nu = 3177476$ photons, therefore we arrive at an estimate for the system efficiency to be $\eta_{system} = 4203/3177476 = 1.32 \times 10^{-3}$.

η_{Lens} is the efficiency of the lens collection[3]. We use a 50x objective with a numerical aperture, $NA = 0.55$. LEDs are fabricated on either two substrates, firstly Si/SiO₂(290nm) with refractive index of Si (n=3.734) and SiO₂(n=1.645) or distributed Bragg reflectors which consist of 10 alternating quarter wave pairs (187.5 nm) of SiO₂ (n=1.46) and NbO₂ (n=2.122) [1, 2].

Numerical simulations allow us to make an improved estimate of the collected light emitted from a dipole on each substrate type. Our NA=0.55. This gives a collection angle of 33.4°.

So for a WSe₂ flake on SiO₂ we find that:

$$\eta_{Lens} = \frac{P_{SiO_2}(33.4)}{P_{SiO_2}(180)} = 2.5\% \quad (B.2)$$

while for WSe₂ flake on a DBR we find that:

$$\eta_{Lens} = \frac{P_{DBR}(33.4)}{P_{DBR}(180)} = 31\% \quad (B.3)$$

This gives us two loss factors depending on the substrate the LED was fabricated onto.

$$\eta_{SiO_2} = 0.18 \times 1.32 \times 10^{-3} \times 0.025 = 5.94 \times 10^{-6} \quad (B.4)$$

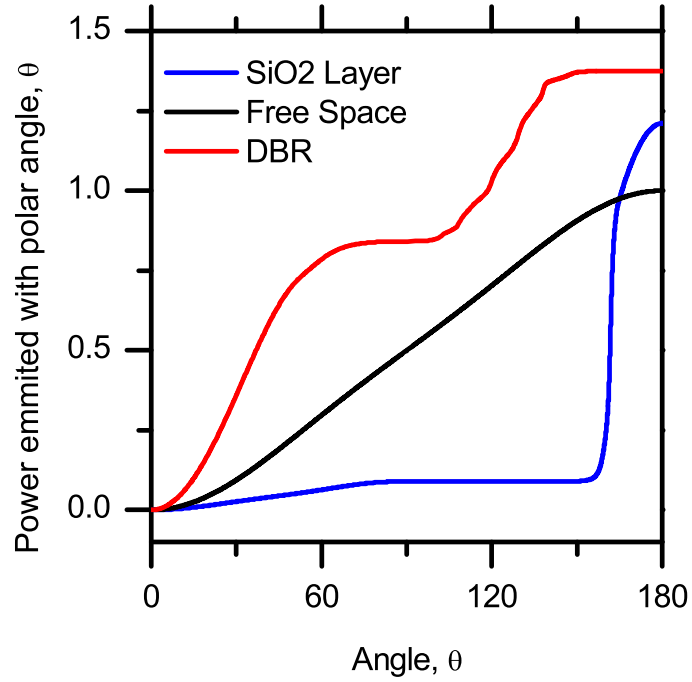


Fig. B.1 Total power emitted within a polar angle for an emitting dipole placed on Si/SiO₂ and on a distributed Bragg reflector (DBR) as well as in free space.

$$\eta_{DBR} = 0.18 \times 1.32 \times 10^{-3} \times 0.31 = 7.37 \times 10^{-5} \quad (\text{B.5})$$

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